

Cambridge Assessment International Education

Cambridge International General Certificate of Secondary Education

CHEMISTRY 0620/53

Paper 5 Practical Test

October/November 2019

CONFIDENTIAL INSTRUCTIONS



This document gives details of how to prepare for and administer the practical exam.

The information in this document and the identity of any materials supplied by Cambridge International are confidential and must NOT reach candidates either directly or indirectly.

The supervisor must complete the report at the end of this document and return it with the scripts.

If you have any queries regarding these confidential instructions, contact Cambridge International stating the centre number, the syllabus and component number and the nature of the query.

email info@cambridgeinternational.org

phone +44 1223 553554 fax +44 1223 553558

This syllabus is regulated for use in England, Wales and Northern Ireland as a Cambridge International Level 1/Level 2 Certificate.

This document consists of 6 printed pages and 2 blank pages.



General information about practical exams

Centres must follow the guidance on science practical exams given in the Cambridge Handbook.

Safety

Supervisors must follow national and local regulations relating to safety and first aid.

Only those procedures described in the question paper should be attempted.

Supervisors must inform candidates that materials and apparatus used in the exam should be treated with caution. Suitable eye protection should be used where necessary.

The following hazard codes are used in these confidential instructions, where relevant:

C corrosive
 HH health hazard
 F flammable
 MH moderate hazard
 T acutely toxic
 O oxidising

N hazardous to the aquatic environment

Hazard data sheets relating to substances used in this exam should be available from your chemical supplier.

Before the exam

- The packets containing the question papers must not be opened before the exam.
- It is assumed that standard school laboratory facilities, as indicated in the *Guide to Planning Practical Science*, will be available.
- Spare materials and apparatus for the tasks set must be available for candidates, if required.

During the exam

- It must be made clear to candidates at the start of the exam that they may request spare materials and apparatus for the tasks set.
- Where specified, the supervisor must perform the experiments and record the results as instructed.
 This must be done out of sight of the candidates, using the same materials and apparatus as the candidates.
- Any assistance provided to candidates must be recorded in the supervisor's report.
- If any materials or apparatus need to be replaced, for example, in the event of breakage or loss, this must be recorded in the supervisor's report.

After the exam

- The supervisor must complete a report for each practical session held and each laboratory used.
- Each packet of scripts returned to Cambridge International must contain the following items:
 - the scripts of the candidates specified on the barcode label provided
 - the supervisor's results relevant to these candidates
 - the supervisor's reports relevant to these candidates
 - seating plans for each practical session, referring to each candidate by candidate number
 - the attendance register.

Specific information for this practical exam

During the exam, the supervisor (NOT the invigilator) must do the experiments in Questions 1 and 2 and record the results on a spare copy of the question paper, clearly labelled 'supervisor's results'.

Materials and apparatus for Question 1

Each candidate will require the following materials and apparatus. Labels do **not** need to include concentrations.

- 100 cm³ of dilute hydrochloric acid of concentration 0.20 mol/dm³, labelled dilute hydrochloric acid for Question 1
- [MH] 50 cm³ of aqueous sodium hydroxide of concentration 0.10 mol/dm³, labelled solution R
- [MH] 50 cm³ of aqueous sodium hydroxide of concentration 0.20 mol/dm³, labelled solution S
 - 50 cm³ of aqueous sodium hydroxide of concentration 0.050 mol/dm³, labelled solution T
 - access to water and distilled water

[C][F][HH] • [MH][N][T]

- methyl orange indicator
- 25 cm³ or 50 cm³ measuring cylinder
- 250 cm³ conical flask
- 50 cm³ burette with stand and clamp
- white tile
- funnel for filling burette
- teat pipette

During the exam, the supervisor (NOT the invigilator) must do the experiments in Questions 1 and 2 and record the results on a spare copy of the question paper, clearly labelled 'supervisor's results'.

Materials and apparatus for Question 2

Each candidate will require the following materials and apparatus. Labels do **not** need to include concentrations.

- [MH] stoppered test-tube containing about 0.5 g of chromium(III) potassium sulfate-12-water (chrome alum), CrK(SO₄)₂•12H₂O, labelled solid U
- [F][MH] stoppered test-tube containing about 2 cm³ of hexane, labelled liquid V
- [MH][N] a small crystal or small pellet of iodine in a stoppered test-tube
 - aqueous barium nitrate of suitable concentration to give a positive sulfate test
 - [C] dilute nitric acid of concentration 1.0 mol/dm³
 - [C] aqueous sodium hydroxide of concentration 1.0 mol/dm³
- [MH][N] aqueous ammonia of concentration 1.0 mol/dm³
 - distilled water
 - 10 cm³ measuring cylinder
 - splints
 - apparatus for a flame test
 - Bunsen burner and means to light it
 - rack of test-tubes
 - boiling tube
 - stopper to fit boiling tube
 - watch-glass
 - teat pipettes
 - spatula

BLANK PAGE

BLANK PAGE

Permission to reproduce items where third-party owned material protected by copyright is included has been sought and cleared where possible. Every reasonable effort has been made by the publisher (UCLES) to trace copyright holders, but if any items requiring clearance have unwittingly been included, the publisher will be pleased to make amends at the earliest possible opportunity.

To avoid the issue of disclosure of answer-related information to candidates, all copyright acknowledgements are reproduced online in the Cambridge Assessment International Education Copyright Acknowledgements Booklet. This is produced for each series of examinations and is freely available to download at www.cambridgeinternational.org after the live examination series.

Cambridge Assessment International Education is part of the Cambridge Assessment Group. Cambridge Assessment is the brand name of the University of Cambridge Local Examinations Syndicate (UCLES), which itself is a department of the University of Cambridge.

Supervisor's report

Syllabus and component number			1		
Centre number					
Centre name	 	 		 	
Time of the practical session	 	 		 	
Laboratory name/number					
Laboratory Harrie/Hulliber	 	 		 	

Give details of any difficulties experienced by the centre or by candidates (include the relevant candidate names and candidate numbers).

You must include:

- any difficulties experienced by the centre in the preparation of materials
- any difficulties experienced by candidates, e.g. due to faulty materials or apparatus
- any specific assistance given to candidates.

D -	-1-	4	

De	claration
1	Each packet that I am returning to Cambridge International contains the following items:
	the scripts of the candidates specified on the barcode label provided
	the supervisor's results relevant to these candidates
	the supervisor's reports relevant to these candidates
	seating plans for each practical session, referring to each candidate by candidate number
	the attendance register
2	Where the practical exam has taken place in more than one practical session, I have clearly labelled the supervisor's results, supervisor's reports and seating plans with the time and laboratory name/number for each practical session.
3	I have included details of difficulties relating to each practical session experienced by the centre or by candidates.
4	I have reported any other adverse circumstances affecting candidates, e.g. illness, bereavement or temporary injury, directly to Cambridge International on a <i>special consideration form</i> .
Sig	ned (supervisor)
Naı	me (in block capitals)

© UCLES 2019 0620/53/CI/O/N/19