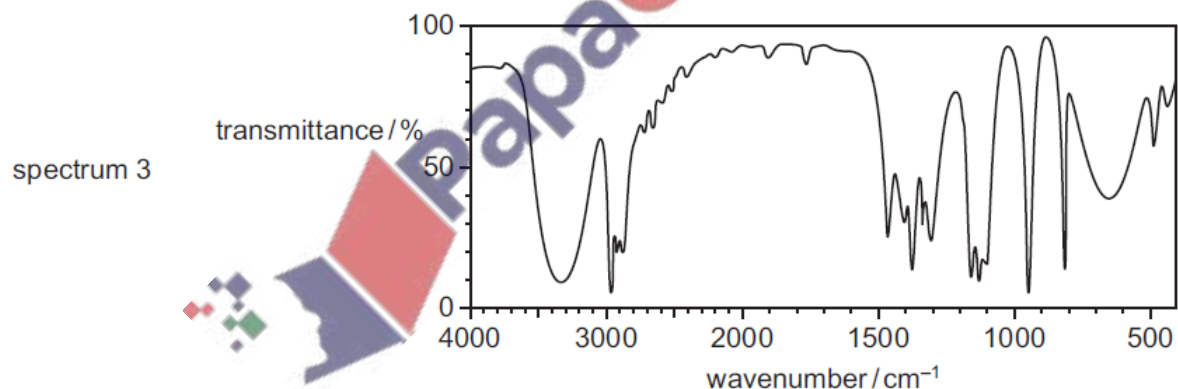
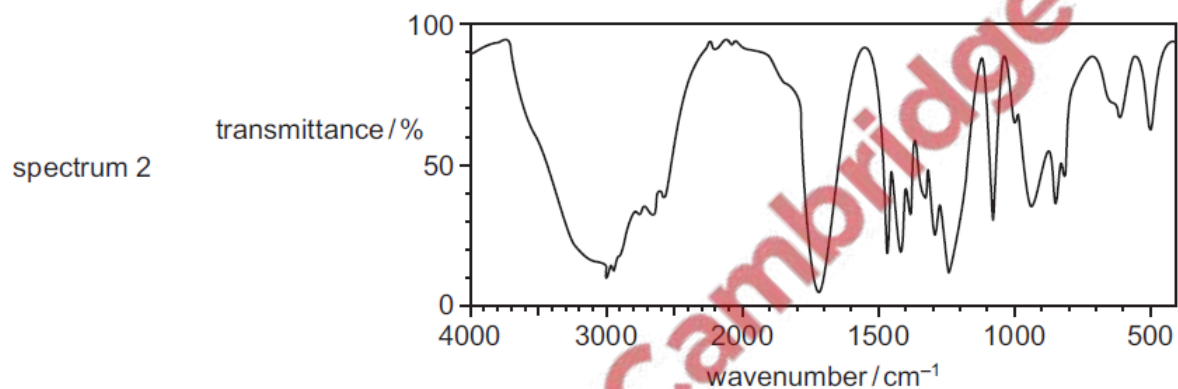
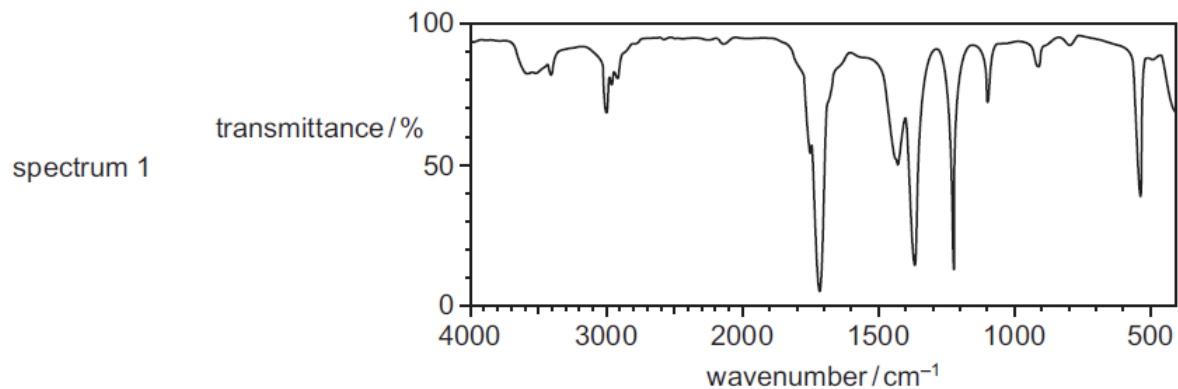


**1. Nov/2021/Paper\_11/No. 30**

The infra-red spectra of three organic compounds are shown.



