Time allowed: 1 hour 45 minutes



# GCSE COMBINED SCIENCE: SYNERGY



Foundation Tier Paper 2F

# Specimen 2018

# **Materials**

For this paper you must have:

- a ruler
- a calculator
- the periodic table (enclosed)
- the Physics equation sheet (enclosed).

## **Instructions**

- Answer all questions in the spaces provided.
- Do all rough work in this book. Cross through any work you do not want to be marked.

#### Information

- There are 100 marks available on this paper.
- The marks for questions are shown in brackets.
- You are expected to use a calculator where appropriate.
- You are reminded of the need for good English and clear presentation in your answers.
- When answering questions 05.4 and 10.1 you need to make sure that your answer:
  - is clear, logical, sensibly structured
  - fully meets the requirements of the question
  - shows that each separate point or step supports the overall answer.

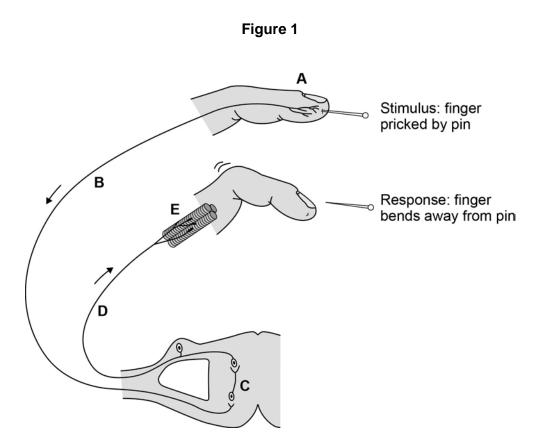
#### Advice

In all calculations, show clearly how you work out your answer.

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Centre number			J	C	an	did	ate	nu	mb	er <sup>l</sup>						
Surname																
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Our nervous system controls our reactions.

**Figure 1** shows the part of the nervous system involved in the rapid response to a stimulus.



0 1 . 1	What is this type of ra	pid response called?	[1 mark]
	Circular action Fast action		
	Forced action		
	Reflex action		

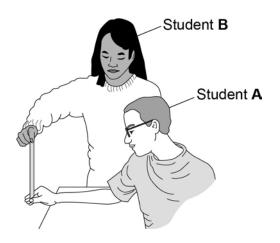
0 1 . 2	Features of the nervous sys	stem are labelled A, B, C, D and E on Figure 1.
	Draw <b>one</b> line from each fea	ature to the correct label from <b>Figure 1</b> .  [3 marks]
	Feature	Label
		Α
	Effector	В
	Relay neurone	С
	Sensory neurone	D
		E

Question 1 continues on the next page

Two students compare their reactions using a ruler.

This is the method used.

- 1. Student A sits with his elbow on a table top.
- 2. Student **B** holds the ruler so the bottom of the ruler is level with the top of student **A**'s thumb.
- 3. Student **B** drops the ruler.
- 4. Student A catches the ruler.
- 5. Record the drop distance.
- 6. Repeat steps 1 to 5 four more times.
- 7. Repeat the whole experiment with student **A** dropping the ruler and student **B** catching it.



Both students are right-handed.

The students are testing the hypothesis:

the drop distance of the ruler is smaller when a right-handed person uses their right hand to catch the ruler.

Student **A** uses his right hand to catch the ruler.

Student **B** uses her left hand to catch the ruler.

0	1	3	Complete the sentence.

Use an answer from the box.

[1 mark]

	control	dependent	independent	
The drop of	distance was t	he		variable.

Question 1 continues on the next page

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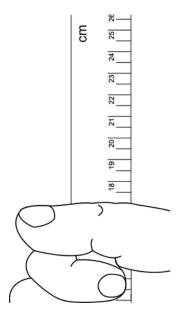
**Table 1** shows the students' results.

Table 1

Student	Drop distance in cm							
Student	Test 1	Test 2	Test 3	Test 4	Test 5			
Student A	17.5	15.5	15.0	23.5	17.0			
Student B	20.5		19.5	21.0	19.0			

0 1 . 4 Figure 2 shows student B's Test 2 result.

Figure 2



Use Figure 2 to complete the missing result for Test 2.

Write the answer in Table 1.

