CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education MARK SCHEME for the November 2003 question papers 0653 COMBINED SCIENCE

0653/01	Paper 1 (Multiple Choice), maximum raw mark 40
0653/02	Paper 2 (Core), maximum raw mark 60
0653/03	Paper 3 (Extended), maximum raw mark 80
0653/05	Paper 5 (Practical), maximum raw mark 30
0653/06	Paper 6 (Alternative to Practical), maximum raw mark 60

These mark schemes are published as an aid to teachers and students, to indicate the requirements of the examination. They show the basis on which Examiners were initially instructed to award marks. They do not indicate the details of the discussions that took place at an Examiners' meeting before marking began. Any substantial changes to the mark scheme that arose from these discussions will be recorded in the published *Report on the Examination*.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the *Report on the Examination*.

• CIE will not enter into discussions or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the November 2003 question papers for most IGCSE and GCE Advanced Level syllabuses.

Grade thresholds taken for Syllabus 0653 (Combined Science) in the November 2003 examination.

	maximum	minimum mark required for grade:				
	mark available	А	С	Е	F	
Component 1	40	-	27	22	19	
Component 2	60	-	34	24	20	
Component 3	80	58	36	-	-	
Component 5	30	21	14	9	7	
Component 6	60	44	35	25	20	

The threshold (minimum mark) for B is set halfway between those for Grades A and C.

The threshold (minimum mark) for D is set halfway between those for Grades C and E.

The threshold (minimum mark) for G is set as many marks below the F threshold as the E threshold is above it.

Grade A* does not exist at the level of an individual component.



INTERNATIONAL GCSE

November 2003

MARKING SCHEME

MAXIMUM MARK: 40

SYLLABUS/COMPONENT: 0653/01

COMBINED SCIENCE Paper 1 (Multiple Choice)

Mark Scheme **IGCSE EXAMINATIONS – NOVEMBER 2003**

Page 1

www.papaCambridge.com Question Question Key Key Number Number 1 21 С в 2 В 22 Α 3 В 23 D С 4 24 В 5 D 25 Α В 26 В 6 7 В Α 27 8 С 28 D 9 С 29 Α С 10 D 30 С 31 11 D 12 В 32 Α С 13 33 Α С С 14 34 D 15 35 Α D 36 16 В С 17 37 D С 18 Α 38 С В 19 39 20 В 40 Α

TOTAL 40

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INTERNATIONAL GCSE

MARK SCHEME

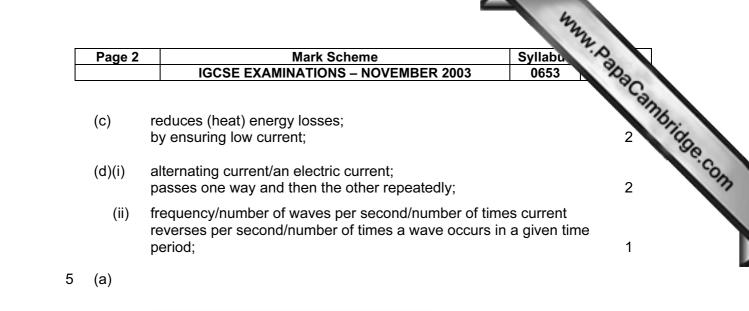
MAXIMUM MARK: 60

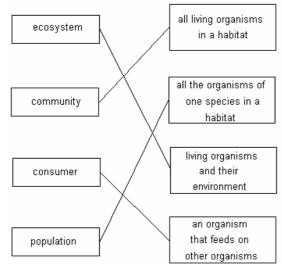
SYLLABUS/COMPONENT: 0653/02

COMBINED SCIENCE Paper 2 (Core)

	Page 1	Mark Scheme Syllabu	Q.
		IGCSE EXAMINATIONS – NOVEMBER 2003 0653	Star.
			PHA
	(a)(i)	kinetic/movement \rightarrow electrical;	orid
	(ii)	chemical (potential); electrical;	www.papacambridge. 2
	(b)	can be used again/replaced/will not run out/ replicated if qualified e.g. wood;	1
	(c)	fission means nuclei break (into smaller pieces); (accept atom splits into daughter nuclei) fusion means nuclei join together;	2
2	(a)(i)	carry oxygen;	1
	(ii)	not enough oxygen (in blood/for cells); cells cannot respire (enough)/reduced respiration rate; so cannot release energy (from food);	2max
	(b)(i)	iron is needed for making haemoglobin; red blood cells contain haemoglobin;	2
		[e.g. iron important part of haemoglobin in red blood cells would g	get both]
	(ii)	red meat/liver; green vegetables/named green vegetable; foods cooked in iron utensils; egg; (not egg white) chocolate; cereals;	
		nuts;	2max
3	(a)(i)	(each molecule contains) two chlorine atoms bonded together; [it is a molecule of two chlorine atoms would be acceptable]	1
	(ii)	(17) protons;	
		18 (neutrons); (ignore any figure)electron(s);	3
	(b)	kills micro-organisms/bacteria/pathogens/sterilises the water; to make the water safe/not harmful; [kills harmful micro-organisms scores both] [reject viruses, germs, algae, bugs]	2
4	(a)	(ignore step up or down) transformer;	1
	(b)	$\frac{v_P}{v_S} = \frac{N_p}{N_s};$	
		[accept if set out correctly using words]	