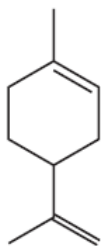
What could the three compounds be?

	spectrum 1	spectrum 2	spectrum 3
<b>A</b>	propanoic acid	propanone	propan-2-ol
<b>B</b>	propanone	propanoic acid	propan-2-ol
<b>C</b>	propanone	propan-2-ol	propanoic acid
<b>D</b>	propan-2-ol	propanoic acid	propanone

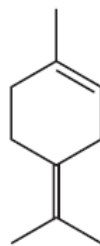
2. Nov/2021/Paper\_11/No. 37

A diketo acid is a compound with two ketone groups and one carboxylic acid group.

limonene



terpinolene



Which statements about the reactions of limonene and terpinolene are correct?

- 1 When reacted with an excess of hydrogen and a nickel catalyst, limonene and terpinolene produce the same compound.
- 2 An excess of hot concentrated acidified  $\text{KMnO}_4$  reacts with limonene and with terpinolene to form different diketo acids.
- 3 The reactions of limonene and terpinolene with an excess of  $\text{Br}_2$  produce positional isomers with the same number of chiral carbon atoms.

The responses **A** to **D** should be selected on the basis of

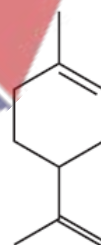
A	B	C	D
1, 2 and 3 are correct	1 and 2 only are correct	2 and 3 only are correct	1 only is correct

No other combination of statements is used as a correct response.

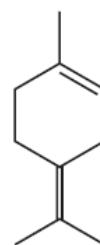
3. Nov/2021/Paper\_13/No. 37

A diketo acid is a compound with two ketone groups and one carboxylic acid group.

limonene



terpinolene



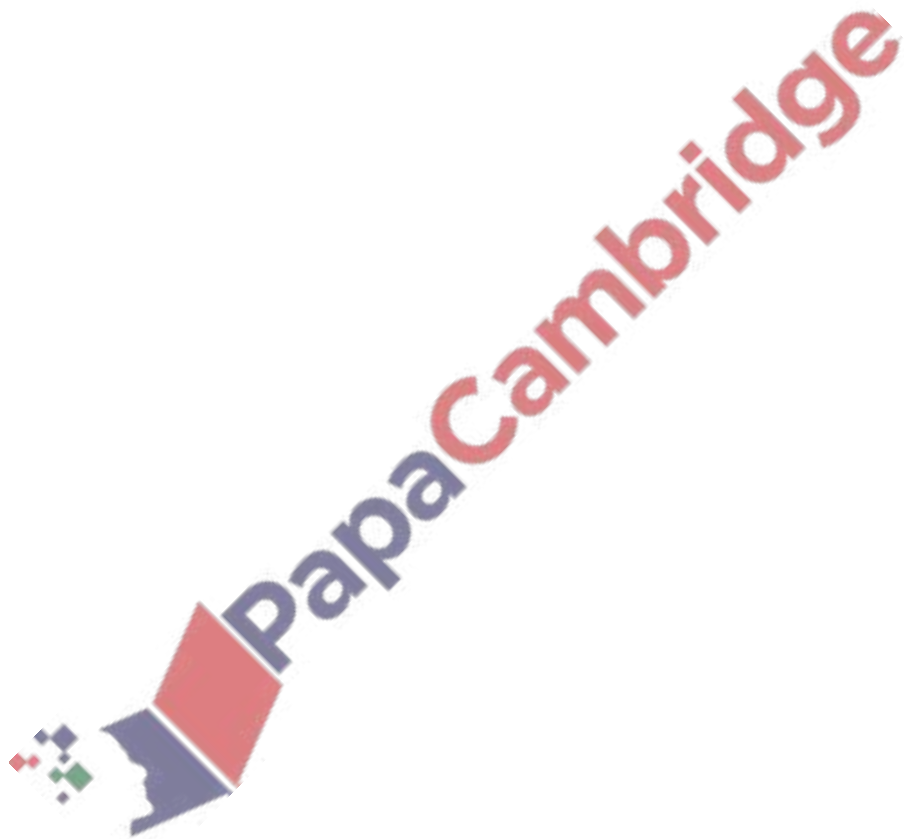
Which statements about the reactions of limonene and terpinolene are correct?

