

Monday 21 May 2012 – Morning

**GCSE GATEWAY SCIENCE
ADDITIONAL SCIENCE B**

B624/02 Unit 2 Modules B4 C4 P4 (Higher Tier)

Candidates answer on the Question Paper.
A calculator may be used for this paper.

OCR supplied materials:
None

Other materials required:

- Pencil
- Ruler (cm/mm)

Duration: 1 hour



Candidate
forename

Candidate
surname

Centre number

Candidate number

MODIFIED LANGUAGE

INSTRUCTIONS TO CANDIDATES

- Write your name, centre number and candidate number in the boxes above. Please write clearly and in capital letters.
- Use black ink. HB pencil may be used for graphs and diagrams only.
- Answer **all** the questions.
- Read each question carefully. Make sure you know what you have to do before starting your answer.
- Write your answer to each question in the space provided. Additional paper may be used if necessary but you must clearly show your candidate number, centre number and question number(s).
- Do **not** write in the bar codes.

INFORMATION FOR CANDIDATES

- The number of marks is given in brackets [] at the end of each question or part question.
- A list of physics equations is printed on page two.
- The Periodic Table is printed on the back page.
- The total number of marks for this paper is **60**.
- This document consists of **28** pages. Any blank pages are indicated.



EQUATIONS

$$\text{speed} = \frac{\text{distance}}{\text{time taken}}$$

$$\text{acceleration} = \frac{\text{change in speed}}{\text{time taken}}$$

$$\text{force} = \text{mass} \times \text{acceleration}$$

$$\text{work done} = \text{force} \times \text{distance}$$

$$\text{power} = \frac{\text{work done}}{\text{time}}$$

$$\text{kinetic energy} = \frac{1}{2}mv^2$$

$$\text{potential energy} = mgh$$

$$\text{weight} = \text{mass} \times \text{gravitational field strength}$$

$$\text{resistance} = \frac{\text{voltage}}{\text{current}}$$

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Question 1 begins on page 4.

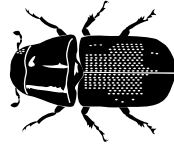
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Answer **all** the questions.

Section A – Module B4

1 Large areas of Canada are covered by forests containing pine trees.

(a) Many of the pine trees in Canada are being killed by small beetles.



The mountain pine beetle feeds from the tubes in the tree that transport sugar.

The beetle also infects the tree with a fungus.

This fungus blocks up the xylem vessels in the tree trunk.

This kills the tree.

(i) Write down the name of the tubes from which the beetle feeds.

..... [1]

(ii) The leaves (needles) of the infected pine trees often change colour because they become short of minerals.

Explain why they become short of minerals.

.....

..... [1]

(iii) The leaves often die because they lack water.

This lack of water causes the stomata to close.

Explain the mechanism that causes the stomata to **close**.

.....

.....

.....

..... [3]

- (iv) Biomass, such as dead trees, may be used in many ways.