[accept working alone or answer alone for second mark but reject if incorrect answer given]





6

	all three right = 2 marks; one right = 1 mark;	2
(b)(i)	food supply/temperature/rainfall/vegetation/diet;	1
(ii)	breed them/strong implication that breeding has occurred; see if the number of stripes is inherited/compare stripes between generations	
	look for patterns in stripes between parent and offspring;	2
(a)(i)	nitrogen;	1
(ii)	(argon) is a noble gas/in Gp 0/inert gas/(atoms) have full outer shell;	1
(iii)	carbon monoxide/nitrogen oxides are released/in the exhaust; these are toxic/can kill (if breathed in); [reject anything to do with carbon dioxide]	2
(b)(i)	same number of atoms of each element on both sides/owtte; [allow numerical response e.g. 4 H's and 2 O's on each side]	1
(ii)	any proportions in a mixture/ fixed ratios in a compound/owtte;	
	or	
	mixture retains properties of components/ compound is new substance/has different properties from elements;	

Page 3	Mark Scheme Syllabu	· Q
	IGCSE EXAMINATIONS – NOVEMBER 2003 0653	Nº Da
	or	
	atoms of different elements not bonded (to each other) in mixture/ are bonded in compound;	WW Baba
	or	
	mixture is separable by physical means/named example/compou cannot be separated this way;	nd 1
(c)	using hydrogen, waste product is water (which is non-polluting); using hydrocarbons waste products include CO ₂ /CO/NOx/other which are harmful;	2
	[so burning H ₂ does not produce CO etc. gets 1 mark]	
(a)	speed = distance ÷ time (or appropriate symbols);	
(4)	288 ÷ 1.2/240;	2
(b)(i)	A/0,0/origin;	1
(ii)	B to C/D to E/F to G; line is horizontal/flat/of gradient zero; [allow line is <i>straight</i> only if qualified by saying that speed constant at 100 mph/40 mph]	2
(iii)	C to D;	1
(a)	one mark for each correct label;;;;	4
(b)(i)	goblet cells make mucus; [ignore excrete mucus] mucus traps bacteria/dirt; cilia sweep (mucus) upward; keeping bacteria/dirt out of lungs; [cilia sweep out bacteria/dirt scores both]	2max
(ii)	cilia stop working/are damaged; more mucus is made/mucus continues to be made; (excess) mucus collects in lungs/bronchioles; bacteria breed in it; [bacteria are not swept out and stay in lungs causing disease with the first point scores both]	2max
(a)	aluminium; electrolyte; positive; cathode;	4
(b)(i)	red-brown/orange/pink/brown/copper solid forms/ (green) gas bubbles/a gas is given off/pungent odour (of chlorine) solution loses colour;	/
(ii)	(copper chloride \rightarrow) copper + chlorine;	1



INTERNATIONAL GCSE

MARK SCHEME

MAXIMUM MARK: 80

SYLLABUS/COMPONENT: 0653/03

COMBINED SCIENCE Paper 3 (Extended)

