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## **CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**International General Certificate of Secondary Education** 

## MARK SCHEME for the October/November 2012 series

## 0653 COMBINED SCIENCE

0653/23

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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www.PapaCambridge.com **Mark Scheme Syllabus** Page 2 IGCSE – October/November 2012 0653

(a) (i) haploid; zygote; dissimilar;

(ii) fertilisation;

(b) (i) A - anther/stamen; B - stigma; [2]

(ii) A; **D** ; [2]

(c) (i)

tube	conditions		
С	water	oxygen	no light
D	no water	oxygen	no light
E	water	no oxygen	no light

[2] (all three tubes correct for 2 marks, two tubes correct for 1 mark)

(ii) (lettuce) seeds need oxygen (for germination); (lettuce) seeds need water/moisture (for germination); (lettuce) seeds do not need light (for germination); (max 2 marks if germination **not** mentioned)

[Total: 13]

[max 2]

[3]

- 2 (a) (i) nitrogen 78(%); oxygen 21(%); [2]
  - (ii) nitrogen/an element is in the Periodic Table/nitric oxide/a compound is not; nitrogen/an element only contains one type of atom/nitric oxide/a compound contains more than one type of atom/element; nitrogen/an element cannot be broken down into simpler substances/nitric oxide/a compound can; the atoms in nitric oxide/a compound are bonded together;

[max 2]

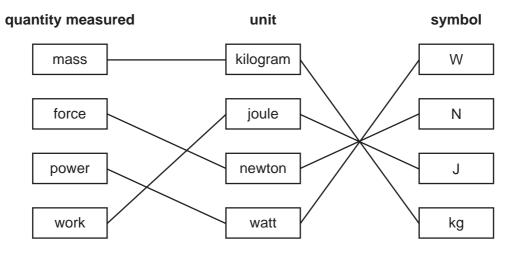
(iii) it represents one molecule of nitric oxide; reference to the bonding of atoms; reference to the 1:1 ratio of N:O;

(iv) oxidation; [1]

(b) (i) ionic/electrovalent; bonding is between metal and non-metal; [2]

Page 3	Mark Scheme	Syllabus	. A.
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	ducts of combustion of magnesium reacted with wat orm an alkaline solution / an alkali ;	er;	[Total: Total: T

3 (a)



(b) A - constant speed/velocity;

**B** - constant acceleration;

[2]

[2]

(c) (distance covered =) speed × time;

 $20 \times 90 = 1800 (m)$ ;

[2]

(d) (i) (resistance =) voltage/current;

$$= 12/2 = 6 (\Omega);$$

[2]

[2]

(ii) (R =) R1 + R2;

$$6 + 6 = 12 (\Omega)$$
;

[Total: 10]

(a) carnivore and consumer;

[1]

**(b) (i)** any number above 20 000;

[1]

(ii) longitudinal;

[1]

(c) (i) idea of reflected randomly/scattered from rough surface/regularly from smooth;

[1]

(ii) idea that it receives ultrasound back from a rough surface (but not from smooth);

[1]

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(d) soil erosion; more carbon dioxide in air/increased greenhouse effect/global warming; extinctions/loss of habitat; 5 (a) chlorination; kills (harmful) microorganisms; AND/OR filtration; removes solids; AND/OR evaporation; removes all impurities/removes water from impurities; [max 4] **(b) (i)** red; dye giving only one spot matches red in P/owtte; [2] (ii) S; [1] [Total: 7] (a) heat; kinetic; [2] **(b) (i)** water turns to a gas/(water) vapour; (as) particles/molecules get further apart; (more) energetic particles escape; heat is needed/used to cause evaporation; [max 2] (ii) heat is needed/used to cause evaporation; (more) energetic particles escape; remaining (particles) have less (thermal) energy; [max 1] (c) in solid:

particles in regular arrangement; particles all touching; in liquid:

particles arranged irregularly;

particles touching; [max 2]

(d) (efficiency =) useful energy out ÷ energy in ;

idea of how little or how much energy is wasted by machine/device; [max 1]

[Total: 8]

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7		B - i) cru bre inc	- incisor/canine; - molar/premolar; sh/grind; ak into small pieces rease surface area a of better access f	of food;	er to digest ;		[max 2]
	(b)		part	ingestion	digestion	absorption	
			mouth	✓	<b>✓</b>		
			stomach		<b>✓</b>		
			small intestine		✓	✓	
	(1	1 mark	per correct row) ;;;				[3] <b>[Total: 7]</b>
8	(a) (i	•	ctile ; ectrical) conductor ;				[2]
	(ii		cture of metals ; by is less malleable.	/hard <u>er</u> /stron <u>ge</u>	<u>r</u> /low <u>er</u> melting រុ	point;	[2]
	(iii	i) cop	oper sulfide + oxyge	en ——► coppe	r + sulfur dioxide	;	[1]
	(b) (i		oper chloride solutic	on/the conductin	g solution ;		[1]
	(ii		orine ; obles/gas given off	;			

copper; reference to copper coloured/brown/pink layer/solid;

[Total: 10]

[4]

Page 6	Mark	Scheme	Syllabus	W. P.
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(a)				ambridge
	component	symbol		E.
	ammeter	—(A)		S.COM

(a)

component	symbol
ammeter	<b>A</b> —
fuse	
variable resistor	<del></del>

**(b) (i)** 3; [1]

[1] (ii) correct symbol in parallel with bulb;

(c) (i) angle of incidence and angle of reflection; [1]

(ii) 45°; [1]

[Total: 7]

[3]