- 1 When reacted with an excess of hydrogen and a nickel catalyst, limonene and terpinolene produce the same compound.
- 2 An excess of hot concentrated acidified  $\text{KMnO}_4$  reacts with limonene and with terpinolene to form different diketo acids.
- 3 The reactions of limonene and terpinolene with an excess of  $\text{Br}_2$  produce positional isomers with the same number of chiral carbon atoms.

The responses **A** to **D** should be selected on the basis of

<b>A</b>	<b>B</b>	<b>C</b>	<b>D</b>
1, 2 and 3 are correct	1 and 2 only are correct	2 and 3 only are correct	1 only is correct

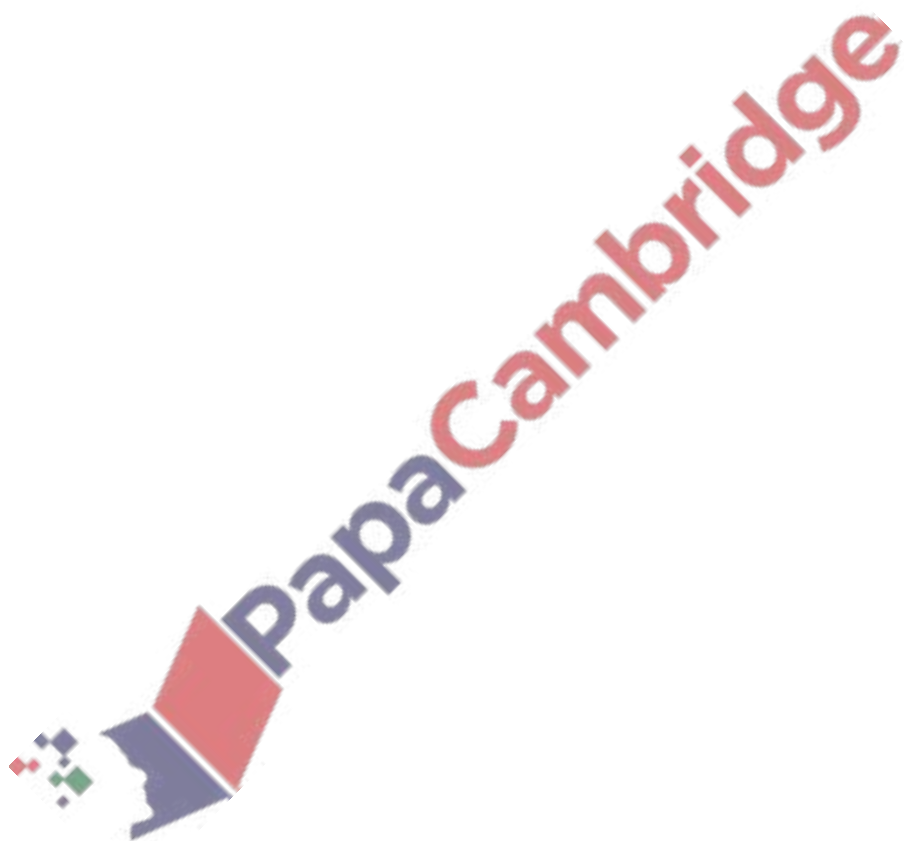
No other combination of statements is used as a correct response.



