UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

## www.papacambridge.com MARK SCHEME for the May/June 2009 question paper

## for the guidance of teachers

## **0653 COMBINED SCIENCE**

0653/02

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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	Mark Scheme: Teachers' version Syllabus IGCSE – May/June 2009 0653	2Do
(a) A ename B dentine C pulp/bl [reject ro	el ; e ; lood vessel/nerve ; pof]	Dana Campings
increase	wn large pieces of food to small ones ; surface area ; t easier for enzymes to act ;	[max 2 ]
(c) calcium/µ D ;	phosphate ;	[2]
		[Total: 7]
<b>(a) (i)</b> all s	ymbols shown ;	
voltr	neter in parallel with lamp only ; ther components in series ;	[3]
	ary current (through lamp)/voltage/PD (across lamp) ;	[1]
[igno	ore refs. to power if with correct statements]	
(iii) (R = = 5.3	3;	[2]
	ept words] ept only suitable symbols, so <i>V/I (= ohms)</i> is accepted but <i>V/A</i> is not]	
(b) damaged	d outer insulation/owtte ;	
short circ	cuit/risk of shock/risk of fire ;	[2]
		[Total: 8]
(a) neon;		
chlorine cobalt ;	,	[3]
<b>(b) (i)</b> 12;		[1]
(ii) carb	on ;	
(nuc	leus/atom has) <u>6 protons</u> /it has a <u>proton number 6</u> ; ept other unambiguous statements]	[2]
(c) hydroger	n ; ate metal e.g. Ca Mg A <i>l</i> Zn Fe ;	
appiopin		
[reject G	acid – could be several correct answers but expect HC $l$ H <sub>2</sub> SO <sub>4</sub> HNO <sub>3</sub> ;	[3]

1 4	Page 3			labus er
			IGCSE – May/June 2009 00	653 7030
(a)	(i)	anth	ner/stamen ;	labus 653 Papacambriog [2]
	(ii)	male	e gametes ;	192
	(iii)		transfer of pollen ; stigma ;	[2]
		ιυa	sugma,	[4]
(b)	(i)	the I	higher the temperature, the more oxygen is used ;	[1]
	(ii)	•	biration ;	
		using	bbic ; g oxygen to produce heat ;	
		by b	preaking down glucose ;	[max 2]
(c)	pho	otosyr	nthesis ;	
<b>\</b> = <i>j</i>	(us	ing) li	ight/sunlight ; ng water and carbon dioxide/correct equation ;	[max 2]
	CON	Dirm	ly water and carbon dioxide/contest equation ,	נוומא בן
(4)	رما	wall	drawn and labelled ;	
(4)	cell	l mem	nbrane labelled immediately inside the cell wall ;	
			drawn and labelled, in the cytoplasm ; ast drawn and labelled, in the cytoplasm ;	
			drawn and labelled ;	[max 4]
				[Total: 13]
(a)	(i)		erage speed =) distance / time ;	
		= 90	00 / 150 = 6 km/h ;	[2]
	(ii)	2 m/s	s ;	[1]
(b)	frict	tion a	ind thrust/upthrust and weight ;	[1]
<i>4</i> \				
(C)		ulatior (pped)	n ; ) air (is an insulator) ;	
	pre	vents	conduction/convection ;	[3]
(d)			= mass ÷ volume / mass = density × volume ;	
	ma	ss = 8	800 × 9 = 7200 (kg) ;	[2]
(e)	(i)	sola	r/sunlight/waves/tides/geothermal/biofuel/hydro (reject nucle	ear); [1]
	<i>(</i> ::)		l/oil/gas/(named) fossil fuel/peat (reject nuclear) ;	[1]
	(ii)	coal	Vollygasi (namou) iossii noipour (rojoor nasioar),	[.]

Page 4		Syllabus Syllabus
	IGCSE – May/June 2009	0653 23
(a)		Syllabus 0653 BabaCannbridge
• •		orig
	P	36
	70–190 °C	
	250–340 °C	
	R below 30 °C	
	$\bigwedge$	
	s	
	190–250 °C	
	(2 marks for 3 correct, 1 mark for 1 or 2 correct)	[2]
(b) (i)	plastic buckets lighter (to carry) ;	
.,	flexible, not bent out of shape in use ;	
	no reaction with content of bucket impermeable ;	
	easily be shaped ;	[max 1]
(ii)	oxygen ;	
. ,	water ;	[2]
(iii)	galvanising/cover in layer of zinc/painting ;	[1]
(iv)	iron ;	[1]
(v)	stainless steel ;	[1]
		[Total: 8]

Page	e 5	Mark Scheme: Teachers' version	Syllabus 0653 Bacanthiuge
		IGCSE – May/June 2009	0653
(a)			an
()		hawks	Dhin
		<b>↑</b>	50
		ا pangolins small birds	
		×	
		ants	
		fungus	
		leaves	
		sms included ;	
		cted correctly by lines ; ct arrow heads ;	[3]
vv			[5]
(b) (i	i) leave	s/trees ;	[1]
	-		
(ii	i) fungu	IS ;	[1]
		s of <u>habitat</u> ; eat ants, which eat leaves ;	
		aves then fewer ants so fewer pangolins ;	[max 2]
			[Total: 7]
			[10tal. 7]
(a) (	(i) weigh	nt of empty lift = 12000 N ;	
(a) (		ined weight = $12800 \text{ N}$ ;	[2]
/:	<b>ii)</b> \// — 「		
()	ii) W = F [acce]	pt (work done =) height × (total) weight]	
	= 12	800 x 9 =115 200 J ;	[2]
	ect to	r incorrect total weight from (i)	
/L.\4	1	·	
(D) (d)	i) vibrat of mo	ions ; lecules/particles ;	
	longit	udinal wave ;	
	comp	ressions and rarefactions ;	[max 2]
(ii	i) loude	r ;	[1]
			[T_4_1. 7]
			[Total: 7]

Page 6	Mark Scheme: Teachers' version	Syllabus er
	IGCSE – May/June 2009	0653 73
(a) A filtratio B evapo	on ; pration/crystallization ;	Syllabus 0653 ObsCambrid
, ,	up the reaction ; s have) a greater surface area (which speeds reaction	on) ; [2]
<b>c) (i)</b> zinc	sulfate ;	[1]
(2 m	opper sulfate + carbon dioxide + water ; arks for 3 correct 1 mark for 2 correct) ct symbols or formulae even if correct]	[2]
(reje	balanced) t have the same number of <u>each type</u> of atom on bo ect same number of atoms needed on both sides) e correct detail e.g. 1 H on left but 2 on right/would i	
	tion is exothermic/heat given out (to surroundings) ;	
(,	<b>.</b>	[.]