## Measurement

## Question Paper 1

| Level | IGCSE |
| :--- | :--- |
| Subject | Chemistry (0620/0971) |
| Exam Board | Cambridge International Examinations (CIE) |
| Topic | Experimental techniques |
| Sub-Topic | Measurement |
| Booklet | Question Paper 1 |

## Time Allowed:

27 minutes
Score: /22
Percentage: /100

## Grade Boundaries:

| 9 | 8 | 7 | 6 | 5 | 4 | 3 | 2 | 1 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $>85 \%$ | $75 \%$ | $68 \%$ | $60 \%$ | $53 \%$ | $48 \%$ | $40 \%$ | $33 \%$ | $<25 \%$ |

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1 A student takes 2 g samples of calcium carbonate and adds them to $20 \mathrm{~cm}^{3}$ samples of dilute hydrochloric acid at different temperatures. She measures how long it takes for the effervescence to stop.

Which apparatus does she use?

|  | balance | clock | filter <br> funnel | measuring <br> cylinder | thermometer |
| :---: | :---: | :---: | :---: | :---: | :---: |
| A | $\checkmark$ | $\checkmark$ | $\checkmark$ | $\checkmark$ | $x$ |
| B | $\checkmark$ | $\checkmark$ | $x$ | $\checkmark$ | $\checkmark$ |
| C | $\checkmark$ | $x$ | $\checkmark$ | $\checkmark$ | $\checkmark$ |
| D | $x$ | $\checkmark$ | $\checkmark$ | $x$ | $\checkmark$ |

2 A student carries out an experiment to find how fast 3 cm pieces of magnesium ribbon dissolve in $10 \mathrm{~cm}^{3}$ samples of sulfuric acid at different temperatures.

Which piece of apparatus does the student not need?
A balance
B measuring cylinder
C stop-clock
D thermometer

3 A student was provided with only a thermometer, a stopwatch and a beaker.
What could the student measure?
A 10.5 g solid and $24.8 \mathrm{~cm}^{3}$ liquid
B $\quad 10.5 \mathrm{~g}$ solid and $25^{\circ} \mathrm{C}$
C $24.8 \mathrm{~cm}^{3}$ liquid and 45 seconds
D $25^{\circ} \mathrm{C}$ and 45 seconds

4 Part of the instructions in an experiment reads as follows.
Quickly add $50 \mathrm{~cm}^{3}$ of acid.
What is the best piece of apparatus to use?
A a burette
B a conical flask
C a measuring cylinder
D a pipette

5 A student investigates how the concentration of an acid affects the speed of reaction with a 0.5 g mass of magnesium at $30^{\circ} \mathrm{C}$.

The student has a beaker, concentrated acid, water and the apparatus below.
P a balance
Q a clock
R a measuring cylinder
S a thermometer
Which pieces of apparatus does the student use?
A P, Q and R only
B P, Q and S only
C $Q, R$ and $S$ only
D P, Q, R and S

6 A student measures the rate of two reactions.
In one reaction, there is a change in mass of the reactants during the reaction.
In the second reaction, there is a change in temperature during the reaction.
Which piece of apparatus would be essential in both experiments?
A balance
B clock
C pipette
D thermometer

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7 Crystals of sodium chloride were prepared by the following method.
$125.0 \mathrm{~cm}^{3}$ of dilute hydrochloric acid was accurately measured into a conical flask.
2 Aqueous sodium hydroxide was added until the solution was neutral. The volume of sodium hydroxide added was measured.

3 The solution was evaporated and the crystals washed with approximately $15 \mathrm{~cm}^{3}$ of water.

Which row shows the pieces of apparatus used to measure the $25.0 \mathrm{~cm}^{3}$ of hydrochloric acid, the volume of aqueous sodium hydroxide and the $15 \mathrm{~cm}^{3}$ of water?

|  | $25.0 \mathrm{~cm}^{3}$ of hydrochloric <br> acid accurately | the volume of aqueous <br> sodium hydroxide added | $15 \mathrm{~cm}^{3}$ of water <br> approximately |
| :---: | :---: | :---: | :---: |
| A | burette | pipette | measuring cylinder |
| B | measuring cylinder | burette | pipette |
| C | pipette | burette | measuring cylinder |
| D | pipette | measuring cylinder | burette |

8 Lead iodide is insoluble in water.
Lead iodide is made by adding aqueous lead nitrate to aqueous potassium iodide.
Which pieces of apparatus are needed to obtain solid lead iodide from $20 \mathrm{~cm}^{3}$ of aqueous lead nitrate?


1


2


4
A
1, 2and 4
B 1, 3 and 5
C 1, 4 and 5
D 2, 4 and 5

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9
The diagram shows an experiment to find the formula of magnesium oxide.