[1 mark]

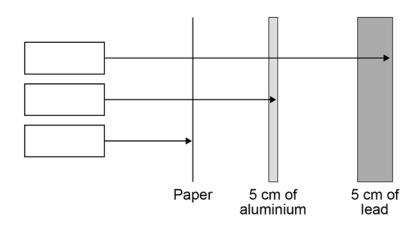
0 1 . 5	What was the resolution of the ruler the students used?	[1 mark]
	Tick <b>one</b> box.	[1 mark]
	0.1 cm	
0 1 . 6	One of the results in <b>Table 1</b> is anomalous.	
	Identify the anomalous result.	
	Give the reason why you chose your answer.	[2 marks]
0 1 . 7	The students are testing the hypothesis:	
	the drop distance of the ruler is smaller when a right-hand their right hand to catch the ruler.	ed person uses
	The results in <b>Table 1</b> are not a good test of the hypothesis.	
	What is one reason why?	[4 mark]
	Tick <b>one</b> box.	[1 mark]
	The drop distances are very variable	
	The drop distance for Student <b>A</b> is sometimes bigger than the drop distance for Student <b>B</b>	
	The results are for the left and right hands of different people	
	The drop distances are not measured accurately enough	

**0 2** Alpha, beta and gamma are types of nuclear radiation.

A teacher sets up a demonstration of the penetration properties of alpha, beta and gamma radiation.

Figure 3 shows the demonstration.

Figure 3



0 2 . 1 Complete **Figure 3** by writing the name of the radiation in each box.

[2 marks]

0 2 . Give **two** safety precautions the teacher should take in the demonstration.

[2 marks]

1

2

Table 2 shows how the count rate of a radioactive source changes with time.

Table 2

Time in seconds	0	40	80	120	160
Count rate in counts/second	600	463	300	221	150

0 2 . 3	Describe the relationship shown in <b>Table 2</b> .	[2 marks]
0 2 . 4	Use <b>Table 2</b> to predict the count rate after 200 seconds.	[2 marks]
	Count rate =	counts/second

Question 2 continues on the next page

0 2 . 5	The half-life of the radioactive source is very short.	
	Give <b>one</b> reason why the source would be much less hazardous after 800 seconds.	[1 mark]

0 3	Blood sugar levels in the body are controlled by insulin.	
0 3 .	1 How does insulin travel around the body?	[1 mark]

Question 3 continues on the next page

0 3 . 2 Table 3 shows the blood sugar levels for two people after eating a meal.

Table 3

Time after eating	Blood sugar levels in mg per 100 cm <sup>3</sup> of blood				
in hours	Person A	Person B			
0	70	130			
1	150	230			
2	90	185			
3	80	165			
4	75	140			

Use data from Table 3 to complete the graph in Figure 4.

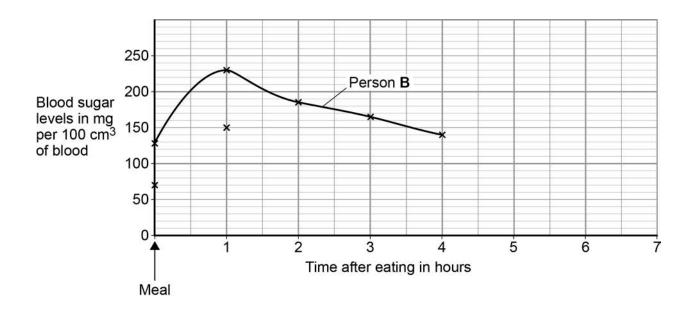
Plot the points for person A.

The first two points have been plotted for you.

Draw a line through all the points.

[3 marks]

Figure 4



0 3 . 3	How long after the meal is person <b>B</b> 's insulin production at its peak?  [1 mark]
0 3 . 4	What is the greatest <b>decrease</b> in the blood sugar level of person <b>B</b> in an hour?
	[2 marks]
	Decrease = mg per 100 cm <sup>3</sup>
0 3 . 5	Estimate how long after eating the meal it will take for person <b>B</b> 's blood sugar level to return to the level before the meal.
	Show your working on <b>Figure 4</b> . [2 marks]

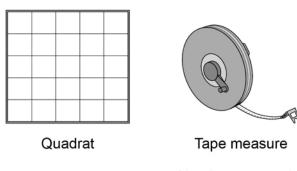
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0 4

A student investigated the number of ribwort plants in a field.

The student used the apparatus shown in **Figure 5**.

Figure 5



Not drawn to scale

This is the method used.

- 1. Place the quadrat in an area where there are lots of ribwort plants in the field.
- 2. Count the number of ribwort plants inside a quadrat.
- 3. Repeat steps 1 and 2 four more times.

