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COMBINED SCIENCE

0653/01

Paper 1 Multiple Choice

October/November 2005

45 minutes

Multiple Choice Answer Sheet Additional Materials:

Soft clean eraser

Soft pencil (type B or HB is recommended)

READ THESE INSTRUCTIONS FIRST

Write in soft pencil.

Do not use staples, paper clips, highlighters, glue or correction fluid.

Write your name, Centre number and candidate number on the answer sheet in the spaces provided unless this has been done for you.

There are **forty** questions on this paper. Answer **all** questions.

For each question there are four possible answers A, B, C and D. Choose the one you consider correct and record your choice in soft pencil on the separate answer sheet.

Read the instructions on the answer sheet very carefully.

Each correct answer will score one mark. A mark will not be deducted for a wrong answer.

Any rough working should be done in this booklet.

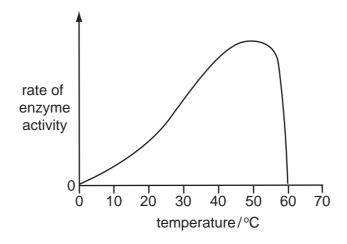
A copy of the Periodic Table is printed on page 20.

1 Which pair of features is found in plant cells but **not** in animal cells?

Α	cell membrane	cell sap
В	cell sap	cell wall
С	cell wall	nucleus
D	nucleus	cell membrane

- 2 Which part of a plant cell is partially permeable?
 - A cell membrane
 - B cell wall
 - **C** chloroplast
 - **D** nucleus

3 The diagram shows how the activity of an enzyme varies with temperature.



At which temperature is this enzyme completely denatured?

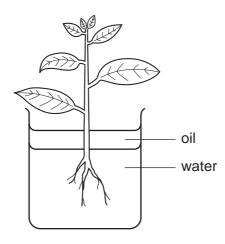
- **A** 0°C
- **B** 40 °C
- **C** 50 °C
- **D** 60 °C

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4 The drawing shows a plant in a container of water. There is a layer of oil on top of the stops the water evaporating. The apparatus weighs 300 g.

After two hours it weighs 296 g.



What is the rate of transpiration?

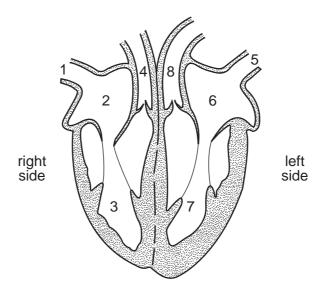
- A 150 g water/hour
- B 148 g water/hour
- C 4g water/hour
- D 2g water/hour
- 5 Which diagram represents the digestion of food molecules in the alimentary canal?





- 6 Which structures in the human breathing system contain goblet cells and cilia?
 - A alveoli and bronchi
 - **B** alveoli and pleural membranes
 - C bronchi and trachea
 - **D** pleural membranes and trachea

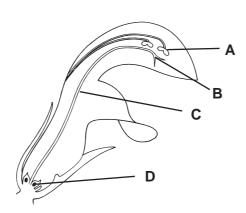
7 The diagram shows a section through a human heart.



In which order does oxygenated blood pass through the heart?

- $\textbf{A} \quad 1 \rightarrow 2 \rightarrow 3 \rightarrow 4$
- $\textbf{B} \quad 4 \rightarrow 3 \rightarrow 2 \rightarrow 1$
- $\textbf{C} \quad 5 \rightarrow 6 \rightarrow 7 \rightarrow 8$
- $\textbf{D} \quad 8 \rightarrow 7 \rightarrow 6 \rightarrow 5$
- 8 What causes the signals passing along the nerves to slow down?
 - A drinking alcohol
 - B eating fat
 - **C** running
 - **D** smoking
- **9** The diagram shows a section through a flower.

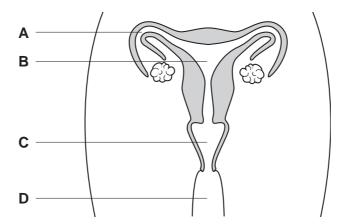
Where does fertilisation occur?



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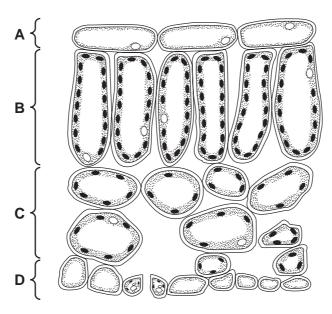
10 The diagram shows the human female reproductive system.

