## Carboxylic acids and derivatives – 2022 June AS Chemistry 9701

1. June/2022/Paper\_11/No.28

How many esters have the molecular formula C<sub>4</sub>H<sub>8</sub>O<sub>2</sub>?

- **A** 2
- **B** 3
- C 4
- **D** 5

### 2. June/2022/Paper\_11/No.37

Which compound produces butan-2-ol and ethanoic acid on hydrolysis?

- A CH<sub>3</sub>CO<sub>2</sub>CH(CH<sub>3</sub>)<sub>2</sub>
- B CH<sub>3</sub>CO<sub>2</sub>CH(CH<sub>3</sub>)CH<sub>2</sub>CH<sub>3</sub>
- C CH<sub>3</sub>CH(CH<sub>3</sub>)CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>
- D CH<sub>3</sub>CH<sub>2</sub>CO<sub>2</sub>CH(CH<sub>3</sub>)CH<sub>2</sub>CH<sub>3</sub>

### 3. June/2022/Paper\_11/No.38

Two 1g samples of Y are reacted separately and completely with sodium and with sodium carbonate. The volumes of the gases produced are collected and measured.

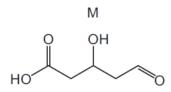
	relative volumes of gases	
	with Na	with Na <sub>2</sub> CO <sub>3</sub>
Υ	2	1

#### What could Y be?

- A CH<sub>3</sub>CH(OH)CH<sub>2</sub>OH
- B CH<sub>3</sub>CH(OH)CO<sub>2</sub>H
- C CH<sub>3</sub>COCH<sub>2</sub>OH
- D CH3COCO2H

# **4.** June/2022/Paper\_12/No.35

The skeletal formula of M is shown.



M is reacted with an excess of LiA1H4. Dilute acid is then added.

What is the molecular formula of the final organic product?

- A C<sub>5</sub>H<sub>6</sub>O<sub>5</sub>
- **B** C<sub>5</sub>H<sub>10</sub>O<sub>4</sub>
- ${f C}$   $C_5H_{10}O_3$
- $D C_5H_{12}O_3$

## **5.** June/2022/Paper\_12/No.38

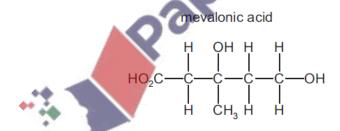
Ethyl butanoate is a flavouring, with a fruity flavour.

Which row is correct?

	alcohol and acid that react to form ethyl butanoate	the mass of water formed when 2.32 g of ester is formed
A	OH and OH	0.36 g
В	OH and OH	0.40 g
С	OH and OH	0.36 g
D	OH and OH	0.40 g

## **6.** June/2022/Paper\_13/No.28

The diagram shows the structural formula of mevalonic acid.



Which reagent and conditions will react with mevalonic acid to produce an organic compound without a chiral carbon atom?

- A heat under reflux with CH<sub>3</sub>OH/H<sup>+</sup>
- **B** heat under reflux with Cr<sub>2</sub>O<sub>7</sub><sup>2-</sup>/H<sup>+</sup>
- C Na at room temperature
- **D** PCl<sub>5</sub> at room temperature

**7.** June/2022/Paper\_13/No.37

An ester is shown.

Which two compounds react to form this ester?

- 2-methylpropan-1-ol and propanoic acid
- 2-methylpropan-2-ol and propanoic acid
- propan-1-ol and 2-methylpropanoic acid С
- D 2-methylpropan-2-ol and ethanoic acid

## **8.** June/2022/Paper\_13/No.38

Qacamido e Which compound can be used to make propanoic acid by treatment with a single reagent?

- CH<sub>2</sub>=CHCH<sub>2</sub>CH<sub>3</sub>
- CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CN
- C CH<sub>3</sub>CH(OH)CN
- D CH<sub>3</sub>CH(OH)CH<sub>3</sub>

(a)	Υd	But-2-ene reacts with KMnO <sub>4</sub> to form organic product, <b>Y</b> . <b>Y</b> does not react with Na <sub>2</sub> CO <sub>3</sub> . A gas is produced when an excess of Na is added to <b>Y</b> .		
	(i)	Describe the conditions for the KMnO <sub>4</sub> used in the reaction to form <b>Y</b> from but-2-ene.		
		[1]		
	(ii)	$24.0\text{cm}^3$ of gas is produced when an excess of Na is added to 0.001 mol of $\boldsymbol{Y},$ when measured under room conditions. Assume that 1 mol of gas occupies $24.0\text{dm}^3$ under room conditions.		
		Deduce a possible structure of <b>Y</b> . Explain your answer.		
		Pale Rale Rale Rale Rale Rale Rale Rale R		

**9.** June/2022/Paper\_23/No.5(a)