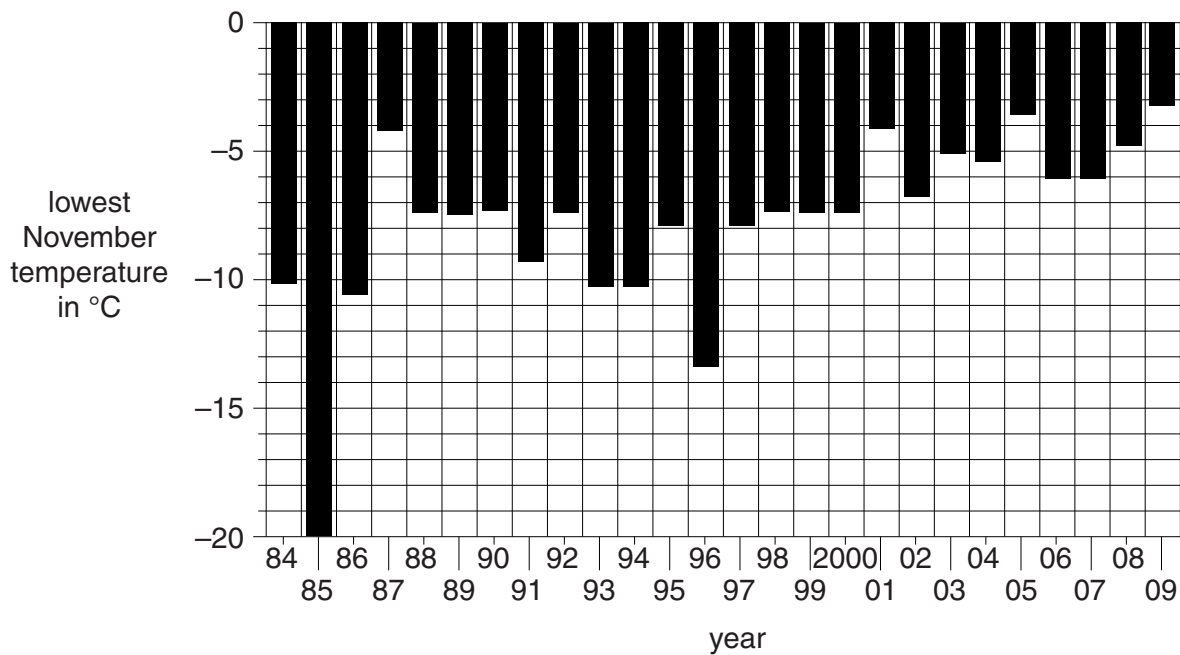
These dead trees are used for fuel and burned **as soon as possible**.

Suggest why it is important that these trees are burned as soon as possible.

..... [1]

- (b) Many of the beetles are killed if temperatures drop below -6°C in November.

The graph shows the lowest November temperature in the pine forest each year from 1984 to 2009.



- (i) In 1988 there was a large outbreak of the beetle but they soon decreased in number.

Use the graph to work out the year when the next major outbreak started.

..... [1]

- (ii) The recent outbreak has lasted for a number of years.

Scientists are worried that the recent outbreak might continue for many more years.

Use the graph to suggest why.

.....

 [2]

[Total: 9]

2 Bill grows lettuces on his farm.

(a) He finds that the lettuces are being eaten by slugs and birds.

The slugs are being eaten by hedgehogs.

Sketch a **pyramid of biomass** to show these feeding relationships.

Write the names of the organisms in the correct boxes in the pyramid.

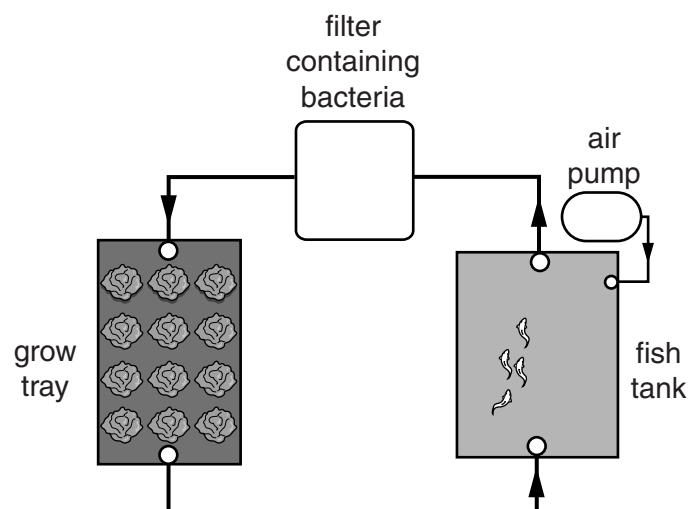
[2]

(b) Bill investigates a new way of growing lettuces.

He grows them in tanks without soil.

Water containing minerals is pumped through the tanks.

The diagram shows the equipment that Bill uses.



Bill keeps fish in a tank. Then he circulates the water through a filter containing bacteria.

The fish make ammonia. The bacteria convert the ammonia into nitrates.

- (i) Name the type of bacteria used in the filter.

..... [1]

- (ii) As well as nitrates, the lettuces need magnesium.

Explain why plants need magnesium.

..... [1]

- (iii) The air pump adds oxygen to the circulating water.

Oxygen is needed by the fish and the bacteria.