0 4 . 1	How could the student improve his method	so that he can collect valid results?  [2 marks]
	Tick <b>two</b> boxes.	
	Count the leaves of each ribwort plant	
	Place more quadrats in the field	
	Place the quadrats randomly	
	Use a smaller quadrat	
	Weigh the ribwort plants	

0 4 . 2 The student calculated that the mean number of ribwort plants per m² was 3.2

The area of the field was 8250 m<sup>2</sup>.

Calculate the total number of ribwort plants in the field.

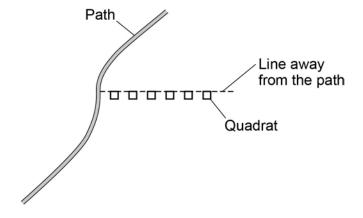
[1 mark]

Total number of ribwort plants =

Another group of students did an investigation in the field.

Figure 6 shows how the students placed their quadrats in this investigation.

Figure 6

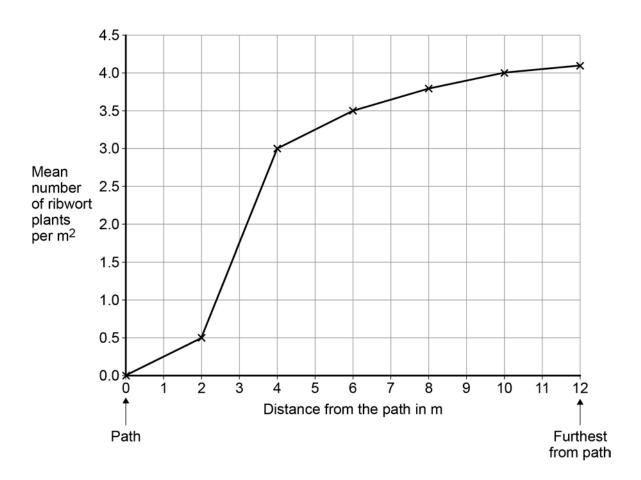


0 4 . 3 What is the name given to the line in Figure 6?

[1 mark]

Figure 7 shows the students' results.

Figure 7



0 4 . 4	Describe the relationship shown in <b>Figure 7</b> .	[2 marks]

0 4 . 5	What is one reason why there are no ribwort plants	next to the path?	[1 mark]
	Tick <b>one</b> box.		[
	There is less light near the path		
	The ribwort plants get walked on		
	There are more nutrients in the soil near the path		
	There are fewer animals near the path		

Turn over for the next question

10
There are no questions printed on this page

0 5	Millions of years ago the Earth's atmosphere was probably like the atmosphere of Mars today.

**Table 4** shows percentages of the main gases in the atmospheres of Earth and Mars today.

Table 4

Gas	Percentage in atmosphere of Mars today	Percentage in atmosphere of Earth today
Carbon dioxide	95.00	0.04
Nitrogen	3.50	78.00
Oxygen	0.50	21.00

For each gas in the table, suggest a reason for the change in the percentage of the gas in Earth's atmosphere.

Carbon dioxide

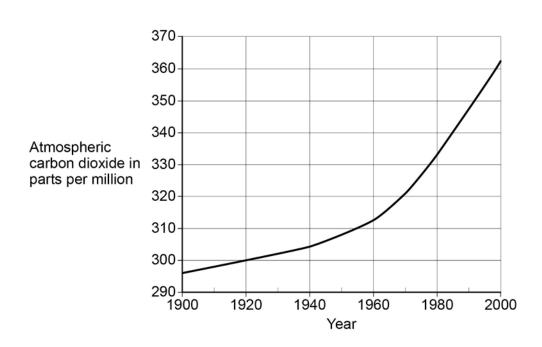
Nitrogen

Oxygen

Question 5 continues on the next page

**Figure 8** shows how the concentration of carbon dioxide in the Earth's atmosphere changed between 1900 and 2000.





0 5 . 2 One of the causes of the increase in carbon dioxide between 1900 and 2000 is increased use of fossil fuels.

Suggest when use of fossil fuels began to increase.

Use data from Figure 8 to explain your answer.

		_	_
[2	ma	ırk	œ1

0 5 . 3	What is the percentage increase in carbon dioxide levels between 1970 and 2000?  [3 marks]		
	Ingrance -	%	
	Increase =	/0	
0 5 . 4	Explain how the changes shown in <b>Figure 8</b> can have harmful effects on the environment.	[4 marks]	

Turn over for the next question

0 6

A person's characteristics can be due to:

- environmental causes
- genetic causes
- both environmental and genetic causes.
- 0 6 . 1 Complete Table 5.

