## The Particulate nature of matter – 2023 June IGCSE Chemistry 0620

### 1. June/2023/Paper 0620/11/No.1

The diagram shows the result of dropping a purple crystal into water.



Which processes take place in this experiment?

	chemical reaction	diffusing	dissolving			
Α	✓	✓	x			
В	✓	X	X			
С	X	X	✓			
D	x	✓	✓			
	23/Paper_ 062  bhysical chang  1 conden	es of ethanol a	are listed.			
	3 freezing	g	00			
	4 boiling					

### 2. June/2023/Paper\_ 0620/12/No.1

- condensation
- evaporation
- 3 freezing
- 4 boiling

In which changes do the particles move further apart?

- A 1 and 2
- 1 and 3
- 2 and 4
- D 3 and 4

#### 3. June/2023/Paper\_ 0620/13/No.1

Nitrogen is heated in a balloon, which expands slightly.

Which statements about the molecules of nitrogen are correct?

- 1 They move further apart.
- 2 They move more quickly.
- 3 They remain the same distance apart.
- 4 Their speed remains unchanged.
- A 1 and 2
- 1 and 4
- C 2 and 3
- **D** 3 and 4

### 4. June/2023/Paper\_ 0620/21/No.1

The diagram shows the result of dropping a purple crystal into water.



Which processes take place in this experiment?

	chemical reaction	diffusing	dissolving		
Α	✓	✓	x		
В	✓	x	x		
С	x	x	✓		
D	×	✓	✓		5
	023/Paper_ 062 physical chang		are listed.	2/0	
	1 conder	nsation			
	2 evapor	ration	.0		
	3 freezin	ıg			
	4 boiling		0		

### 5. June/2023/Paper 0620/22/No.1

- condensation
- 2 evaporation
- freezing
- boiling

In which changes do the particles move further apart?

- A 1 and 2
- 2 and 4
- 3 and 4

# 6. June/2023/Paper\_0620/23/No.1

Nitrogen is heated in a balloon, which expands slightly.

Which statements about the molecules of nitrogen are correct?

- 1 They move further apart.
- 2 They move more quickly.
- 3 They remain the same distance apart.
- Their speed remains unchanged.
- 1 and 2
- 1 and 4
- 2 and 3
- 3 and 4

#### **7.** June/2023/Paper\_ 0620/23/No.8

Which row describes the structure of a solid metal and explains the property?

	structure of solid metal	property of solid metal
Α	lattice of negative ions in a sea of electrons	conducts electricity because the electrons are free to move
В	lattice of negative ions in a sea of electrons	is malleable because the layers of ions can slide over each other
С	lattice of positive ions in a sea of electrons	conducts electricity because the ions are free to move
D	lattice of positive ions in a sea of electrons	is malleable because the layers of ions can slide over each other

### 8. March/2023/Paper\_ 0620/12/No.1

The arrangements of particles in solids, liquids and gases are different.

Which statement about the molecules in ice, water or steam is correct?

- A The H<sub>2</sub>O molecules are on average closest together in steam
- B The H<sub>2</sub>O molecules are on average furthest apart in water.
- C The H<sub>2</sub>O molecules in steam have the second highest average velocity.
- **D** The H<sub>2</sub>O molecules in ice are able to vibrate.

#### 9. March/2023/Paper 0620/12/No.2

The melting points and boiling points of three elements, at 1 atm pressure, are shown.

	K	melting point /°C	boiling point /°C
4	argon	-189	-186
1	nitrogen	-210	-196
्री	oxygen	-218	-183

Separate samples of argon, nitrogen and oxygen are stored at -200 °C and at 1 atm pressure.

How many samples are liquids?

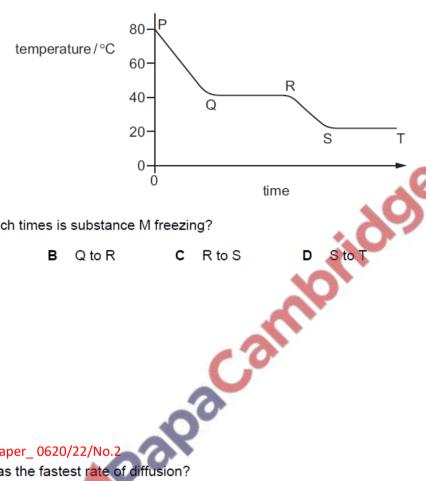
**A** 0 **B** 1 **C** 2 **D** 3

10. March/2023/Paper\_ 0620/22/No.1

Substance M is a solid at 30 °C.

The substance is heated to 80 °C and its temperature measured as it cools down to room temperature.

The cooling curve is shown.



Between which times is substance M freezing?

- A PtoQ
- B Q to R

# 11. March/2023/Paper\_ 0620/22/No.2

Which gas has the fastest rate of diffusion?

- A Ar
- $C_2H_6$
- С HC1
- D H<sub>2</sub>S