Oxygen is also needed by the lettuce roots.

Complete these sentences to describe why oxygen is needed by the roots.

Plant roots use oxygen in the process of to release energy.

Some of this energy is used to absorb minerals by a process called

.....

The energy is needed because the minerals are absorbed against a

.....

[3]

[Total: 7]

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3 Read this information about preserving food.



In the 18th century, Napoleon was travelling in hot countries with his army.

The hot conditions meant that their food rotted faster.

The food could be preserved by adding salt but this made the soldiers thirsty.

The French government offered a large sum of money to anybody who could find a solution.

Nicolas Appert solved the problem.

He put food into glass bottles. He used steam to heat the bottles to high temperatures.

He then removed the steam and quickly sealed each bottle with a cork and wax.

Appert won the prize and set up his own factory.

(a) Microbes caused Napoleon's food to rot or decompose.

(i) Give the name used to describe the type of nutrition shown by these microbes.

..... [1]

(ii) Explain why food rots much faster in hot countries.

..... [1]

(b) Before Appert devised his method, food could be preserved by adding salt.

Explain how adding salt preserves food.

.....

..... [2]

[Total: 4]

Section B – Module C4

4 Miles is doing the washing.

He washes a shirt.

(a) Miles washes his shirt at a low temperature.

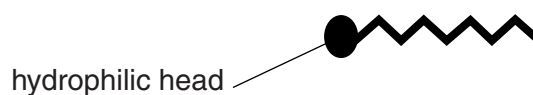
This saves him money.

Write down one **other advantage** of washing clothes at a low temperature.

..... [1]

(b) Miles uses a washing powder containing a detergent.

Look at the diagram of a detergent molecule.



Write about how the detergent removes grease from clothes.

Use ideas about the structure of the detergent molecule.

You may wish to draw a **labelled** diagram.

.....
.....
..... [3]

(c) Miles wants to clean his best suit.

He decides to have it dry cleaned.

(i) Explain what is meant by **dry cleaning**.

.....
..... [1]

(ii) Explain why some clothes are dry cleaned.

.....
..... [1]

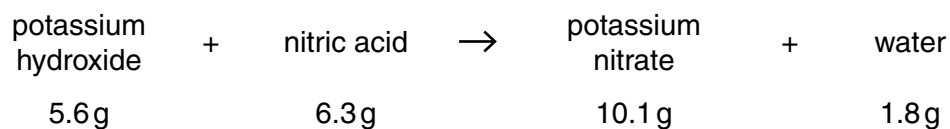
[Total: 6]

- 5 Kylie makes a fertiliser called potassium nitrate.

She adds an alkali called potassium hydroxide to nitric acid.

- (a) Kylie uses 5.6 g of potassium hydroxide and 6.3 g of nitric acid.

She makes 10.1 g of potassium nitrate.



Kylie does the experiment again.

This time she uses 16.8 g of potassium hydroxide and 18.9 g of nitric acid.

Calculate the mass of potassium nitrate she will make.

.....

answer g [1]

- (b) Kylie does the experiment a third time.

She expects to make 5.5 g of potassium nitrate.

She actually makes 4.4 g.

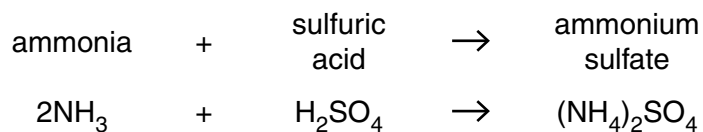
Show by calculation that her percentage yield is 80%.

.....

 [2]

- (c) Kylie makes another fertiliser called ammonium sulfate.

Look at the word and symbol equations for this reaction.



Calculate the mass of ammonium sulfate that can be made from 3.4g of ammonia.

Relative atomic masses, A_r

N = 14, H = 1, S = 32, O = 16.

.....

.....

.....

.....

answer g [3]

- (d) Most fertilisers contain the essential element nitrogen.

The nitrogen in fertilisers helps crops grow bigger.

Explain **how** nitrogen in fertilisers helps crops grow bigger.

.....