If a woman uses an IUD (intra-uterine device) as a contraceptive, where would it be placed



11 The diagram shows a section through a leaf.

During photosynthesis, where would the greatest conversion of light energy to chemical energy take place?



12 What can cause animals of the same species to vary?

	genes	environment
Α	✓	✓
В	✓	X
С	x	✓
D	×	X

How many consumer levels are there?

- Α 1
- В 4
- C 5
- D 6
- 14 Which fact about crude oil shows that it is a mixture?
 - A Crude oil can be burned as a fuel.
 - Crude oil can be separated into fractions by distillation.
 - Crude oil is a fossil fuel formed over millions of years. C
 - Crude oil is a thick, black liquid. D

A 0-C-C

$$C = C$$

$$C = C$$

16 A solid is ionic.

Which property confirms this fact?

- A its behaviour as an electrolyte
- B its melting point
- **C** its solubility in water
- **D** the shape of its crystals

www.PapaCambridge.com 17 The diagram shows a solid element dropped into a bowl of water. The element cate burns with a lilac flame.



What is the element?

- aluminium
- В magnesium
- C potassium
- D sodium
- **18** Element **X** has a high melting point and forms a green chloride.

Where in the Periodic Table is **X** most likely to be found?

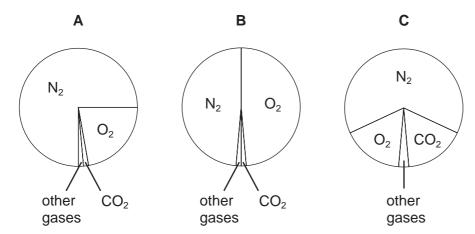
- Group O Α
- В Group I
- С Group VII
- Transition elements D
- **19** The diagram shows a lorry delivering a large container of a corrosive chemical to a factory.



Which metals are used for the lorry and for the container?

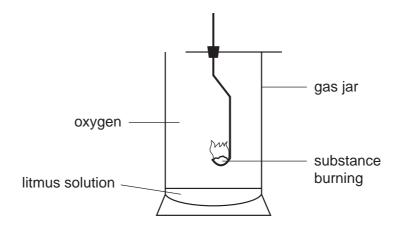
	lorry	container
Α	aluminium	stainless steel
В	mild steel	mild steel
С	mild steel	stainless steel
D	stainless steel	mild steel

20 Which pie chart correctly shows the proportions of gases in the air?



D N₂ O₂ Other gases

21 The diagram shows an experiment on combustion.

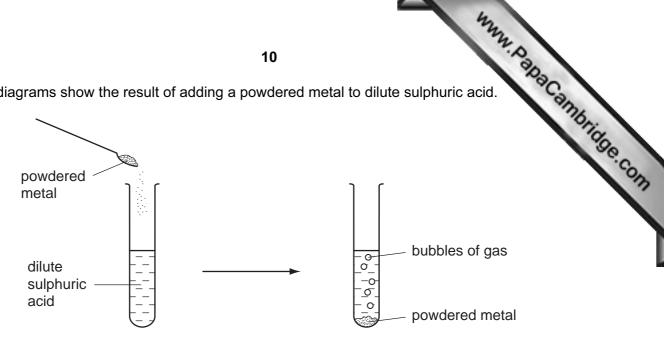


The litmus solution turns red.

Which substance is burning?

- A copper
- **B** magnesium
- C sulphur
- **D** zinc

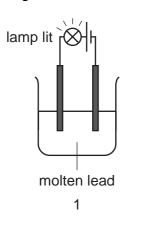
22 The diagrams show the result of adding a powdered metal to dilute sulphuric acid.

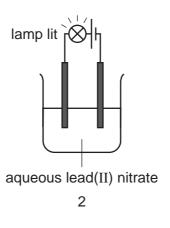


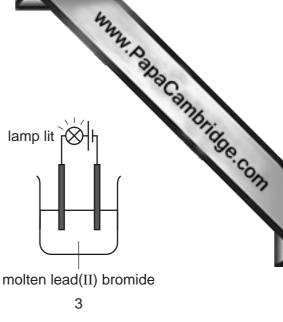
Which of the metals copper, magnesium and zinc react in this way?

