

1. June/2022/Paper\_41/No.6(b, f, g)

(b) Asparagine is an amino acid that contains a chiral carbon atom and displays stereoisomerism.

Separate samples of asparagine are dissolved in  $\text{CDCl}_3$  and analysed using carbon-13 and proton ( $^1\text{H}$ ) NMR spectroscopy.

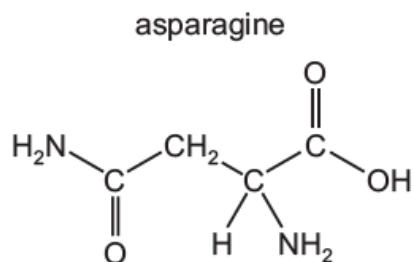
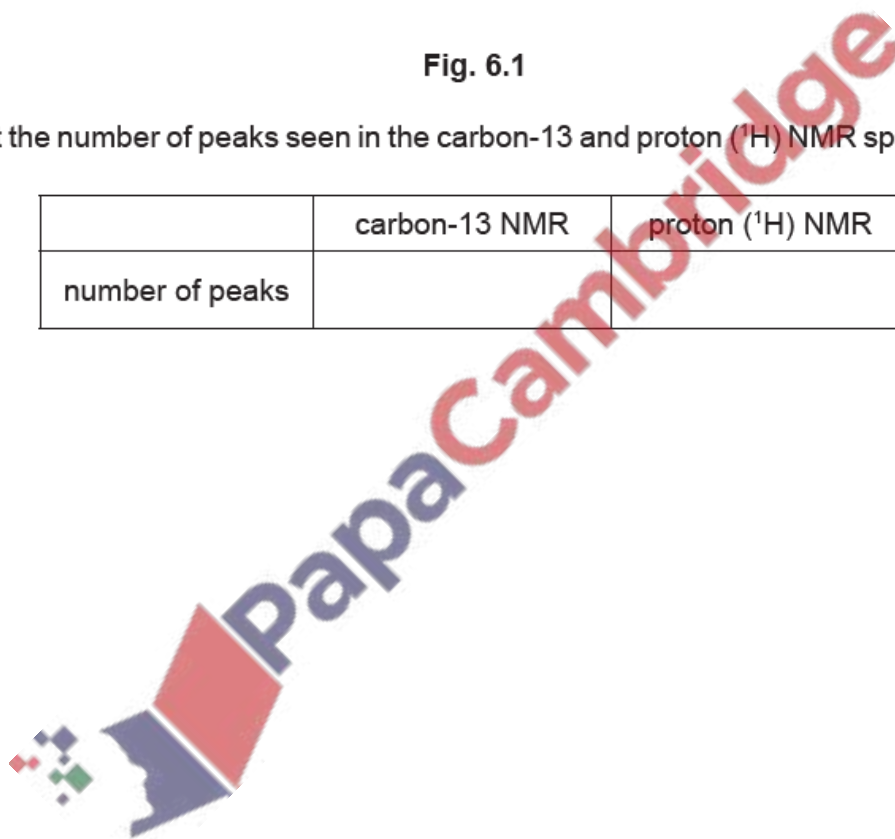


Fig. 6.1

Predict the number of peaks seen in the carbon-13 and proton ( $^1\text{H}$ ) NMR spectra of asparagine.

	carbon-13 NMR	proton ( $^1\text{H}$ ) NMR
number of peaks		

[1]



(f) Thin-layer and gas-liquid chromatography can be used to analyse mixtures of substances.

Each type of chromatography makes use of a stationary phase and a mobile phase.

(i) Complete Table 6.1 with an example of each of these.

Table 6.1

	stationary phase	mobile phase
thin-layer chromatography		
gas-liquid chromatography		

[1]

(ii) An unknown amino acid is analysed using thin-layer chromatography. Two chromatographs of the unknown amino acid and four reference amino acids, P, Q, R and S, are obtained using two different solvents.