	Page 1	Mark Scheme Syllabu IGCSE EXAMINATIONS – NOVEMBER 2003 0653	2
			C.
1	(a)	amylase/carbohydrase; fats/lipids; amino acids/peptides/polypeptides; stomach/small intestine/duodenum/ileum;	A A A A A A A
	(b)(i)	attracts animals/animals eat them; animals carry, fruits/seeds, to a new place;	2
	(ii)	crush/chop (seed)/cut in half/make a solution; add biuret (solution)/add copper sulphate and pot. hydroxide (solution); lose this mark if heated purple if protein present ;	3
2	(a)(i)	ray reflects at surface of mirror ; (lose this mark if arrows going wrong way) straight lines drawn; angles approximately equal and ray enters eye;	3
	(ii)	image (of lamp) shown behind the mirror; approximately 39 mm behind it and level with the lamp;	2
	(b)(i)	speed/both tranverse/both can travel through vacuum;	1
	(ii)	frequency/wavelength;	1
3	(a)(i)	limestone/calcium carbonate;	1
	(ii)	iron oxide + carbon monoxide/carbon \rightarrow iron + carbon dioxide;;	2
	(iii)	oxygen removed (from iron oxide)/electron gained by iron;	1
	(b)(i)	Fe ³⁺ ; <u>balancing of charges</u> used as evidence;	2
	(ii)	working; <i>e.g. (56 x 2) + (16 x 3)</i> 160; (ignore units)	2
4	(a)(i)	light (intensity)/air movement/humidity;	1
	(ii)	so that light could enter; for photosynthesis;	2
	(b)(i)	 water was lost from plants A and B; by transpiration/by evaporation; as water vapour/from the leaves/through stomata; water could not escape from C/words to that effect ; 	ах 3
	(ii)	 it was in warmer conditions; which increased transpiration; as water evaporated (in the leaves) more rapidly; water (vapour) diffused (out of the leaf) more rapidly; 	ax 3

Page 2	Mark Scheme Sylla	bu . S
	IGCSE EXAMINATIONS – NOVEMBER 2003 065	53 Yac
(iii)	 1 plant had closed its stomata; 2 because plant was running short of water; 3 most of the water in the soil had been lost; 4 less difference in water concentration between the leaf and 5 so smaller diffusion gradient; 	the air; max 2
5	 1 particles touch in solid and liquid; 2 particles widely spaced in gas; 3 particles vibrate in solid; 4 particles move within liquid; 5 particles move more freely in gas; 6 solid particles, strongly attracted to each other/strong forces 7 liquid particles, strongly attracted to each other/strong forces 8 gas particles, not attracted to each other/no forces between 	s between them; s between them;
6 (a)(i)	methane + oxygen \rightarrow carbon dioxide + water;;	2
(ii)	(fractions) used as fuel/burnt; sulphur converted to sulphur dioxide/sulphur dioxide is formed (sulphur dioxide may) produce acid rain/be harmful if breathed damage stonework/kill plants/acidify lakes/kill fish ;	
(b)	mix with bromine (solution/water); solution decolourised;	2
(c)(i)	5;	1
(ii)	each ethene molecule has two carbons/ref to 10 ÷ 2;	1
7 (a)(i)	A has twice the mass of B; gravity was the same for both so equal weight means equal m similar argument;	nass/ 2
(ii)	60 (cm); explanation using idea of moments;	2
(b)(i)	shown on mid line in both; in bottom half of flask and top half of glass;	2
(ii)	1 conical flask has wider base; 2 so more difficult to move c of g outside this; 3 conical flask has lower centre of gravity;	max 2
8 (a)	organ;	1
(b)(i)	 1 at least 6 of these cells drawn (shape recognisably similar); 2 each cell touching at least one other; 3 all orientated in the same direction; 4 more than one 'row' of cells shown; 	max 3
(ii)	1 (onion epidermal cell) has no chloroplasts/chlorophyll; 2 it is underground/does not receive light; 3 chloroplasts/chlorophyll, absorb light;	

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	IGCSE EXAMINATIONS – NOVEMBER 2003	0653
O(-)(i)	hattan asid .	any .
9 (a)(i)	battery acid ;	"I'd
(ii)	$H^+ + OH^- \rightarrow H_2O;;$	Vilabu 0653 ABACAMbridge 2
(b)	sodium chloride;	
	carbon dioxide;	
	water;	
	any order	3
(c)	1 mix acid and copper oxide;	
	2 warm/stir;	
	3 copper oxide in excess/add copper oxide until no more disso	olves;
	4 filter/decant;	
	5 allow, filtrate/solution, to form crystals or evaporate some wa	ater; 4 max
10(a)	1 causes ionisation;	
	2 damages DNA/chromosomes/genes;	
	3 causes mutations;	
	4 destroys/damages, cells;	
	5 causes cancer;	2 max
	it is charged;	
(b)	it is charged; positively charged;	
(b)	0	ld); 2 max
(b)	positively charged;	ld); 2 max



