

Cambridge International AS & A Level

CHEMISTRY

Paper 3 Advanced Practical Skills 1

CONFIDENTIAL INSTRUCTIONS

9701/35

May/June 2023

This document gives details of how to prepare for and administer the practical exam.

The information in this document and the identity of any materials supplied by Cambridge International are confidential and must NOT reach candidates either directly or indirectly.

The supervisor must complete the report at the end of this document and return it with the scripts.

INSTRUCTIONS

If you have any queries regarding these confidential instructions, contact Cambridge International stating the centre number, the syllabus and component number and the nature of the query.
 email info@cambridgeinternational.org
 phone +44 1223 553554

General information about practical exams

Centres must follow the guidance on science practical exams given in the Cambridge Handbook.

Safety

Supervisors must follow national and local regulations relating to safety and first aid.

Only those procedures described in the question paper should be attempted.

Supervisors must inform candidates that materials and apparatus used in the exam should be treated with caution. Suitable eye protection should be used where necessary.

The following hazard codes are used in these confidential instructions, where relevant:

- **C** corrosive
- **HH** health hazard**F** flammable

- MH moderate hazard
- T acutely toxic
- **O** oxidising
- N hazardous to the aquatic environment

Hazard data sheets relating to substances used in this exam should be available from your chemical supplier.

Before the exam

- The packets containing the question papers must **not** be opened before the exam.
- It is assumed that standard school laboratory facilities, as indicated in the *Guide to Planning Practical Science*, will be available.
- Spare materials and apparatus for the tasks set must be available for candidates, if required.

During the exam

- It must be made clear to candidates at the start of the exam that they may request spare materials and apparatus for the tasks set.
- Where specified, the supervisor must perform the experiments and record the results as instructed. This must be done out of sight of the candidates, using the same materials and apparatus as the candidates.
- Any assistance provided to candidates must be recorded in the supervisor's report.
- If any materials or apparatus need to be replaced, for example, in the event of breakage or loss, this must be recorded in the supervisor's report.

After the exam

- The supervisor must complete a report for each practical session held and each laboratory used.
 - Each packet of scripts returned to Cambridge International must contain the following items:
 - the scripts of the candidates specified on the bar code label provided
 - the supervisor's results relevant to these candidates
 - the supervisor's reports relevant to these candidates
 - seating plans for each practical session, referring to each candidate by candidate number
 - the attendance register.

Specific information for this practical exam

During the exam, the supervisor (not the invigilator) must do all the experiments and record the results on a spare copy of the question paper, clearly labelled 'supervisor's results'.

If chemicals are prepared in more than one batch, clearly labelled supervisor's results must be provided for each batch. The candidates using each batch must be listed on the supervisor's report.

Apparatus

The apparatus listed must be provided to each candidate.

 $1 \times 50 \,\mathrm{cm^3}$ measuring cylinder

 $1 \times 250 \, \text{cm}^3$ plastic or glass measuring cylinder

 $1 \times 250 \,\mathrm{cm^3}$ beaker

 $1 \times 100 \,\mathrm{cm^3}$ beaker

2 × stand and clamp

 $1 \times \text{tub}$ suitable for acting as a trough (minimum capacity 1 dm^3)

 $1 \times$ side-arm conical flask (labelled 'X') with bung connected to approximately 50 cm of plastic/rubber delivery tube OR additional 150 cm³ or 100 cm³ conical flask (labelled 'X') with one-hole bung connected to approximately 50 cm of plastic/rubber delivery tube

 $1 \times$ thermometer (-10 °C to 110 °C at 1 °C)

 $1 \times \text{plastic or cardboard cup, capacity approximately } 150 \,\text{cm}^3$

1 × glass rod

- 1 × stop-clock to measure to an accuracy of 1 second
- 2 × teat/dropping pipette

1 × spatula

1 × test-tube holder

1 x hard-glass test-tube

8 × test-tube*

1 × test-tube rack

balance, single-pan, direct reading, minimum accuracy 0.01g (1 per 8-12 candidates) weighing to 200g 1 × wash bottle containing distilled water

1 × pen for labelling glassware

paper towels

red and blue litmus papers

aluminium foil

wooden splints

the apparatus normally used in the centre for use with limewater in testing for carbon dioxide

*Candidates are expected to rinse and reuse test-tubes where possible.

Additional test-tubes should be available.

Materials
The materials listed in the table must be provided to each candidate. Materials must be labelled only as specified in the 'label' column. The identities of chemicals labelled with letter codes, e.g. FA 1, may be different from their descriptions in the question paper. Candidates must use the descriptions given in the question paper. For example, candidates may be supplied with sulfuric acid, labelled as FA 1, but be told in the question paper that FA 1 is phosphoric acid.