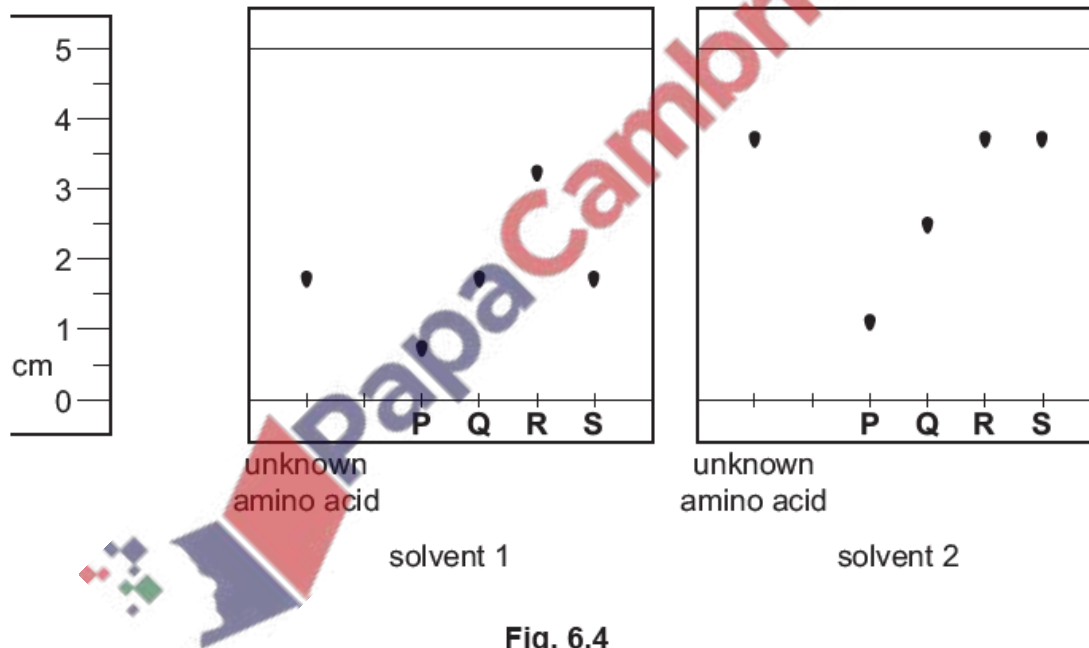
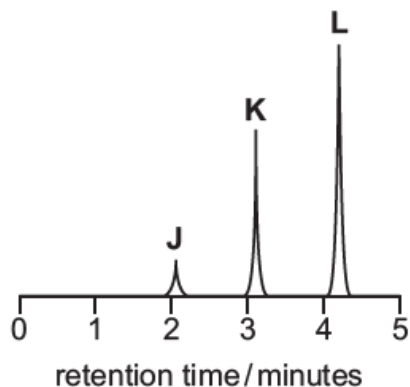


Fig. 6.4

Identify the unknown amino acid. Justify your answer.

.....  
.....  
..... [1]

(g) A mixture containing three organic compounds is analysed by gas chromatography and mass spectrometry. The gas chromatogram is shown.



