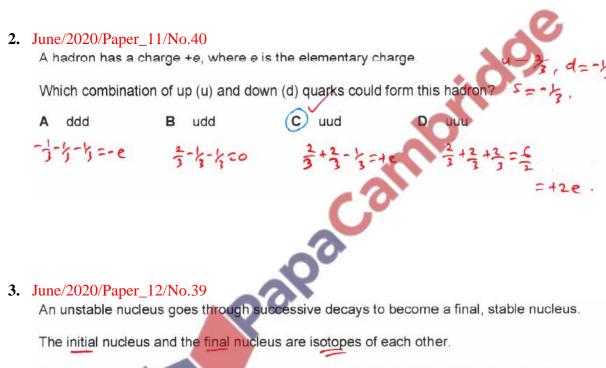
## Particle Physics – 2020 June AS Physics

#### 1. June/2020/Paper\_11/No.39

An element has two isotopic forms.

What are the nuclear arrangements of these two isotopes?

- A They have different nucleon numbers and different proton numbers. x
- **B**) They have different nucleon numbers but the same proton number.
- C They have the same nucleon number and the same proton number.
- D They have the same nucleon number but different proton numbers. 📈

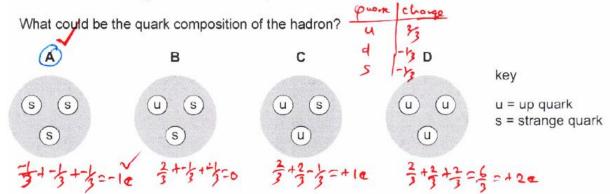


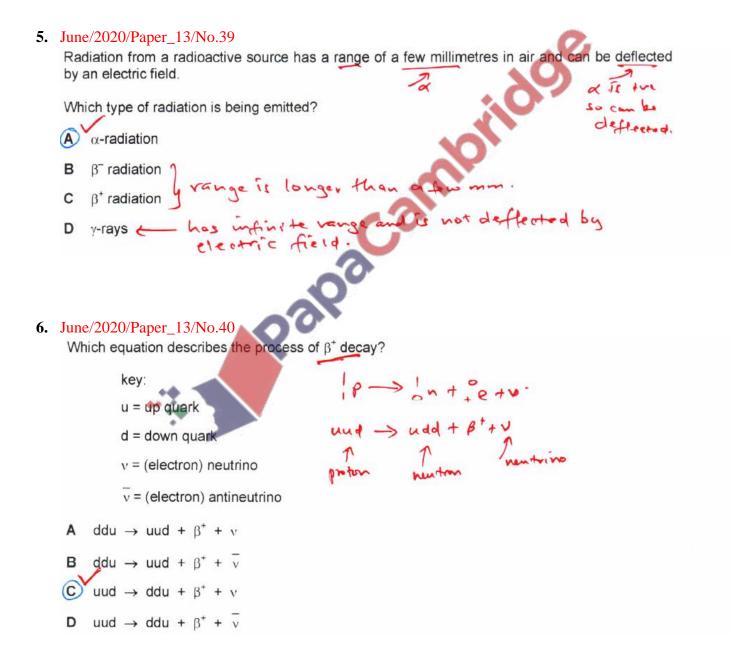
How many  $\alpha$  and  $\beta^{-}$  particles could have been emitted during the decay sequence?

	par	ticle	$^{\times}_{2}M \rightarrow ^{*-}_{N}N + ^{*}_{d}$	1 z is the proton
	α	β-		, number.
A	1	0	2-1 N - 2-1 P + - B	1 - Isotopos are atoms of the
B	1	2		1 same element
С	2	0	$2-1$ $P \rightarrow x-4M + B$	with different
D	2	1	2-1 (2-1×1) y x-y.	gun her of neutrons.

#### **4.** June/2020/Paper\_12/No.40

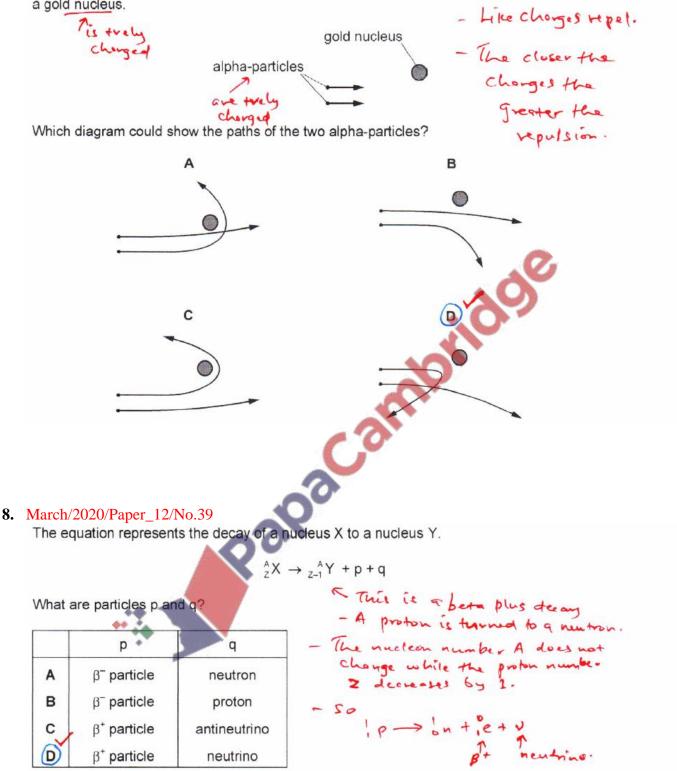
A hadron has a charge of -e and is composed of three quarks.





### 7. March/2020/Paper\_12/No.38

Two alpha-particles with the same kinetic energy are moving towards, and are then deflected by, a gold nucleus.



# 9. March/2020/Paper\_12/No.40

Which row gives the correct type and quark composition for the named particle?

	particle	type	quark composition	
A	neutron	hadron 🧹	u u d 🛪	key
в	neutron	lepton ≍	u d d 🛩	u = up quark
C	proton	hadron	u u d 🖌	d = down quark
D	proton	lepton 🗙	u d d 🗸	

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- Both neutron and proton are hadrons.
    Papacambridge
neutron - udd
proton - und.
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