4. **March/2021/Paper\_12/No. 29**

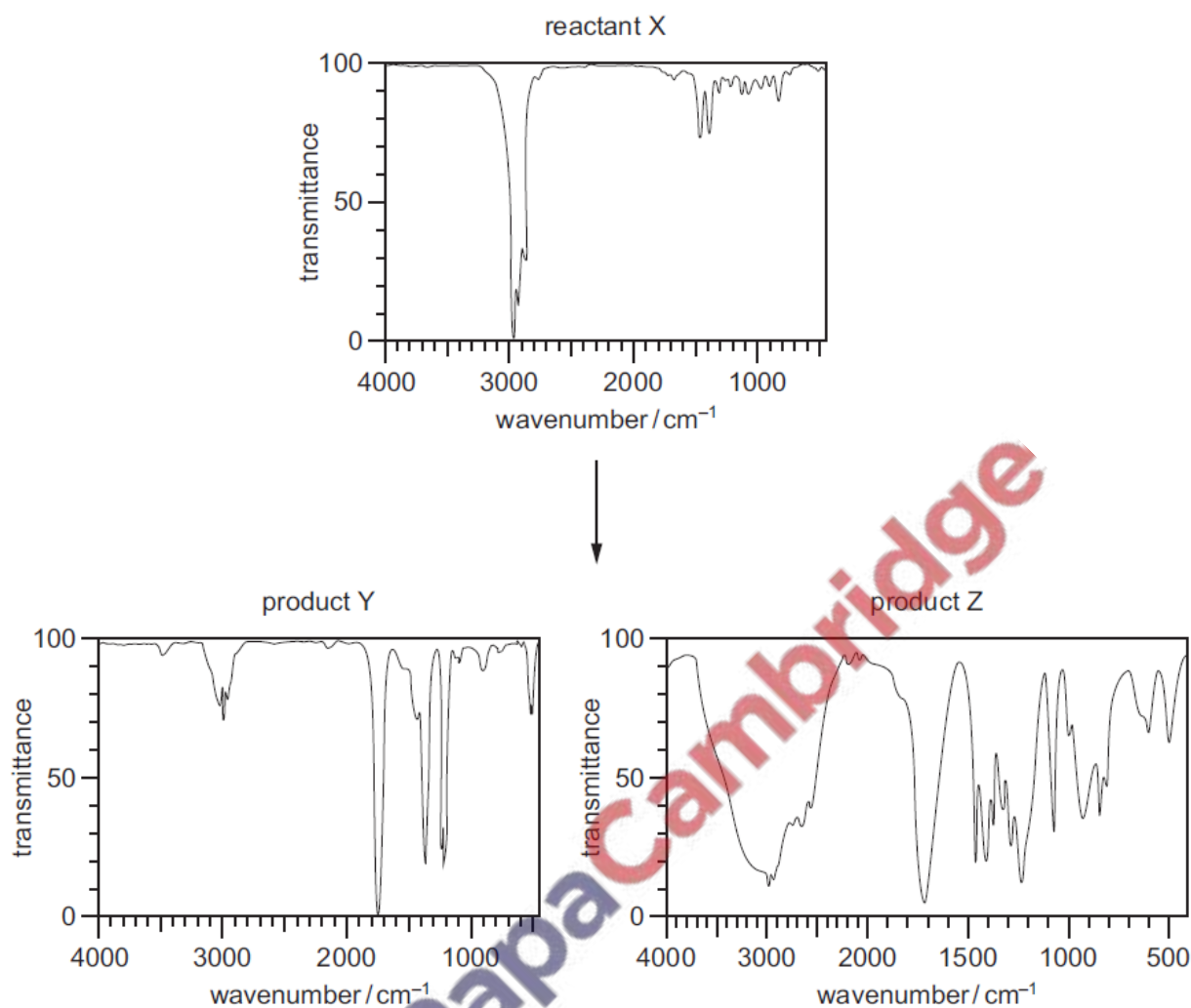
Which reaction gives butanoic acid as one of its products?

- A acid hydrolysis of butyl ethanoate
- B alkaline hydrolysis of butyl ethanoate
- C acid hydrolysis of ethyl butanoate
- D alkaline hydrolysis of ethyl butanoate



5. March/2021/Paper\_12/No. 30

When reactant X is treated with a suitable reagent, products Y and Z are formed. Infrared spectra of X, Y and Z are shown.



Which row could be correct?

	X	Y	Z
A	2,3-dimethylpent-2-ene	propanone	butanone
B	2-methylpent-2-ene	propanone	propanoic acid
C	pent-2-ene	ethanoic acid	propanoic acid
D	propyl propanoate	propan-1-ol	propanoic acid

6. **March/2021/Paper\_12/No. 40**

An organic compound, T, does **not** fizz when aqueous sodium carbonate is added to it.

Compound T contains 27.6% by mass of oxygen.

What could be the identity of T?