INTERNATIONAL GCSE

MARKING SCHEME

MAXIMUM MARK: 30

SYLLABUS/COMPONENT: 0653/05

COMBINED SCIENCE Practical

Р	age 1	Mark Scheme	Syln Syln
		COMBINED SCIENCE – NOVEMBER 2003	0653
1 (a)		temperatures at time 0 mins included time 0-10 mins all temps. decrease and B less than A	Sylle Babacambrid
(b)		suitable scale for temperature correct plotting of points two smooth curves drawn	3
	(ii)	tube A	1
(c)		yes explanation involving results or in terms of heat transfer	2
(d)		lines continued as smooth curves	1
			total 10
2 (a)		blue colour	1
(b)	(i)	white ppt. chloride ion	2
	(ii)	litmus turns blue ammonia gas	2
(c)		test for copper ion correctly described	3
(d)		ammonium chloride and copper	2
			total 10
3 (c)(Table correctly calculating mass of nitrate/100g at least three temperatures recorded temperatures within 4°C of expected values 70-78, 62-70, 55-63, 50-58	1 1 4
(e)		correct plotting curve drawn smooth curve	3
(f)		correctly read from graph	1
(-)		······································	total 10



INTERNATIONAL GCSE

MARKING SCHEME

MAXIMUM MARK: 60

SYLLABUS/COMPONENT: 0653/06

COMBINED SCIENCE Alternative to Practical

Page 1	Mark Scheme	S
	IGCSE EXAMINATIONS – NOVEMBER 2003	

1. (a)	Average values correct as in table.	(-1 for each error, 2 errors = 0 marks)	
(۵)	, worage values control as in tablet		-

			-	www.
)	1		Scheme Sy	h A per
			DNS – NOVEMBER 2003	653 23
			table. (-1 for each error, 2 errors	4 653 er 653 er 655 er 60 e 60 e
	alcol	hol concn. /%	average heart rate per minute	°.C.
		0	210	ON
		1	192	
		2	174	
		3	146	
		4	92	
		5	46	
		6	34	
		7	24	
		8	18	

[2]

(b)	suitable scales (1) points plotted correctly (1) smooth curve drawn (1)	[3]
(c)(i) (ii)	(gradual) fall in heart rate (1) steeper fall than in (i) (1)	[2]
(d)	slower reaction/reaction time increased	[1]
(e)(i) (ii)	counting error/variation in individual daphnia/warming effect of light different temperatures/ any other appropriate reason longer count time/repeat several times at each alcohol strength/ check temperatures/any other appropriate (any one)	[1] [1]
	Lotal 10 ma	irke
	Total 10 ma	irks
2. (a)	25, 3, 44, cm ³	[3]
2. (a) (b)(i)		_
	25, 3, 44, cm ³	[3]
(b)(i)	25, 3, 44, cm ³ copper or zinc, (no reaction with water) iron (1)	[3] [1]

Total 9 marks

Page	2 Mark Scheme Syle	A per
	IGCSE EXAMINATIONS – NOVEMBER 2003 0655	SP3
		Sen .
8. (a)	70, 62, 55°C	n, Baha Cambridge [3] [1]
(b)	140 g	[1]
(c)	points plotted (2) (-1 for each error) smooth curve (not straight line) (1)	[3]
(d)	40g of potassium nitrate in 100g water at 60°C	[1]
(e)	heat to evaporate (1) allow to cool (1)	[2]
	Total 10	marks
l. (a)(i) (ii)	57 43	[2]
(b)	Table with 3 columns correctly headed and 2 rows (or vice versa), (data correctly entered (1) (-1 overall if 0 time omitted)	(1) [2]
(c)	tube A	[1]
(d)	(yes) (no mark for this) A stayed warm for longer/surrounding tubes acted as insulation/ any reference to mechanism of heat loss/smaller difference in temperature across the wall of tube A compared with tube B	[3]
(e)	repeat and average/put all tubes in a water bath at first/measure volumes accurately/any sensible suggestion (any 2)	[2]
	Total 10	marks
5. (a)	test 1 carbon or copper oxide test 3 not a carbonate test 4 chloride (ions)	
	test 5 ammonia	[4]
(b)	fumes with HC1	[2]
(c)(i) (ii)	light (1) blue precipitate (1) deep (1) blue solution(1) (any 3 points)	[3]
(d)	ammonium chloride copper oxide	[2]
	Total 11	

Page 3	Mark Scheme Syln		er er
	IGCSE EXAMINATIONS – NOVE	MBER 2003 0653	Pacan
	adio (wave) sound (wave)		oapaCambridge.co.
(b)	The further away the source, the weaker	is the sound OWTTE	[1]
()()	3.0 s 3.8 +/- 0.1s		[2]
(d)(i)	1000/3 = 333 m/s		[1]
(ii)	1000/3.8 = 263 m/s		[1]
• •	The first (1), because the other one may of the observer (1) OWTTE	be affected by the responses	; [2]
(f)	repeat the experiment and average the re	esults	[1]