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## **UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS**

International General Certificate of Secondary Education

## MARK SCHEME for the May/June 2006 question paper

## 0620 CHEMISTRY

0620/05

Paper 5, maximum raw mark 40

These mark schemes are published as an aid to teachers and students, to indicate the requirements of the examination. They show the basis on which Examiners were initially instructed to award marks. They do not indicate the details of the discussions that took place at an Examiners' meeting before marking began. Any substantial changes to the mark scheme that arose from these discussions will be recorded in the published *Report on the Examination*.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the *Report on the Examination*.

The minimum marks in these components needed for various grades were previously published with these mark schemes, but are now instead included in the Report on the Examination for this session.

• CIE will not enter into discussion or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the May/June 2006 question papers for most IGCSE and GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

Page 1	Mark Scheme	Syllabu
	IGCSE – May/June 2006	0620

1	Table	e of	resu	ılts
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1	Tab	able of results			
	Vol	umes completed for 10 seconds (1) and 20 seconds (1)	Mbridge		
	Vol	umes decreasing (1)			
	Cor	mparable to supervisor (2)	[5]		
	(a)	Graph points correctly plotted (2), -1 for each incorrect			
		Best fit lines (1) labels (1)	[4]		
	(b)	fizzing/bubbles/black (1)	[1]		
	(c)	(i) Experiment 1 (1)	[1]		
		(ii) strongest/more concentrated solution (1) more collisions (1)	[2]		
	(d)	(i) e.g. amount of catalyst added different/starting the timer/inaccurate measurement of peroxide max 2	[2]		
		<ul><li>(ii) e.g. measure mass of catalyst/time using data logger/use a burette or pipette to measure peroxide max 2</li></ul>	[2]		
	(e)	filter (1) mass of catalyst same before and after (1)			
		/repeat experiment and compare volumes of gas given off (1) max 2	[2]		
Exp	perim	nent 5			
Glo	wing	splint (1) relights (1) oxygen (1)	[3]		
			[Sub total = 22]		
2	(a)	white (1)	[1]		
	(b)	paper goes pink/colourless/condensation description/ solid yellow max 2	[2]		
	(c)	(i) white (1) precipitate (1) soluble in excess or similar (1)	[3]		
		(ii) white precipitate (1) soluble in excess or similar(1)	[2]		
		(iii) white precipitate (1)	[1]		
		(iv) no reaction/unchanged (1)	[1]		
	(d)	fizz/bubbles (1) limewater milky (1)	[2]		
	(e)	water (of crystallisation) present (1)	[1]		
	(f)	sulphate present (1) chloride/halide absent (1)	[2]		
	(g)	<b>B</b> is zinc (1) sulphate (1)			
		C is (zinc) carbonate (1)	[3]		
			[Sub total = 18]		