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for the guidance of teachers

9701 CHEMISTRY

9701/31

Paper 31 (Advanced Practical Skills 1), maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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Cambridge is publishing the mark schemes for the May/June 2012 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

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Page 2	Mark Scheme: Teachers' version	Syllabus
	GCE AS/A LEVEL – May/June 2012	9701

Page 2	e 2 Mark Scheme: Teachers' version Syllabus GCE AS/A LEVEL – May/June 2012 9701			_
Question	Sections	Indicative material	Cam	6.
1 (a)	PDO layout	I Constructs a table for results	1	7
-	PDO recording	cheme: Teachers' version Syllabus JA LEVEL – May/June 2012 9701 Indicative material 9701 I Constructs a table for results II Appropriate headings and units for data given. Volume / V in cm ³ , / cm ³ or (cm ³) Time/t in seconds, /s or (s)	1	
	PDO recording	III All times recorded to the nearest second.	1	
	MMO decision	IV 3 additional volumes chosen with intervals not less than 2.00 cm ³ and all volumes of FA 1 greater or equal to 6.00 cm ³	1	
	MMO collection	V In all 3 additional experiments water is added to make a total of 20.00 cm ³	1	
	MMO quality	Round times to nearest second. VI + VII Compare time for 20.00 cm ³ of FA 1 with that of supervisor.	2	
		VIII + IX Compare time for 10.00 cm ³ of FA 1 with that of supervisor. The range for award of 1 or 2 depends on the supervisor value.	2	
		Supervisor value: < or = 15δ for 2 is 2 and for 1 is 4 16 to 30 δ for 2 is 3 and for 1 is 6 31 to 45 δ for 2 is 4 and for 1 is 8 46 τ o 60 δ for 2 is 5 and for 1 is 10 > 60 δ for 2 is 6 and for 1 is 12		[9]
(b)	PDO display	(i) Working to show ans = 5×10^{-5} mol	1	
	ACE interpretation	(ii) 0.5 x ans to (b)(i) = 2.5 × 10 ⁻⁵ mol	1	
	PDO display	(iii) Working to show that: $(2.5 \times 10^{-5}) / 0.050 =$ $(5 \times 10^{-4} \text{ mol dm}^{-3})$	1	[3]
(c)	ACE interpretation	Rate correctly calculated using ans (b)(iii) / time (or 4.25×10^{-4}). Min 2 s.f. rounded correctly and minimum 4 results.	1	
	PDO recording	Unit for rate given as mol dm ⁻³ s ⁻¹ .	1	[2]

Page 3 Mark Scheme: Teachers' version Syllabus			Mary .
	Page 3	Mark Scheme: Teachers' version	Syllabus A
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luestion	Sections	Indicative material	17	3
(d)	PDO layout	I Rate on <i>y</i> -axis and volume on <i>x</i> -axis. Axes clearly labelled (ignore units)	1 1	1
		 Linear scale chosen to use at least half of each axis (need not include 0, 0) If no point at 0, 0 cannot count for > half. 	1	
		III Plotting of points. Minimum of 3 readings.	1	
		IV Draws a line of best fit. Minimum 4 readings including 0, 0 (if plotted).	1	[4]
(e)	ACE conclusion	Rate is proportional to peroxodisulfate concentration Rate increases as concentration (volume) increases would score one	2	[2]
(f)	ACE interpretation	 (i) correctly calculates (0.5 / time from Expt 1) × 100. Minimum of 2 s.f. 	1	
		(ii) ans (b)(iii) Expt 1 time + 0.5 × 10 ⁶ mol dm ⁻³ s ⁻¹ or Rate– (% from (i) × rate)	1	
		(iii) Any reasonable suggestion e.g. difficult to judge colour change / measurement of volumes / variation in T	1	
	ACE improvement	use of colorimeter / burettes for all volumes / (thermostatic) waterbath. Not air conditioning.	1	[4]
(g)	ACE conclusion	 (ii) Thiosulfate concentration / number moles / volume is doubled (1) Time is longer/ reaction is slower with more thiosulfate (1) 	2	[2]

Page 4 Mark Scheme: Teache GCE AS/A LEVEL – Ma			ners' version Syllabus			r		
Question	See	ctions		dicative ma	aterial	Syllabus 9701 9701 9701 9701 9701 9701 9701 9701	Car	non
FA	5 =	CuC <i>l</i> ₂ ; FA 6 = NaOH;	FA 7 = Pb(NO ₃) ₂ ; FA	8 = K ₂ CrO	4; FA 9 = MgSO ₄		
2 (a)	MM	10 collection	in: ex	ue ppt sol in cess (1) ot 'dark ue'	White pp (1) Ignore 'excess'. White pp soluble in excess (greenish- brown ppt (1) Not 'orange, red, red / brown' Ignore excess. Mo reaction n / yellow		[5]
(b) A	AC	E conclusion	Cı	u ²⁺ in FA 5	AND CrC	04 ^{2–} in FA 8	1	
			Pt	²⁺ in FA 7	AND OH	in FA 6	1	
			Ci	[–] in FA 5			1	[3]
1 F 1	ΜM	1O decision	I	Add Pb (N	IO ₃) ₂ or Ba	aCl_2 or $Ba(NO_3)_2$	1	
	ΜN	1O decision	Ш	Add HNO	₃ or HC <i>l</i>		1	
	PD	O recording		 III Presents observations in a single table – no extra reagents. Must be > 2 'boxes' 		1		
	ΜN	1O collection	IV	White ppt			1	
	ΜM	1O collection	v	No SO ₂ ev	volved or p	opt insoluble	1	
	AC	E conclusion	VI	sulfate			1	[6]