- copper only
- В copper and magnesium only
- magnesium and zinc only C
- zinc only D
- 23 Which of the reactions shown is a thermal decomposition?
 - calcium carbonate → calcium oxide + carbon dioxide Α
 - methane + air → carbon dioxide + water В
 - C sodium carbonate + hydrochloric acid → sodium chloride + water + carbon dioxide
 - sodium hydroxide + hydrochloric acid → sodium chloride + water D

24 The diagram shows the results of three experiments.



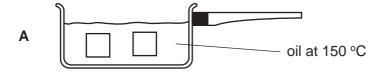


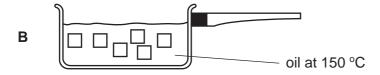


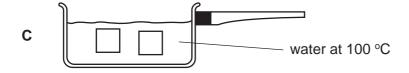
In which experiment is an electrolyte present?

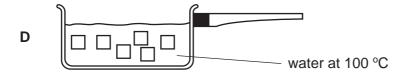
- A 1 and 2 only
- B 1 and 3 only
- C 2 and 3 only
- **D** 1, 2 and 3
- 25 A sweet potato is cut into pieces and cooked.

In which pan does the potato cook most quickly?

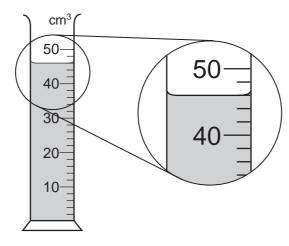








- 26 Which structure shows a polymer that is also a hydrocarbon?
 - A H-C-C-H
 - B C=C
- 27 Which equation shows the complete combustion of ethane, C₂H₆?
 - $\textbf{A} \quad C_2H_6 + 3O_2 \rightarrow CO + CO_2 + 3H_2O$
 - $\textbf{B} \quad 2C_2H_6 + 3O_2 \! \to 4C + 6H_2O$
 - **C** $2C_2H_6 + 5O_2 \rightarrow 4CO + 6H_2O$
 - $\label{eq:D} \textbf{D} \quad 2C_2H_6 + 7O_2 \! \to \! 4CO_2 + 6H_2O$
- **28** A measuring cylinder is used to measure the volume of a liquid.

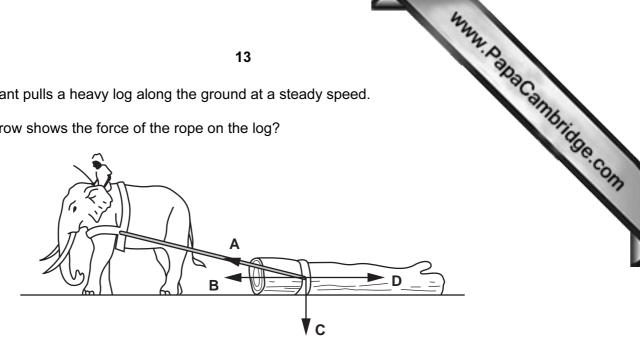


What is the volume of the liquid?

- \mathbf{A} 43 cm³
- **B** 46 cm³
- **C** 48 cm³
- **D** 54 cm³

29 An elephant pulls a heavy log along the ground at a steady speed.

Which arrow shows the force of the rope on the log?



30 The table shows the length of a wire as the load on it is increased.

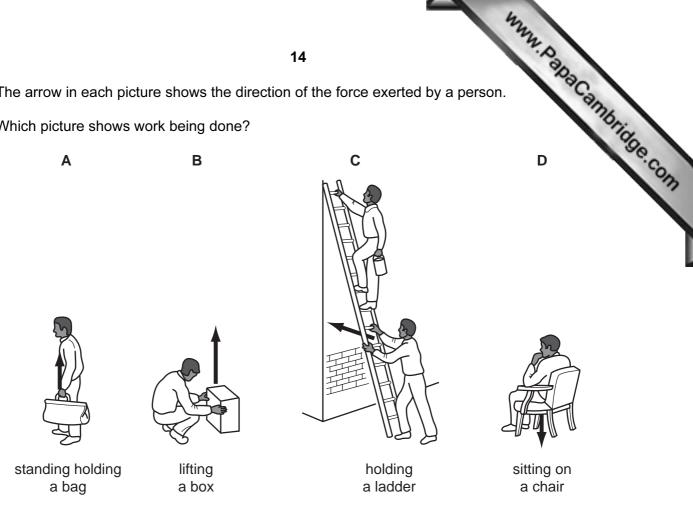
load/N	0	10	20	30
length/cm	50.0	52.1	54.1	56.3

Which subtraction should be made to find the extension caused by the 20N load?

