

Additional Materials: ruler

## READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use a pencil for any diagrams or graphs.

Do not use staples, paper clips, highlighters, glue or correction fluid.

DO NOT WRITE IN ANY BARCODES.

Answer all questions.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [] at the end of each question or part question.

For Examiner's Use		
1		
2		
3		
Total		

This document consists of 9 printed pages and 3 blank pages.



1 Some students carried out tests for vitamin C.

They were provided with three vitamin C solutions, **S1**, **S2** and **S3**.

**S1** had a concentration of 0.2% vitamin C. **S2** had a concentration of 0.05% vitamin C. The concentration of **S3** was not known.

- The students measured 1 cm<sup>3</sup> of starch solution into a test-tube.
- They added 1 cm<sup>3</sup> of solution **S1**.
- The students added iodine solution, counting drop by drop, until a blue colour appeared. This was the end-point for solution **S1**.
- They repeated the test on solutions **S2** and **S3**.

These are the results that the students recorded.

HH HH HH III S1 sz ##1 HH HH I drops **S**3

(a) Record the students' observations in a suitable table using the space below.

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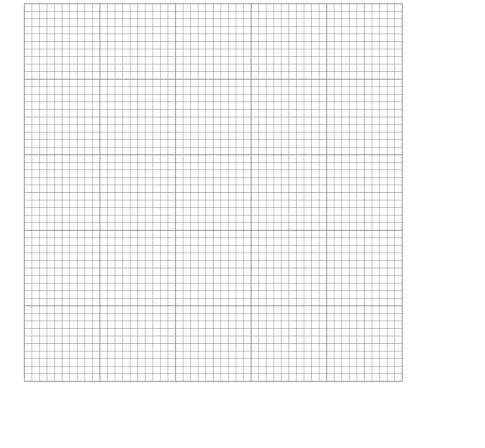
	42	
	3 Use these results to suggest the approximate vitamin C concentration of S3. Give reasons for your answer.	
(b)	Use these results to suggest the approximate vitamin C concentration of S3.	
	Give reasons for your answer.	hid
		1
	[3]	
(c)	Suggest <b>four</b> ways in which you could improve this method to find the concentration of an unknown vitamin C solution.	
	1	
	2.	
	3.	
	4.	
	[4]	
		1

www.papacambridge.com (d) Fig. 1.1 shows the results of a similar investigation into the concentration of vitual in five fruit juices. The students counted the number of drops of iodine solution use reach the end-point for each fruit juice.

Pineapple 5 Blackcurrant 48 Orange 16 Strawberry 22 Lemon 12



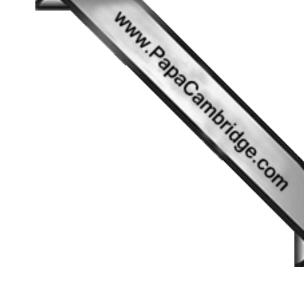
(i) On the grid below plot the data from Fig. 1.1 to show the variation in the number of drops of iodine solution required to reach the end-point.



(ii) State which fruit juice has the highest concentration of vitamin C.

[1] .....

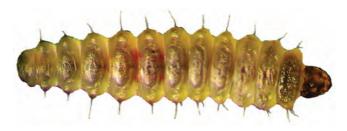
[Total: 17]



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Question 2 begins on Page 6

2 Fig. 2.1 shows a photograph of the larva of an insect.



6



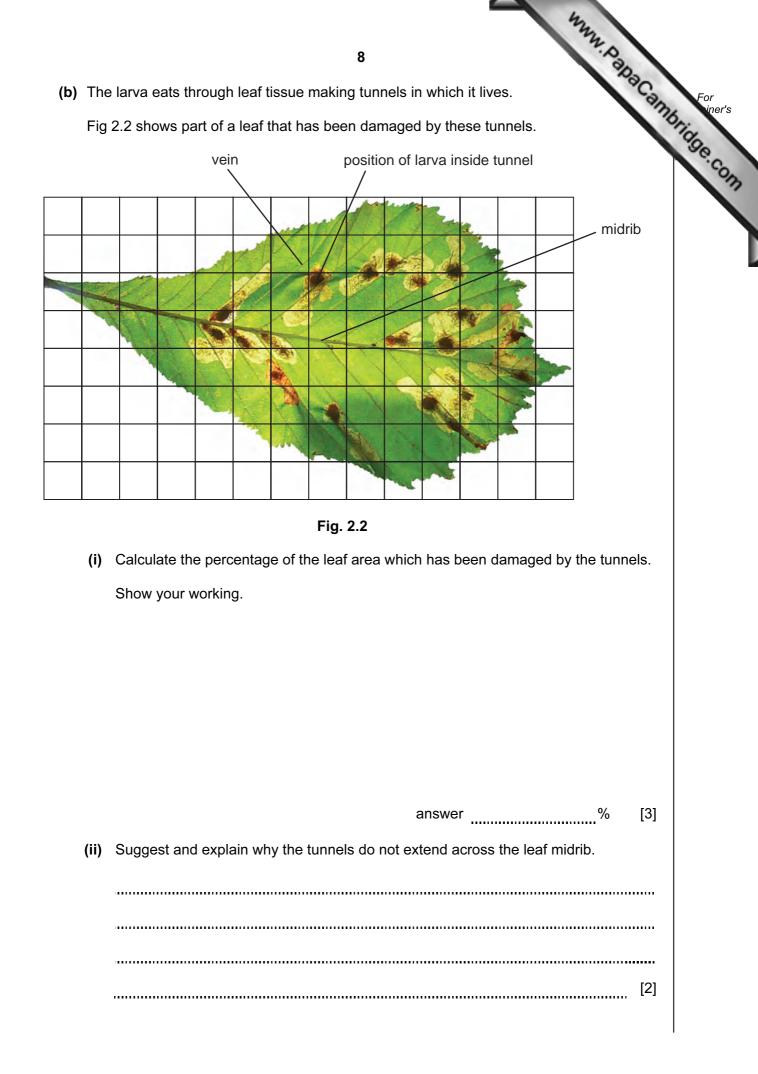
(a) (i) In the space below make a large drawing of the larva shown in Fig. 2.1.Labels are **not** needed.

