WWW. Pals

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2009 question paper for the guidance of teachers

0653 COMBINED SCIENCE

0653/02

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• CIE will not enter into discussions or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the October/November 2009 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

| Page 2 | Mark Scheme: Teachers' version | Syllabus | er er |
|--------|--------------------------------|----------|-------|
| | IGCSE – October/November 2009 | 0653 | 100 |

(a) A contains starch; **B** contains protein; (b) (i) protein; in living organisms; that acts as a catalyst; [max 2] (ii) cannot digest / breakdown starch; (reject food) cannot absorb, starch / sugar / glucose; into the blood; cells do not get sugar; cannot use (starch / sugar) for respiration; [max 3] (iii) genes / chromosomes / DNA / mutation; [1] (reject references to inheritance or blood) (iv) no starch in their food / starch not present in meat; [1] [Total: 9] (a) (i) hydrogen; 2 [1] (ii) sulfur dioxide; reacts with / dissolves in rainwater / forms sulfurous / sulfuric acid; [2] (allow one mark for reference to CO₂ and carbonic acid) (b) (i) covalent; [1] (ii) two oxygen atoms shown joined to central carbon; [2] by double bonds; [Total: 6] 3 (a) D to C to A to B [2] **(b)** alpha radiation completely absorbed by / cannot penetrate paper; [1] (c) mutates (cells); causes cancer; [max 2] radiation burns; (d) shielding (e.g. gloves, lead lined clothes etc.); monitoring (radiation badges etc.); limited exposure time; [max 2] [Total: 7]

| | Page 3 | | Mark Scheme: Teachers' version | Syllabus |
|---|-----------|--|--|----------|
| | | | IGCSE – October/November 2009 | 0653 |
| 4 | 4 (a) (i) | | ease (soil erosion) ; not protected from rain by leaves ; | Cambric |

4 (a) (i) increase (soil erosion); soil not protected from rain by leaves; soil not held by roots; easily washed away;

[max

(ii) decrease (species diversity);loss of habitats;loss of food supplies / disrupts food chains;more hunting (by humans);

[max 2]

(b) (i) oil palm → rats → owls;; (all organisms and energy transfer)

[2]

[1]

(ii) oil palm is producer, rats and owls are consumers; (all organisms required, ignore references to the Sun)

[Total: 7]

5 (a) (i) \rightarrow salt; + water; [2] (allow H_2O)

(ii) hydrogen / H / H⁺;

[1] [1]

(iii) NaOH; (allow NaHO)

[1]

(b) (i) iron; (allow Fe)

(ii) a barrier of zinc / zinc plated onto the steel / covered with zinc; which prevents (the steel from) rusting / keeps air / oxygen and water away from the steel;

[2]

(iii) metals react with acid / the container would soon react and break; (allow corroded / dissolved)

[1]

(c) (i) 9;

[1]

(ii) contains (only) hydrogen and carbon / is made of hydrogen and carbon;

[1]

(iii) words and/or diagram which conveys (many) small molecules / molecules of propene join together / propene monomers link together; to form a (long) chain (molecule);

[2]

(ignore mistakes in displayed formulae if meaning is clear)

[Total: 12]

| | 3 | | | | |
|---|---------|-------------------|--|------------------|----------|
| | Page 4 | 1 | Mark Scheme: Teachers' version | Syllabus | er |
| | | | IGCSE – October/November 2009 | 0653 | |
| 6 | (a) (i) | 15 s ; | | Syllabus 0653 | ambri |
| | (ii) | 30 s; | | | E. Car |
| | (iii) | CD and | I GH / 60 − 80 (s) and 140 − 160 (s) ; | | [1] |
| | | | eed / constant velocity ; balanced forces / equal and opposite forces ; | | [2] |
| | (c) (i) | may co poisono | ntain carbon monoxide / (products of) incomplete cous ; | ombustion ; | [2] |
| | (ii) | | al ; hermal ; / movement / sound / light ; | | [max 3] |
| | | | | [То | tal: 10] |
| 7 | (a) (i) | label to | palisade cell ; | | [1] |
| | (ii) | label to | stoma ; | | [1] |
| | (b) (i) | which is controls | s DNA / genetic information ; s inherited ; s activity of the cell (by controlling enzymes present) enes on chromosomes ; | | [max 2] |
| | (ii) | | s what enters / leaves the cell ; allow direct references to oxygen, water, carbon did | oxide) | [1] |

(c) (i) for photosynthesis;

(ii) xylem (vessel);

(d) (i) gas;

(in which) water is combined with carbon dioxide;

to provide turgor / support / hold shape;

(ii) diffusion / (evapo)transpiration;

(reject evaporation)

[1]

[max 2]

[1]

[1]

[Total: 10]

| Page 5 | | 5 | Mark Scheme: Teachers' version | Syllabus | er |
|--------|--------|---------------------|---|------------|------------------|
| | | | IGCSE – October/November 2009 | 0653 | No. |
| 8 | (a) (i |) con | duction ; | | ding |
| | (ii |) (pla | astic is) good insulator / poor conductor ; | | A. Papa Cambridg |
| | (b) (i |) A , l | pecause it is a (diagram of a) solid / close (packed) and | | [1] |
| | (ii |) B , l | pecause it is a (diagram of a) liquid / close (packed) and | d random ; | [1] |
| | (c) (i | (rej (vol = 5 | nsity =) <u>mass</u> ÷ volume ect unconventional symbols or units in formula) lume is) 200 (cm³) ; 40 ÷ 200 / 2.7 (g / cm³); | | [3] |
| | | (allo | ow ecf for incorrect volume) | | |
| | (ii | 5.4 | N; | | [1] |
| | | | | | [Total: 8] |
| 9 | (a) (i | • | and Y (both needed) ; rbonates react with acid to produce) carbon dioxide ; | | [2] |
| | (ii | | nsition metal compounds are (often) coloured (other tha | n white) ; | [2] |
| | (b) (i |) (str | ong) heat ; | | [1] |
| | (ii | | pper oxide +) carbon; → (copper +) carbon dioxide; ow carbon monoxide and carbon oxide) | | [2] |
| | (iii | , | dation / reduction / redox ; bon gains oxygen / copper oxide loses oxygen ; | | [2] |
| | (iv | | nplete circuit in which the copper forms a part ; | | |

circuit would indicate that copper is a conductor i.e. copper is in series with a power

[2]

[Total: 11]

source and a current indicator;