CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

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0620 CHEMISTRY

0620/51

Paper 5 (Practical), maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2012 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.

				Syllabus 0620 eted (1) han 20 °C below original (1)	
	Page 2		Mark Scheme	Syllabus 7.0 r	
1	(9)	Table of	IGCSE – October/November 2012	0620 Parca	
I	(e)	Table of results for Experiments all initial temperature boxes completed correctly as instructed (1)			
			emperature boxes completed correctly as instruct	hen 20°C bolow original (1)	0
			ge temperatures completed correctly (1)		m
			mpleted in seconds (1) ignore: dps		
			ing in order (comparable to supervisor) (1)	[5]	
		<u></u>	<u></u>		
	(f)		otted correctly (4)		
		smooth li	ine graph (1)	[5]	
	(g)	average	temperature 72 °C (1)		
		value fro	m graph (1)		
		extrapola	ation shown on grid (1)	[3]	
	(h)	as an inc	dicator/check presence of iodine owtte (1)	[1]	
	(i)	., .	eriment 5/when temperature is 70 (1)	[1]	
			est temperature (1)	[0]	
		paru	icles have more energy/more collisions (1)	[2]	
	(j)	time lonç	ger/more/increase (1)		
		speed slo	ower/decrease (1)	[2]	
	(k)	more <u>ac</u>	<u>curate (</u> 1)	[1]	
~	(-)	··	····	[4]	
2	(a)	рН 5— <i>і</i> (1) ignore colours	[1]	
	(b)	(i) white	e (1) precipitate (1) dissolves owtte (1)	[3]	
		(ii) white	e (1) precipitate (1) dissolves owttte (1)	[3]	
	(c)	no reacti	ion/no change/no precipitate/no observation (1)	[1]	
	1-,	10			
	(d)	white (1)	precipitate (1)	[2]	