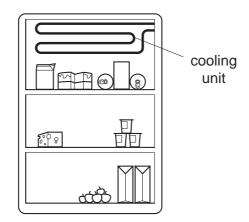
- 54.1 cm 0 cm
- **B** 54.1cm 50.0cm
- **C** 54.1cm 52.1cm
- **D** 56.3 cm 54.1 cm

31 The arrow in each picture shows the direction of the force exerted by a person.

Which picture shows work being done?



32 The diagram shows a cooling unit in a refrigerator.



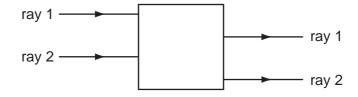
Why is the cooling unit placed at the top?

- Cold air falls and warm air is displaced upwards.
- В Cold air is a bad conductor so heat is not conducted into the refrigerator.
- C Cold air is a good conductor so heat is conducted out of the refrigerator.
- D Cold air stops at the top and so prevents convection.

www.PapaCambridge.com 33 At the end of a long race, a runner is wrapped in a thin, plastic blanket that has a si surface.

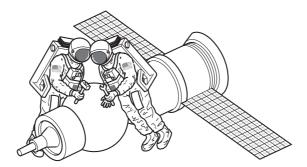
Which type of heat loss is the shiny surface intended to reduce?

- conduction
- В convection
- evaporation
- D radiation
- 34 Rays of light enter and leave a box.



What could be inside the box to make the rays behave as shown?

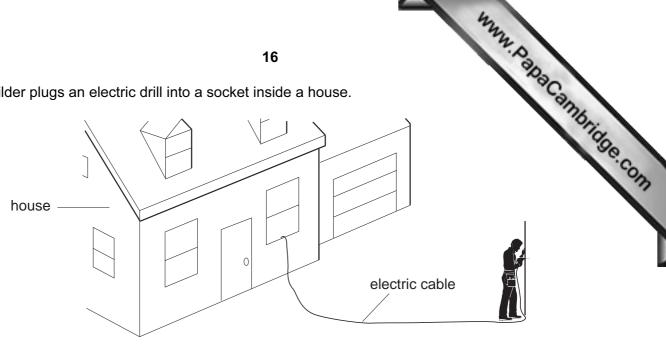
- a converging lens
- a parallel-sided glass block
- a plane mirror C
- a triangular prism D
- 35 Two astronauts without radios can only communicate in space if their helmets are touching. There is no air in space.



What does this show about sound?

	through a solid	through a vacuum
Α	can travel	can travel
В	can travel	cannot travel
С	cannot travel	can travel
D	cannot travel	cannot travel

36 A builder plugs an electric drill into a socket inside a house.



He uses the drill outdoors. It starts to rain heavily.

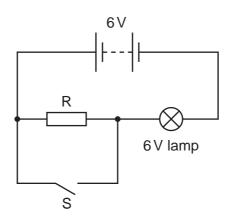
Why is it dangerous to continue using the electric drill in the rain?

- The drill could give the builder an electric shock. Α
- The drill could overheat. В
- C The fuse could blow.
- The rain could rust the drill. D
- 37 The table shows the voltage and current ratings for four electric heaters.

Which heater has the least resistance?

	voltage/V	current/A
Α	110	5.0
В	110	10
С	230	5.0
D	230	10

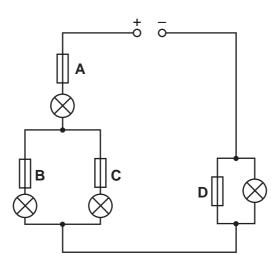
www.PapaCambridge.com 38 When the circuit shown is connected with switch S open, the 6V lamp glows.