peak	J	K	L
area/mm <sup>2</sup>	8	44	58

Fig. 6.5

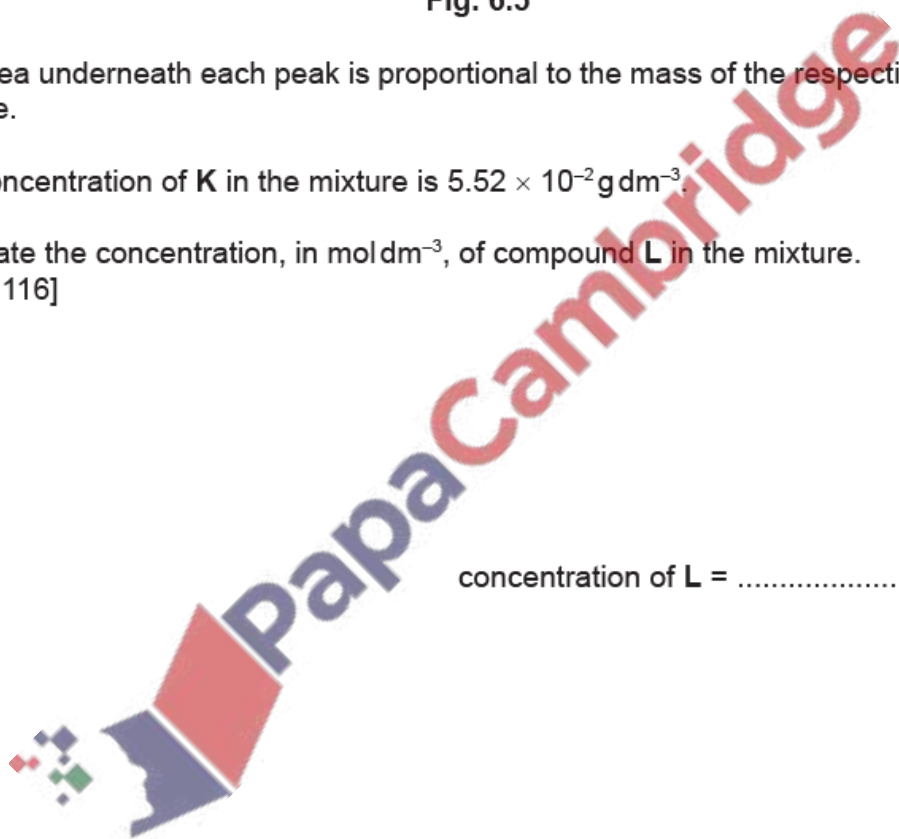
The area underneath each peak is proportional to the mass of the respective compound in the mixture.

The concentration of K in the mixture is  $5.52 \times 10^{-2} \text{ g dm}^{-3}$ .

Calculate the concentration, in  $\text{mol dm}^{-3}$ , of compound L in the mixture.

[ $M_r$ : L, 116]

concentration of L = .....  $\text{mol dm}^{-3}$  [1]



(a) State the uses of TMS and D<sub>2</sub>O in NMR spectroscopy.

TMS .....

D<sub>2</sub>O .....

[1]

(b) The three isomeric ketones with molecular formula C<sub>5</sub>H<sub>10</sub>O are:

- pentan-2-one
- pentan-3-one
- 3-methylbutanone.

(i) Complete Table 7.1 to show the number of peaks observed in the proton (<sup>1</sup>H) NMR spectrum and in the carbon-13 NMR spectrum for each compound listed.

Table 7.1

ketone	number of peaks observed in the proton ( <sup>1</sup> H) NMR spectrum	number of peaks observed in the carbon-13 NMR spectrum
pentan-2-one		
pentan-3-one		
3-methylbutanone		

[2]

(ii) State all the ketones with molecular formula C<sub>5</sub>H<sub>10</sub>O that have:

a doublet in their proton (<sup>1</sup>H) NMR spectrum

.....

a singlet in their proton (<sup>1</sup>H) NMR spectrum.

.....

[2]

(c) Cortisone,  $C_{21}H_{28}O_5$ , is a naturally occurring chemical that contains chiral carbon atoms.

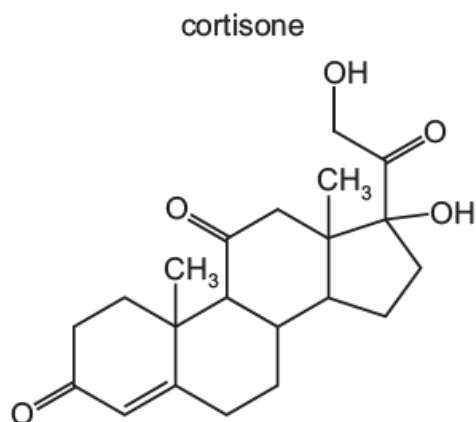


Fig. 7.1

(i) Deduce the number of chiral carbon atoms in one molecule of cortisone.

..... [1]

(ii) Cortisone is reacted with an excess of  $NaBH_4$ .

State the molecular formula of the organic compound formed.

..... [1]

(iii) Cortisone is an optically active molecule.

Explain what is meant by optically active.

.....  
.....  
..... [1]

[Total: 8]

