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UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the May/June 2011 question paper for the guidance of teachers

0653 COMBINED SCIENCE

0653/22

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

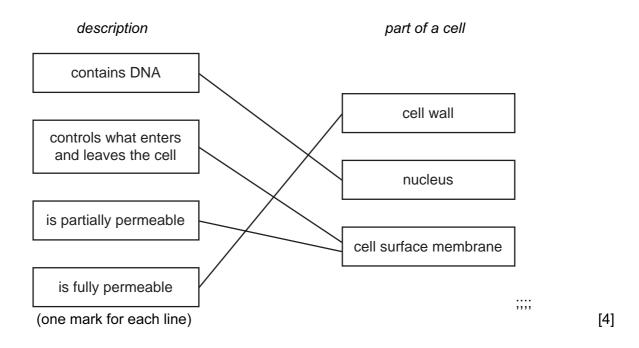
• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2011 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

	Page 2	Mark Scheme: Teachers' version	Syllabus
		IGCSE – May/June 2011	0653
1	(ii) i	gravity/weight; idea of balanced forces/equal and opposite; no <u>acceleration</u> ;	Cambridge con
	(b) (i) :	X on a horizontal part of the graph: (not at 50)	[1]

- (a) (i) gravity/weight;
 - (ii) idea of balanced forces / equal and opposite; no acceleration;
 - [1] **(b) (i) X** on a horizontal part of the graph; *(not at 50)*
 - (ii) Y in correct position; [1]
 - (iii) at end of graph / on the vertical part of graph at 110 s; [1]
 - [Total: 6]

2 (a)



- (b) (i) enzymes; [1]
 - (ii) enzymes, denatured / destroyed; [1]
- (c) (i) increases; steadily / linear, increase; from 0.6 to 1.1 $(g/cm^3)/by 0.5 (g/cm^3)$;

- (ii) these foods contain calcium; needed for bones; [2]
- (iii) any citrus fruit / blackcurrants / other valid food source; [1]

[Total: 11]

[max 2]

	D۵	ge 3	<u> </u>	Mark Scheme: Teachers' version	Syllabus	\ -
	Ра	ge s)	IGCSE – May/June 2011	0653	2
3	(a)	(i) (ii)	hydr	res / pops ; rogen is given off ; A and C did not react / two did not react / cannot d	ecide between A and	ORC AMBRIDGE
	(b)	(i) (ii)	copp	water / calcium hydroxide / slaked lime ; per sulfate + carbon dioxide + water ;; correct scores 2 marks, two correct scores 1 mark)		[1] [2]
			(all c	sorroot socres 2 marks, two correct socres 1 marky		[Total: 6]
4	(a)	(i)		s + arrows showing upward movement from the hea s + arrows showing downward movement round the		[2]
		(ii)	hotte hot a	est A ; est C ; air rises / cold air sinks; air less dense than cold air (vice versa);		[max 3]
	(b)		oR risk	of fire / overheating; ause socket overloaded; of electrocution / shock (if touched); ause insulation damaged / live wires exposed;		[max 2]
	(c)	(i)	no C	CO ₂ production / no global warming / no depletion of t	ossil fuels ;	[1]
		(ii)	radia	ation leaks / nuclear accidents / problems of storage	of nuclear waste ;	[1]
						[Total: 9]
5	(a)	(i)	peta	als / nectaries ;		[1]
		(ii)	anth	ner/stamen ;		[1]
	(b)	(i)	carb usin	tosynthesis ; oon dioxide combined with water ; g <u>energy</u> from (sun)light ; ergy) captured by chlorophyll ;		[max 3]
		(ii)		respiration/for energy/to make nectar/any name cess;	ed energy-consuming	[1]

[Total: 6]

	Page 4	Mark Scheme: Teachers' version	Syllabus Y
		IGCSE – May/June 2011	0653
6	(ii) refe time acti	I/peat; erence to: escale / time to renew; on of, heat / pressure; on of microorganisms / decay;	[max 2]

- (a) (i) coal/peat;
 - (ii) reference to: timescale / time to renew; action of, heat/pressure; action of microorganisms / decay;

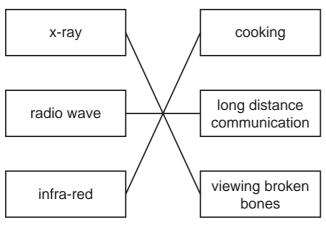
[1]

- **(b) (i)** <u>fractional distillation</u> / <u>fractionation</u>;
 - (ii) too viscous / difficult to ignite; [1]
- (c) (i) 20-22%; [1]
 - (ii) some of it has been used to burn the fuel; [1]
 - (iii) carbon monoxide / nitrogen oxides, produced; [2] toxic to humans;
- (d) $(CH_4 +) 2O_2 \longrightarrow CO_2 + 2H_2O ;;;$ (one mark for each correct formula) [3]

[Total: 12]

- 7 cell; switch; [3] (a) (i) lamp;
 - (ii) correct symbols linked together; [2] in series;

(b) type of wave use of wave



[1]

[Total: 6]

	D^	ao F	Mark Scheme: Teachers' version Syllabus	1
	Pa	ge 5	Mark Scheme: Teachers' version Syllabus IGCSE – May/June 2011 0653	Bar
((a)		population ; community ; consumer ;	DaCambrio.
((b)	(i)	diffusion ; from, alveoli / air sacs ;	[2]
		(ii)	more oxygen can be absorbed from the air/compensates for lack of oxygen; more oxygen, carried by blood/supplied to cells; for respiration/for energy;	[max 2]
((c)	idea	to species diversity; a of their importance in food chain/provide food for pumas/so pumas won't	
			er, e.g. tourism/moral arguments ;	[max 2]
				[Total: 9]
((a)	(i)	beta/gamma, too penetrating; can pass through smoke; alarm would not stop/current would not flow/beta or gamma not ionising enough; beta or gamma would be a hazard to people;	[max 2]
		(ii)	highly ionising; can damage cells / cause mutation / cause cancer / damages DNA;	2]
((b)	(gra	anite) rocks / ground / radon / cosmic radiation ;	[1]
((c)	wea	ar gloves / lead shield / wear radiation badge ;	[1]
·				[Total: 6]
0 ((a)	(i)	group 1 period 2;	[1]
		(ii)	lithium, is (very) reactive/easily combines with other elements; oil forms protective barrier/oil prevents reaction with, air/oxygen/water;	[2]
		(iii)	lithium atoms have two shells / first shell can contain only two electrons; lithium atoms have three electrons; (correctly re-drawn diagram scores 2 marks)	[2]
	((iv)	(yes) lithium is a metal / on left of Periodic Table; (or no) (although a metal it is so reactive) it is covered with corrosion / oxide which acts as an insulator;	[max 1]

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(b) (i) ion is electrically charged and atom is neutral/ion and atom have different numbers of electrons/ion has a full outer shell;

(ii) label line to left electrode;

(iii) chlorine;

[Total: 9]