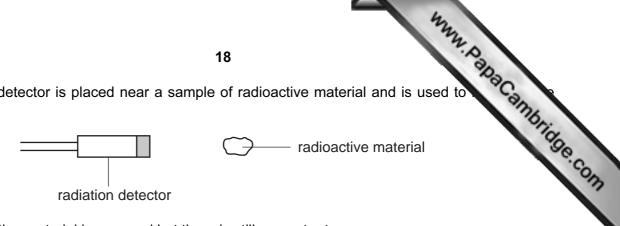
What happens to the brightness of the lamp when switch S is closed?

- It becomes brighter. Α
- В It remains the same.
- C It becomes dimmer.
- It goes off. D
- **39** In the circuit shown, one of the fuses blows and all the lamps go out.

Which fuse blows?



40 A radiation detector is placed near a sample of radioactive material and is used to count rate.



The radioactive material is removed but there is still a count rate.

Why is this?

- It takes a long time for all emissions from the material to reach the detector.
- В The detector has become radioactive.
- C The radioactive material has not finished decaying.
- D There is always some background radiation.

19

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DATA SHEET The Periodic Table of the Elements	
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							1	
0	4 He Helium	20 Ne Neon	40 Ar Argon	84 Kr Krypton 36	131 Xe Xenon	Rn Radon 86		175 Lu Lutetium
\		19 T Fluorine	35.5 C1 Chlorine	80 Br Bromine	127 I lodine	At Astatine 85		173 Yb Ytterbium
N		16 O Oxygen 8	32 S Sulphur	79 Se Selenium 34	128 Te Tellurium	Po Polonium 84		169 Tm Thulium
٧		14 N itrogen 7	31 P Phosphorus 15	75 AS Arsenic 33	122 Sb Antimony 51	209 Bi Bismuth		167 Er
2		12 C Carbon 6	28 Si Silicon	73 Ge Germanium 32	119 Sn Tin	207 Pb Lead		165 Ho Holmium
Ξ		11 Boron 5	27 A 1 Aluminium 13	70 Ga Gallium 31	115 In Indium	204 T 1 Thallium		162 Dy Dysprosium
				65 Zn Zinc 30	112 Cd Cadmium 48	201 Hg Mercury 80		159 Tb Terbium
				64 Copper	108 Ag Silver 47	197 Au Gold		157 Gd Gadolinium
				59 Nickel	106 Pd Palladium	195 Pt Platinum 78		152 Eu Europium
				59 Co Cobalt	103 Rh Rhodium	192 Ir Iridium 77		Samarium
	T Hydrogen			56 Fe	Ruthenium	190 Os Osmium 76		Pm Promethium
				Mn Manganese	Tc Technetium 43	186 Re Rhenium 75		144 Nd Neodymium
				52 Cr Chromium 24	96 Mo Molybdenum 42	184 W Tungsten 74		141 Pr Praseodymium
				51 V Vanadium 23	93 Nob ium 41	181 Ta Tantalum 73		140 Cer ium
				48 T Titanium	2r Zirconium 40	178 Hf Hafnium 72		1
				Scandium 21	89 ≺ (trium	139 La Lanthanum 57 *	AC Actinium 89	l series eries
=		9 Be Berylium	24 Mg Magnesium 12	40 Ca Calcium	Strontium	137 Ba Barium 56	226 Ra Radium 88	*58-71 Lanthanoid series 90-103 Actinoid series
_		7 Li Lithium	23 Na Sodium	39 K Potassium 19	85 Rb Rubidium 37	133 Csesium 55	Fr Francium 87	*58-71 L
		1	III IV V VII VIII VIII	III IV VII VII	III III	1 1 1 1 1 1 1 1 1 1	1 1 1 1 1 1 1 1 1 1	1 1 1 1 1 1 1 1 1 1

	Md No Lr	2
175 Lu Lutetium	د ع	Lawrenciu 103
173 Yb Ytterbium	0Z OZ	Nobelium 102
169 Tm	PW	Mendelevium 101
167 E rbium	E P	100
165 H olmium	9	Einsteinium 99
162 Dy Dysprosium	5	Californium 98
159 Tb		Berkelium 97
157 Gd Gadolinium	Cm Cm	Curium 96
152 Eu	Am	Americium 95
150 Sm Samarium	Pu	Plutonium 94
Pm Promethium	N D	Neptunium 93
144 Neodymium	60 238 C	Uranıum 92
141 Pr	. Pa	Protactinium 91
Cerium	232 Th	l horium 90

b = proton (atomic) number

Key

a = relative atomic mass X = atomic symbol