Put a tick to show what each characteristic is due to.

[3 marks]

# Table 5

	Characteristic due to			
Characteristic	Environmental causes	Genetic causes	Both environmental and genetic causes	
Eye colour				
A scar				
Weight				

0 6 . 2 Draw **one** line from each key term to the correct definition.

[2 marks]

Key term	Definition
	The set of alleles for a characteristic
Genotype	The genus of an organism
	The inheritance of chromosomes
Phenotype	The mutation of genes
	The physical characteristics of an organism

Question 6 continues on the next page

0 6 . 3 Farmers use selective breeding to control the characteristics in cows.

**Table 6** shows the stages of selective breeding in cows.

Complete **Table 6** to show the correct order of the stages.

The first stage has been numbered for you.

[2 marks]

Table 6

Stage in selective breeding	Order of stage
Cows are bred over many generations	
Parents are bred together	
Cows with the desired characteristics are chosen	1
Calves with the most desired characteristics are bred together	

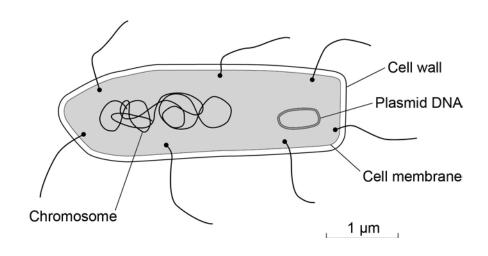
0 6 . 4	Farmers selectively breed cows for many different reasons.				
	Suggest <b>two</b> characteristics that cows may be bred for.				
	Do <b>not</b> suggest coat colour. [2 marks]				
	1				
	2				
	2				
0 6 . 5	Selective breeding can lead to problems.				
	Suggest how problems caused by selective breeding in cows can have negative financial effects for the farmer.				
	[2 marks]				

Turn over for the next question

**0 7** Bacteria can cause disease.

Figure 9 shows some features of a Salmonella bacterium.

Figure 9



0 7 . 1 Draw **one** line from each feature of the *Salmonella* bacterium to the function.

[2 marks]

Feature	Function	
	Controls the movement of substances into and out of the cell	
Cell membrane	Carries genetic information	
Plasmid DNA	Provides support and protection	
Flasifild DIVA		
	The site of protein synthesis	

0 7 . 2	How is Salmonella spread between people?		
	Tick <b>one</b> box.		[1 mark]
	Animal bites		
	Contaminated food		
	Sneezing		
	Sexual contact		
0 7 . 3	Give <b>two</b> ways you could stop Salr	monella from spreading.	[2 marks]
	1		
	2		

Question 7 continues on the next page

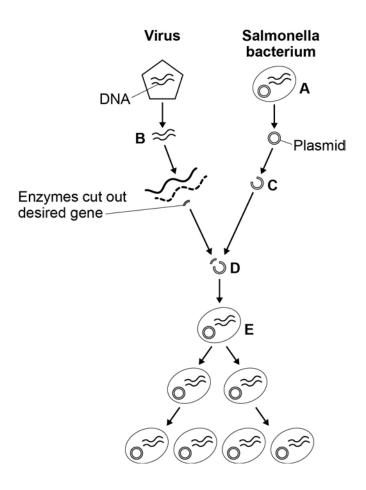
accine to n a tablet.
Y
[1 mark]
[1

0 7 . 5

The *Salmonella* bacterium used in the vaccine is genetically modified using part of a virus.

Look at Figure 10.

Figure 10



Complete the sentences.

Use the letters from **Figure 10**.

[3 marks]

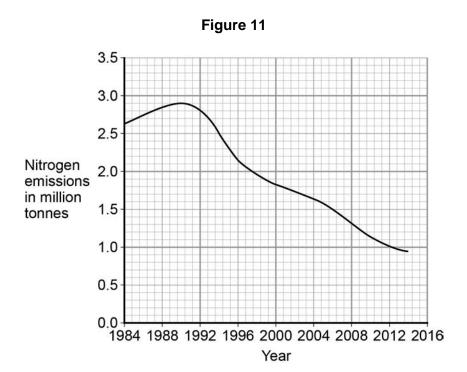
Bacteria reproduce quickly in part

DNA with the desired gene is removed from the virus in part .

The chosen gene is inserted into the plasmid in part

(	0 8	3	Oxides of nitrogen are produced when fuels are burnt.	
	0   8	3 . 1	Write a balanced symbol equation for the production of nitrogen dioxide (NO <sub>2</sub> ) from nitrogen and oxygen.	