- 1 propanal
- 2 ethyl butanoate
- 3 3-methylpentanoic acid

7. **June/2021/Paper\_11/No.24**

The compound cetyl palmitate,  $C_{15}H_{31}CO_2C_{16}H_{33}$ , is a waxy solid.

Cetyl palmitate is heated under reflux with an excess of aqueous sodium hydroxide.

Which products will be formed?

- A  $C_{15}H_{31}ONa$  and  $C_{16}H_{33}CO_2Na$
- B  $C_{15}H_{31}CO_2Na$  and  $C_{16}H_{33}ONa$
- C  $C_{15}H_{31}OH$  and  $C_{16}H_{33}CO_2Na$
- D  $C_{15}H_{31}CO_2Na$  and  $C_{16}H_{33}OH$

8. **June/2021/Paper\_11/No.30**

Butanoic acid is prepared from 1-bromopropane.

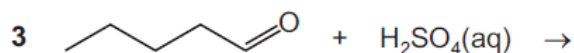
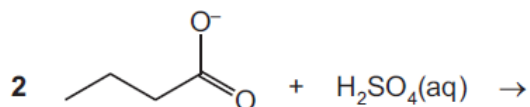
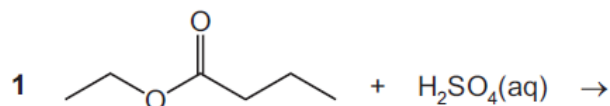
This synthesis requires a sequence of two reactions.

Which compound is prepared in the first stage of the synthesis?

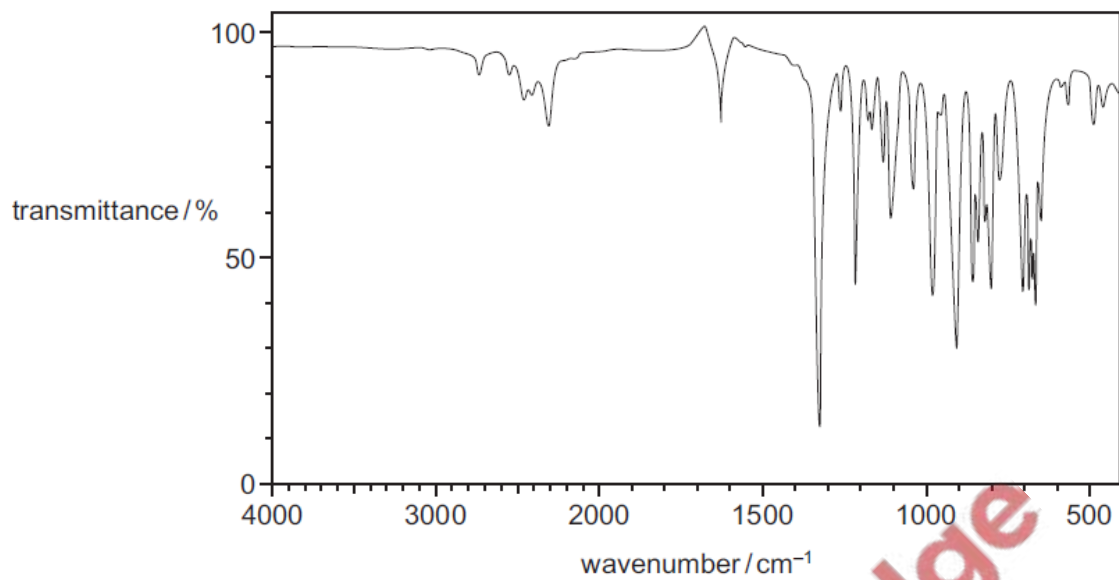
- A 1-aminopropane
- B propan-1-ol
- C butanal
- D butanenitrile

9. **June/2021/Paper\_12/No.40**

Which mixtures form a carboxylic acid as one of the products?

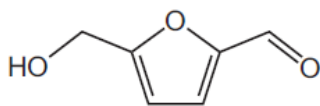


The infra-red spectrum of molecule Z is shown.

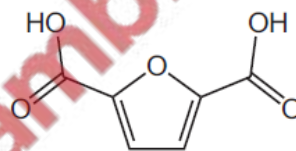


What could be the identity of Z?

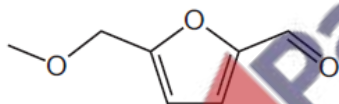
**A**



**B**



**C**



**D**