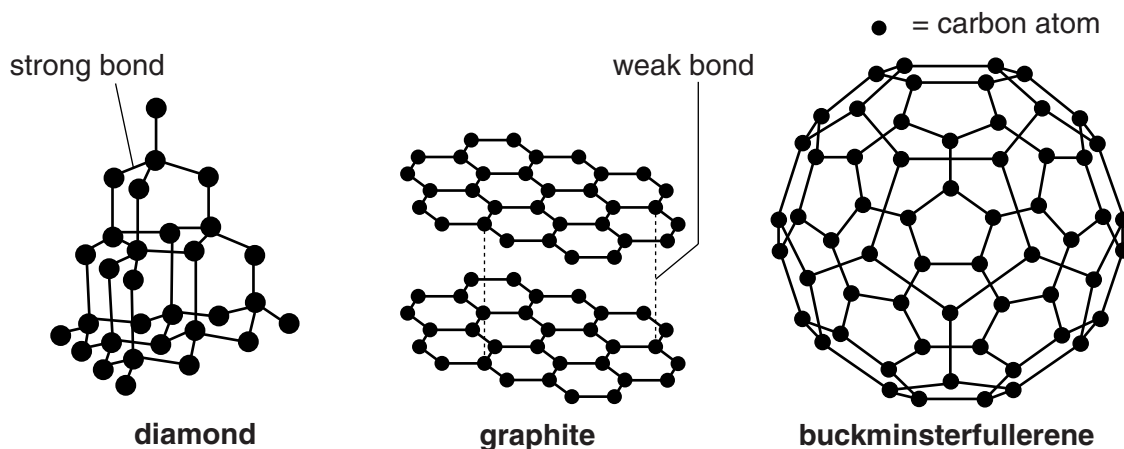
.....

..... [2]

[Total: 8]

6 Look at the diagrams.

They show the structures of diamond, graphite and buckminsterfullerene.



- (a) Diamond and graphite are giant structures.

Diamond and graphite have **high melting points**.

Explain why they have high melting points. Use the diagrams to help you.

.....
 [1]

- (b) Graphite is used to make lead in pencils.

Explain why.

.....

 [2]

- (c) Buckminsterfullerene has the formula C_x .

Write down the value of x.

Choose from

20

40

60

80

100

120

answer

[1]

(d) Buckminsterfullerene can be made into **nanotubes**.

Write down two **uses** of nanotubes.

1

2 [2]

[Total: 6]

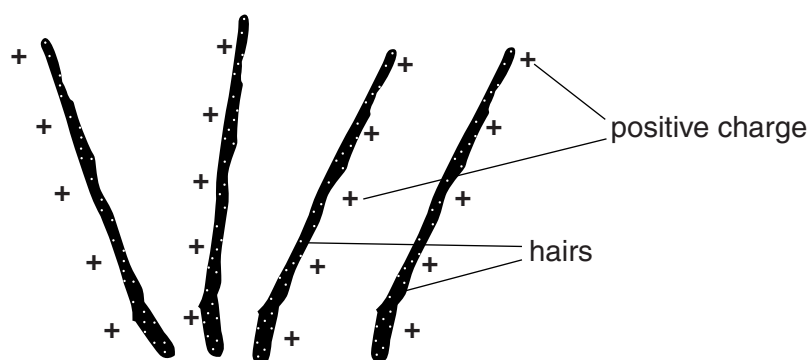
Section C – Module P4

- 7 This question is about static electricity.

Danni combs her hair.



Hairs become positively charged by friction and they repel each other.



- (a) (i) Why do the hairs repel each other?

.....
..... [1]

- (ii) Explain how the hairs have become **positively** charged.

.....
..... [1]

(b) What charge does the comb have after Danni combs her hair?

answer

Explain your answer.

.....

