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## **UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**International General Certificate of Secondary Education** 

## MARK SCHEME for the May/June 2007 question paper

## 0653 COMBINED SCIENCE

0653/03

Paper 3 (Extended Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

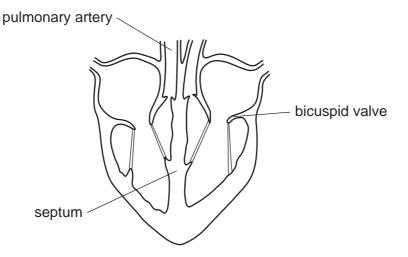
All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

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(a) one mar	k for each correct label ; ; ;	Cambridge
	pulmonary artery	Talle
		S.COM



**(b)** contains more muscle;

to provide, more force / high(er) pressure; to push blood further round the body; right ventricle only pushes blood to lungs;

[max 2]

[3]

(c) aorta wall is thicker;

aorta lumen is smaller; aorta wall is more elastic;

vein has valves;

[max 2]

(d) muscle does not get oxygen;

so cannot respire;

so cannot contract;

[max 2]

2 (a)  $A_2 = 0.015 A$ 

 $A_3 = 0.15 A$ ;

 $V_1 = 3 V;$ 

 $V_2 = 3 V$ ;

[2]

**(b) (i)** Vp/Vs = Np/Ns; (or rearranged)

25 000 / 400 000 = 20 000 / Ns; (or alternative working method)

 $(Ns) = 320\ 000$ ;

[3]

(ii) changing current causes changing magnetic field;

changing magnetic field induces voltage in secondary coil;

[2]

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Page 3	Mark Scheme	Syllabus	2 er
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3	(a)	in mixture: (particles / molecules) of the different gases are not bonded / gases have the same (chemical) properties as when not mixed / any proportions are possible / can be separated by physical methods;		ambridge [1]
	(b)	(i)	carbon monoxide / steam ; carbon / soot ;	[2]
		(ii)	shake with limewater; goes cloudy; shows carbon dioxide;	
			test with, cobalt chloride (paper) / anhydrous copper sulfate; goes from, blue to pink / white to blue; shows water;	[max 4]
	(c)	(i)	KOH;	[1]
		(ii)	$H^+ + OH^- \longrightarrow H_2O$ ;; (left hand side and right hand side)	[2]
4	(a)	more species in the rainforest ; of plants and bats / figures quoted ;		[2]
	(b)	14 species found <u>only</u> in the rainforest ;		[1]
	(c)	bats go to flowers for nectar; pollination; (not 'pollen dispersed') ref. to fertilisation following pollination; beans form after, fertilisation / pollination;  stops rain hitting the ground directly; more roots to soak up the water; less run-off; roots hold the soil;		[max 2]
	(d)			[max 2]
	(e)	(i)	it reduces the number of pods infected; compared with, the control / no treatment; but does not completely eliminate infection / use of figures;	[max 2]
		(ii)	takes time for the b.c. fungus to work; any other relevant suggestion (related to a particular stage of the curve);	[2]

Page 4	Mark Scheme	Syllabus	
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5 (a) (i) B because the line is horizontal;

- (ii) change of speed = 0 to 28 s / a = (v-u)/t; 1.4 m/s<sup>2</sup>;
- (iii) force = mass x acceleration; =  $1400 \times 1.4 = 1960 \text{ N}$ ; [2]
- (iv) working; 1036 m; [2]
- (b) (i) road material expands when hot; [1]
  - (ii) rubber, can be compressed / is elastic / can stretch; [1]
- (c) (less) explanation relating to resistances in parallel; [1]
- (a) any group 1 or calcium / strontium / barium;
   reference to hydrogen;
   (only) these metals produce hydrogen (rapidly) / at room temp / in cold water, when they react with water;
   [3]
  - (b) (i) oxidation / redox; [1]
    - (ii) oxygen / water / substances from the air, have reacted with the, iron / steel;rust is (hydrated) iron oxide;ref. to the combined mass of iron and other substances;[max 2]
- 7 (a) similar shape with optimum at lower temperature; [1]
  - **(b)** as temperature rises (below optimum) movement of molecules increases; more frequent collisions / more energetic collisions; between enzyme and substrate;

beyond optimum enzymes denature; they are proteins; lose their shape at high temperatures;

(c) plant cells at lower temperatures / plant enzymes work better where they live; [1]

[max 4]

Page 5	Mark Scheme	Syllabus
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8	(a)	(i)	infra-red / microwaves ;	ambride
		(ii)	300 000 000 m/s ;	195
		(iii)	frequency / wavelength;	[1]
	(b)	(i)	breakdown of an (unstable) nucleus ;	[1]
		(ii)	Geiger-Müller tube ;	[1]
		(iii)	moves, towards negative plate / away from positive plate; moves, towards positive plate / away from negative plate; unaffected by plates;	[3]
		(iv)	ionises; damages cells or DNA or mutates; cancer; skin burns; radiation sickness;	[max 2]
9	(a)	(i)	bromine ;	[1]
		(ii)	to form an electrolyte / to melt the lead bromide; enables <u>ions</u> to move; so that an electric current will flow through it;	[max 2]
	(b)	(i)	+2 ; two -1 bromide ions balance the charge on the lead ion ;	[2]
		(ii)	(36) because bromine atom has 35 electrons / same number of electrons as proton nur has gained one electron / has single negative charge so one extra electron ;	mber ; [2]
	(c)	(i)	shared pair ; all other non-bonding electrons shown ;	[2]

[2]

(ii) Si +  $2Cl_2$   $\longrightarrow$  SiC $l_4$ ;; (formula and balanced)