label	per candidate	identity	notes
FA 1	70 cm ³	1.0 mol dm ⁻³ hydrochloric acid	Dilute 500 cm ³ of 2.0 mol dm ^{-3} HC <i>l</i> to 1.00 dm ³ with distilled water. For 2.0 mol dm ^{-3} HC <i>l</i> see preparation instructions in the current syllabus.
FA 2 [F] For experiment 1	0.20 ± 0.02g	magnesium ribbon	Provide 0.20 ± 0.02 of magnesium ribbon [F] in a container labelled 'For experiment 1'. The magnesium ribbon must be from the same source for the supervisor and all candidates within a group.
FA 3	40 cm ³	2.0 mol dm ⁻³ hydrochloric acid	See preparation instructions in the current syllabus.
FA 4 [F] For experiment 2	0.40 ± 0.02g	magnesium ribbon	Provide 0.40 ± 0.02 g of magnesium ribbon [F] cut into 1 cm strips in a container labelled 'For experiment 2'. The magnesium ribbon must be from the same source for the supervisor and all candidates within a group.
FA 5	0.5 g	magnesium carbonate	Provide 0.5 \pm 0.1 g of MgCO ₃ in a stoppered container. Any basic magnesium carbonate is suitable.
FA 6 [MH]	10 cm ³	0.10 mol dm ⁻³ iron(II) sulfate and 0.10 mol dm ⁻³ potassium iodide in aqueous sulfuric acid	Dissolve 27.8 \pm 0.2g of FeSO ₄ •7H ₂ O [MH] and 16.6 \pm 0.2g KI in approximately 50 cm ³ of 1.0 mol dm ⁻³ H ₂ SO ₄ [MH] and dilute to 1.0 dm ³ with distilled water. For 1.0 mol dm ⁻³ H ₂ SO ₄ see preparation instructions in the current syllabus.
starch	5 cm ³	2% starch	See preparation instructions in the current syllabus.
hydrogen peroxide	5 cm ³	10 vol aqueous hydrogen peroxide	Dilute 100 cm ³ 100 vol H ₂ O ₂ [C] to 1.0 dm ³ or dilute 500 cm ³ 20 vol H ₂ O ₂ [MH] to 1.0 dm ³ .

Igodi	candidate	identity	notes
dilute hydrochloric acid	10 cm ³	$2.0 \mathrm{mol}\mathrm{dm}^{-3}\mathrm{HC}l$	
	10 cm ³	$2.0 \text{mol}\text{dm}^{-3}\text{HNO}_3$	
dilute sulfuric acid [MH]	10 cm ³	$1.0 \text{mol}\text{dm}^{-3}\text{H}_2\text{SO}_4$	See preparation instructions in the current
aqueous ammonia [C][MH][N]	20 cm ³	2.0 moldm ⁻³ NH ₃	syllabus.
aqueous sodium hydroxide [C]	20 cm ³	2.0 moldm ⁻³ NaOH	If necessary, each of these reagents can be
aqueous barium chloride or	10 cm ³	$0.1 \mathrm{mol}\mathrm{dm}^{-3}\mathrm{BaC}l_2$ or	provided as a communal supply for groups of up to 6 candidates.
aqueous barium nitrate		$0.1 \text{ mol dm}^{-3} \text{ Ba}(\text{NO}_3)_2$	Invigilators must be alert to the risk of
	10 cm ³	saturated aqueous calcium hydroxide, Ca(OH) ₂	
aqueous silver nitrate	10 cm ³	$0.05 \text{mol}\text{dm}^{-3}\text{AgNO}_3$	
acidified aqueous potassium manganate(VII) [MH]	10 cm ³	0.01 moldm ⁻³ KMnO ₄ in 0.5 moldm ⁻³ H ₂ SO ₄	

An excess of at least 10% of each material must be prepared to cover accidental loss.

5

All solutions must be thoroughly mixed.

•

•

If you are unable to source any of these chemicals, you must contact Cambridge International as far as possible in advance of the exam for advice. •

BLANK PAGE

6

Permission to reproduce items where third-party owned material protected by copyright is included has been sought and cleared where possible. Every reasonable effort has been made by the publisher (UCLES) to trace copyright holders, but if any items requiring clearance have unwittingly been included, the publisher will be pleased to make amends at the earliest possible opportunity.

To avoid the issue of disclosure of answer-related information to candidates, all copyright acknowledgements are reproduced online in the Cambridge Assessment International Education Copyright Acknowledgements Booklet. This is produced for each series of examinations and is freely available to download at www.cambridgeinternational.org after the live examination series.

Cambridge Assessment International Education is part of Cambridge Assessment. Cambridge Assessment is the brand name of the University of Cambridge Local Examinations Syndicate (UCLES), which is a department of the University of Cambridge.

Supervisor's report

Syllabus and component number			/	
Centre number				
Centre name	 	 		

Time of the practical session	

Laboratory name/number

Give details of any difficulties experienced by the centre or by candidates (include the relevant candidate names and candidate numbers).

You must include:

- any difficulties experienced by the centre in the preparation of materials
- any difficulties experienced by candidates, e.g. due to faulty materials or apparatus
- any specific assistance given to candidates.

If chemicals have been prepared in more than one batch, list the candidates using each batch.

Declaration

- 1 Each packet that I am returning to Cambridge International contains all of the following items:
 - the scripts of the candidates specified on the bar code label provided
 - the supervisor's results relevant to these candidates
 - the supervisor's reports relevant to these candidates
 - seating plans for each practical session, referring to each candidate by candidate number
 - the attendance register.
- 2 Where the practical exam has taken place in more than one practical session, I have clearly labelled the supervisor's results, supervisor's reports and seating plans with the time and laboratory name/number for each practical session.
- 3 I have included details of difficulties relating to each practical session experienced by the centre or by candidates.
- 4 I have reported any other adverse circumstances affecting candidates, e.g. illness, bereavement or temporary injury, directly to Cambridge International on a *special consideration form*.

Signed	 /isor)
Name (in block capitals)	