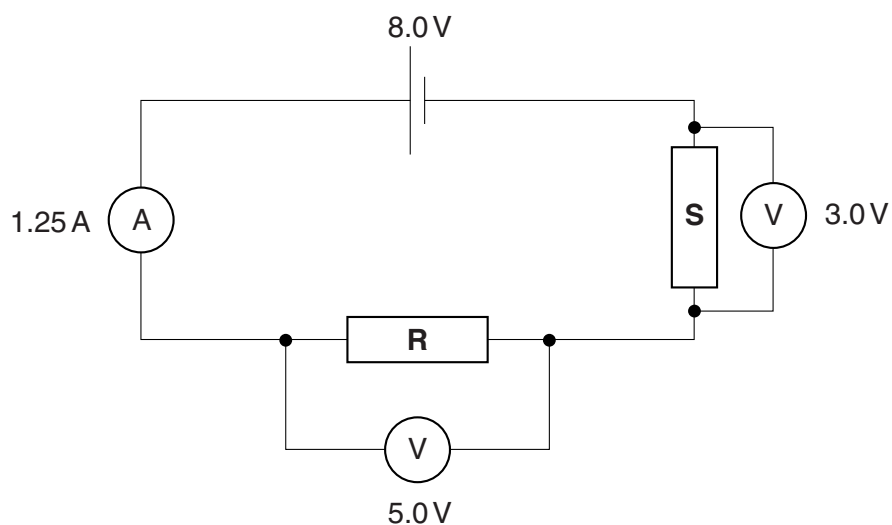
.....

..... [2]

[Total: 4]

8 Ramiz is investigating electrical circuits.

He builds a circuit with two resistors in it.



(a) Ramiz measures

- the current in the circuit
- the voltage across each resistor.

Calculate the **resistance** of resistor **R**.

The equations on page 2 may help you.

.....

.....

.....

resistance of **R** = ohms (Ω)

[2]

- (b) (i) Ramiz adds another identical cell to his circuit.

This doubles the supply voltage.

This increases the current flowing through the resistor.

What happens to the **value** of the **voltage** across resistor **R**?

.....
..... [1]

- (ii) He adds a third resistor to the circuit.

It is placed in series with the other resistors.

What happens to the current in the circuit?

.....
..... [1]

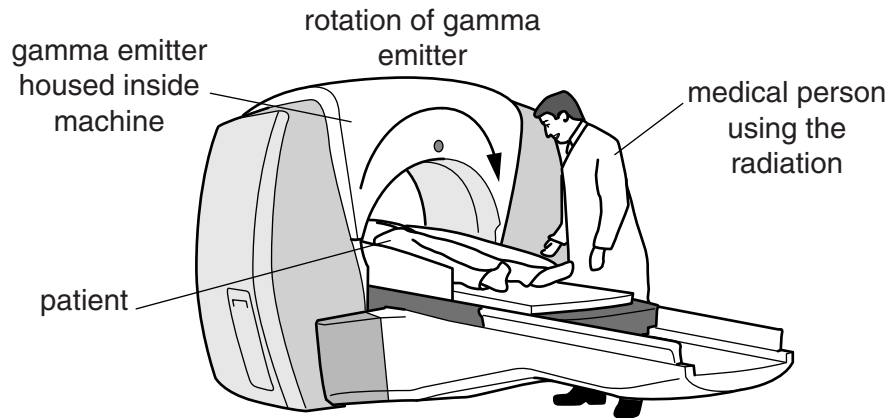
[Total: 4]

20
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9 Some radioactive isotopes emit gamma rays.

They are used to treat cancerous tumours.



(a) Explain how the gamma rays are used to treat the cancerous tumour.

In your answer, include ideas about

- the beam of gamma rays
- why the beam is rotated
- how damage to healthy human tissue is kept to a minimum.

.....

.....

.....

.....