Which piece of apparatus would be needed in addition to those shown?
A a balance
B a measuring cylinder
C a spatula
D a thermometer

10 The diagram shows three pieces of apparatus that are used for measuring the volume of a liquid.


What are these pieces of apparatus?

|  | 1 | 2 | 3 |
| :---: | :---: | :---: | :---: |
| A | burette | measuring cylinder | pipette |
| B | burette | pipette | measuring cylinder |
| C | measuring cylinder | burette | pipette |
| D | measuring cylinder | pipette | burette |

11 The four pieces of apparatus shown below are used in chemical experiments.

burette
 measuring
cylinder

pipette

thermometer

Which statement about the apparatus is correct?
A The burette measures the volume of liquid added in a titration.
B The measuring cylinder measures the mass of a substance used in an experiment.
C The pipette measures the volume of gas given off in a reaction.
D The thermometer measures the density of a solution.
$12 \mathrm{P}, \mathrm{Q}, \mathrm{R}$ and S are pieces of apparatus.


Which row describes the correct apparatus for the measurement made?

|  | apparatus | measurement made |
| :---: | :---: | :---: |
| A | P | the volume of acid added to alkali in a titration |
| B | Q | $\mathrm{cm}^{3}$ of acid to add to calcium carbonate in a rate-determining experiment |
| C | R | $7 \mathrm{~cm}^{3}$ of a gas given off in a rate-determining experiment |
| D | S | $2 \mathrm{~cm}^{3}$ of alkali for use in a titration |

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13 A student uses the apparatus shown in the diagram below to measure the volume of carbon dioxide gas made when different masses of marble chips are added to $25 \mathrm{~cm}^{3}$ of dilute hydrochloric acid.


Which other items of apparatus are needed?
A funnel and balance
B funnel and stopwatch
C measuring cylinder and balance
D measuring cylinder and stopwatch

14 During a reaction, the following changes take place.
1 The temperature rises.
2 A gas is given off.
Which apparatus is required to measure the rate of this reaction?
A balance and burette
B balance and gas syringe
C gas syringe and burette
D gas syringe and stopclock

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15 Which piece of apparatus is used to measure variable quantities of liquid in a titration?


16 A student put $25.0 \mathrm{~cm}^{3}$ of dilute hydrochloric acid into a conical flask.
The student added 2.5 g of solid sodium carbonate and measured the change in temperature of the mixture.

Which apparatus does the student need to use to obtain the most accurate results?
A balance, measuring cylinder, thermometer
B balance, pipette, stopwatch
C balance, pipette, thermometer
D burette, pipette, thermometer

17 During an experiment a measurement is recorded in $\mathrm{cm}^{3}$.
Which apparatus is used?
A balance
B measuring cylinder
C stopclock
D thermometer

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$25 \mathrm{~cm}^{3}$ of an alkali are added to $20 \mathrm{~cm}^{3}$ of an acid. The temperature change is measured. Which apparatus is not needed in the experiment?

A $25 \mathrm{~cm}^{3}$ measuring cylinder
B $\quad 100 \mathrm{~cm}^{3}$ beaker
C balance
D thermometer

19 The diagram shows part of a thermometer.


What is the reading on the thermometer?
A 30.2
B 30.3
C 31.7
D 31.8

20 A student needs to measure four different volumes of a solution accurately. The volumes are $10 \mathrm{~cm}^{3}, 25 \mathrm{~cm}^{3}, 50 \mathrm{~cm}^{3}$ and $60 \mathrm{~cm}^{3}$.

The apparatus available includes a $25 \mathrm{~cm}^{3}$ pipette.
Which volumes could be measured using this pipette?
A $10 \mathrm{~cm}^{3}$ and $25 \mathrm{~cm}^{3}$
B $25 \mathrm{~cm}^{3}$ and $50 \mathrm{~cm}^{3}$
C $25 \mathrm{~cm}^{3}$ only
D $50 \mathrm{~cm}^{3}$ and $60 \mathrm{~cm}^{3}$

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21 A student needs to measure $22 \mathrm{~cm}^{3}$ of water at $40^{\circ} \mathrm{C}$.
Which apparatus is required?
A beaker and stopwatch
B beaker and thermometer
C measuring cylinder and stopwatch
D measuring cylinder and thermometer

22 The diagrams show liquids in a burette and a measuring cylinder.

burette

measuring cylinder

Which row shows the correct readings for the burette and the measuring cylinder?

|  | burette | measuring <br> cylinder |
| :---: | :---: | :---: |
| A | 27.8 | 42 |
| B | 27.8 | 44 |
| C | 28.2 | 42 |
| D | 28.2 | 44 |