[5]

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	true -	
	7	
(ii)	7 Measure the length of the larva in Fig. 2.1 and in your drawing.	For iner's
	length of larva in Fig. 2.1	iner's
	length of larva in your drawing [2]	Se.co.
(iii)	Calculate the magnification of your drawing compared with the larva in Fig. 2.1.	12
	Show your working.	

magnification [2]



- www.PapaCambridge.com (iii) Suggest two reasons why the leaf in Fig. 2.2 may die and fall off. 1. ..... ..... 2. ..... [2]
- (c) The larva in Fig. 2.1 becomes a moth.

Fig. 2.3 and Fig. 2.4 show the moth.

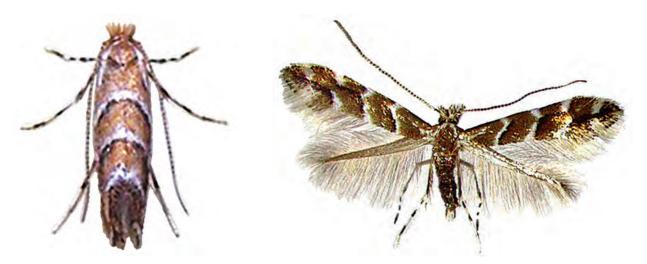




Fig. 2.4

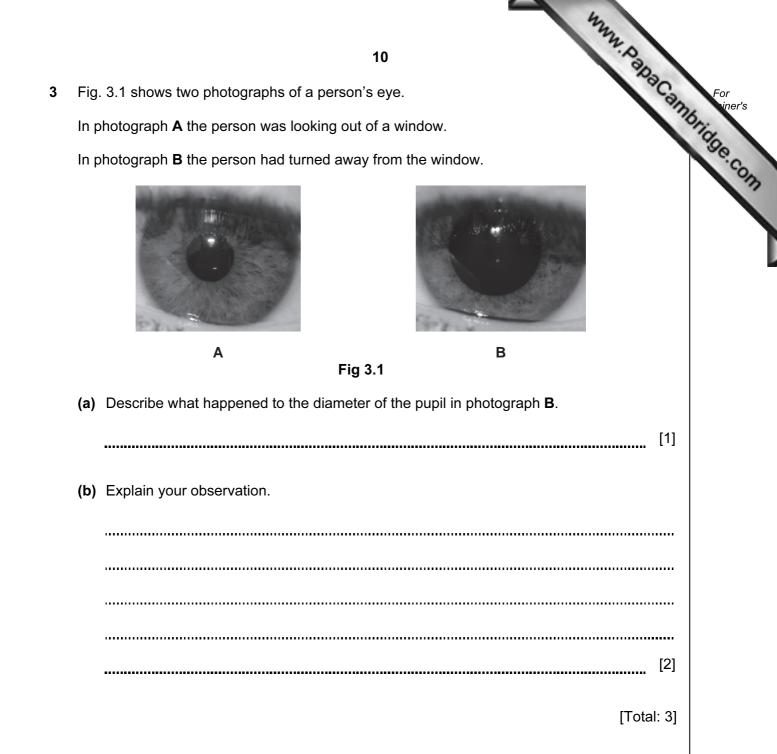
Look at Fig. 2.3 and Fig. 2.4.

(i) State **one** visible feature of this moth which is used to classify it as an arthropod.

......[1] (ii) State three visible features of this moth which are used to classify it as an insect. 1. .....

2. 3. [3] .....

[Total: 20]





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