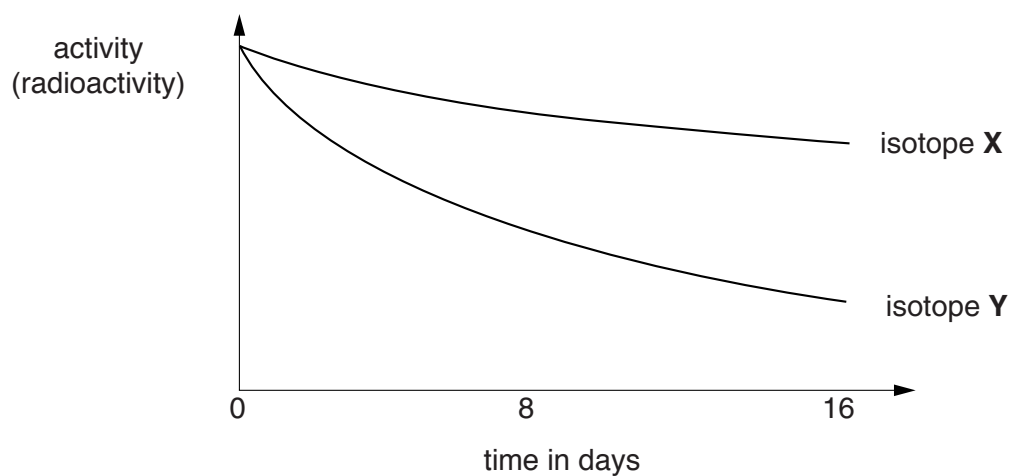
.....

..... [3]

(b) Radioisotopes are used as **tracers** in hospitals.

These isotopes have a relatively short **half-life**.

The graph shows the decay of two radioisotopes with different half-lives.



Explain

- what is meant by half-life
- **which** isotope would be the most suitable to use as a tracer. Give reasons for your answer.

The explanation has been started for you.

Half life is

.....

Isotope is the **most** suitable because

.....

.....

..... [2]

- (c) (i) Complete the sentences using phrases from this list.

a beta particle

a gamma ray

an X-ray

an alpha particle

An unstable atom emits a helium nucleus.

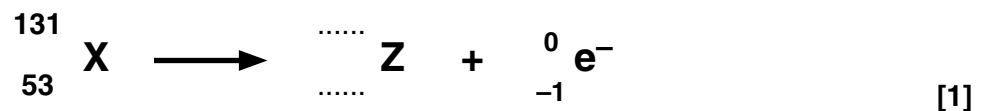
Scientists call this helium nucleus

An unstable atom emits a fast moving electron.

Scientists call this fast moving electron [2]

- (ii) When radioactive isotope **X** decays it emits a β particle to form a new element, **Z**.

Complete the nuclear equation for the decay of isotope **X**.



[Total: 8]

10 Doctors use **ultrasound** to treat people with kidney stones.

There are two parts to the treatment.

First the doctor locates the kidney stones. He then carries out the treatment.



(a) Complete the sentences to describe what happens.

The kidney stones are located by doing an ultrasound The waves strike the kidney stones and to the detector outside the body.

Different ultrasound waves are then focused on the kidney stones to the stones by vibrating them vigorously at a very high [2]

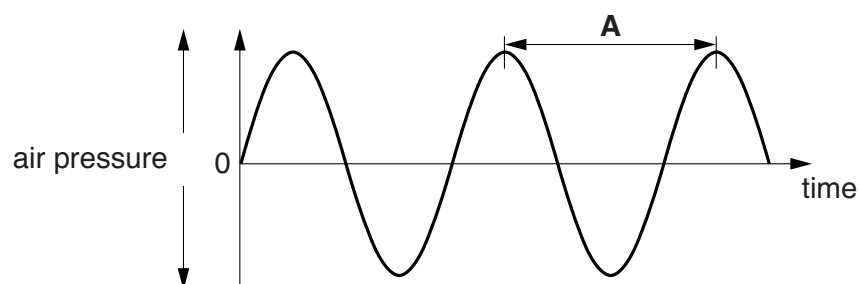
(b) Humans **cannot** hear ultrasound.

Put a tick (✓) next to the correct reason.

- | | |
|---|--------------------------|
| The ultrasound waves are not loud enough. | <input type="checkbox"/> |
| The ultrasound waves travel as transverse waves. | <input type="checkbox"/> |
| The ultrasound waves are just areas of air pressure changes. | <input type="checkbox"/> |
| The ultrasound waves do not have enough energy. | <input type="checkbox"/> |
| The ultrasound frequency is above the range of human hearing. | <input type="checkbox"/> |

[1]

(c) Ultrasound waves can be shown as a wave diagram.



The wave feature shown by the letter **A** is the **wavelength**.

Describe what is meant by the **frequency** of a wave.

.....
..... [1]

[Total: 4]

END OF QUESTION PAPER

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* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.