0 8 . 2 Figure 11 gives information about emissions of oxides of nitrogen in the UK.



Calculate the percentage decrease in emissions of oxides of nitrogen from 1990 to 2014.

Give your answer to three significant figures.		
Percentage decrease =	%	

0 8 . 3 Give one advantage of reducing the emissions of oxides of nitrogen.

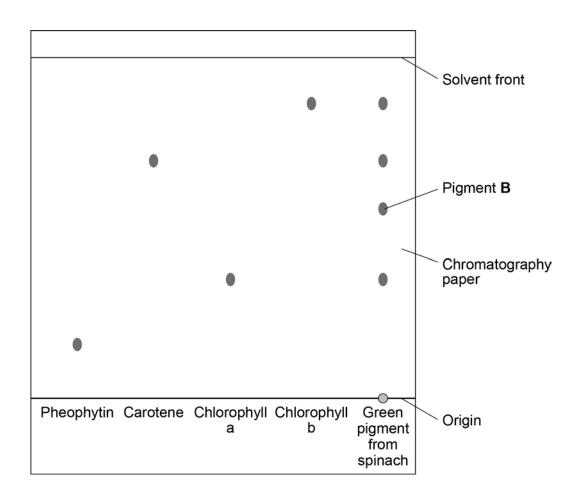
[1 mark]

**0 9** A student used paper chromatography to identify the pigments in spinach leaves.

She used propanone as a solvent.

Figure 12 shows the student's results.

Figure 12



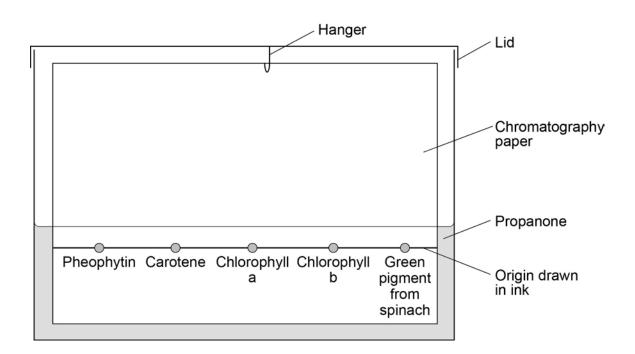
0 9 . 1	Name the mobile ph	ase and the stationary phase in the student's experiment.
		[2 marks]
	Mobile phase	
	Stationary phase	

0 9 . 2	What does Figure 12 tell you about the green pigment from spinach?	[3 marks]
0 9 . 3	Write the equation that links distance moved by solvent, distance moved by solute and $R_{\rm f}$ value.	[1 mark]
0 9 . 4	Use <b>Figure 12</b> to calculate the R <sub>f</sub> value for pigment <b>B</b> .	[3 marks]
	R <sub>f</sub> value =	
	· ———	

Question 9 continues on the next page

0 9 . 5 Another student set up the apparatus shown in Figure 13.

Figure 13



This student did not set up the apparatus correctly.

Identify the errors the student made.

Explain how the errors she made would affect her results.	[4 marks]

# Turn over for the next question

1 0 As	A student investigated the specific heat capacity of metals.		
10.1	Describe an experiment the student could do to measure the specific heat capacity of a metal.	[6 marks]	

The student calculated the specific heat capacity of four metals.

Table 7 shows the student's results.

Table 7

Metal	Mass of material in kg	Time in minutes	Temperature change in °C	Change in thermal energy in J	Calculated specific heat capacity of material in J/kg °C
Aluminium	1	10	2	4 780	2 390
Brass	1	10	4	4 660	1 165
Copper	1	10		4 600	657
Steel	1	10	5	4 690	938

1 0 . 2	Use data from <b>Table 7</b> to calculate the temperature Use the correct equation from the Physics Equation		s]
	Temperature change	=	°C
1 0 . 3	What is the independent variable in the student's involved one box.	estigation? [1 mar	k]
	Mass of material		
	Power used		
	Time in minutes		
	Type of material		

1 0 . 4	The student calculated the specific heat capacity of aluminium to be 2390 J/kg °C.
	The 'true' specific heat capacity of aluminium is 900 J/kg °C.
	Suggest why the student's result for aluminium is different from the 'true' value.  [2 marks]
1 0 . 5	The teacher suggested that putting bubble wrap round the metal block would change the results.
	How would using bubble wrap change the results?
	Give a reason for your answer.  [2 marks]

# **END OF QUESTIONS**

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