CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the May/June 2013 series

0653 COMBINED SCIENCE

0653/32

Paper 3 (Extended Theory), maximum raw mark 80

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This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2013 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.

Page 2	Mark Scheme	Syllabus	.03
	IGCSE – May/June 2013	0653	735

- 1 (a) (i) Group 1 elements all metals and Group 7 elements all non-metals; Group 4 elements non-metals at top and metals at bottom/contain both types of element;
 - (ii) melting point decreases down Group 1 so francium has lowest m. pt; melting point increases down Group 7 so astatine has highest m. pt;
- [2]

(b) (i) cobalt chloride paper; changes from blue to pink;

OR

anhydrous copper sulphate; changes from white to blue;

[2]

(ii) PbO + $H_2 \rightarrow Pb + H_2O;$

[2]

(iii) calcium has a high reactivity; calcium reactivity greater than hydrogen; calcium too strongly bonded to oxygen;

[max 2]

[Total: 10]

2 (a) (i) $W = F \times D$; = 1400 × 10 = 14 000 J;

[2]

(ii) KE = $\frac{1}{2}$ mv²; = $\frac{1}{2}$ × 5000 × 1.5 × 1.5 = 5625 J;

[2]

(b) density = mass / volume; = $5000 / 5 = 1000 \text{ kg/m}^3$;

[2]

(c) (i) below 20 Hz; human threshold is about 20 Hz;

[2] [1]

(ii) number of vibrations per second;

[1]

(iii) sound waves with frequencies above human threshold/above 20000 Hz;

[Total: 10]

	Page 3		}	Mark Scheme	Syllabus		
				IGCSE – May/June 2013	0653	130	
3	(a)	(i)	geot	ropism/gravitropism;		Papa Cambridge	
		(ii)		ers held up;		8	
				re insects can reach them; ollination;		[2]	
				,			
	(b)			rface has grown more than upper surface; es from 1 st graph;			
		aux	in coi	ncentrates on lower surface;			
				gures from 2 nd graph/deduction that auxin has moved	d away from uppe	er surface;	
		mo	re aux	kin causes more growth;		[max 3]	
						[Total: 6]	
4	(a)	(i)	therr			ro1	
			tnerr	mal and conduction;		[2]	
		(ii)	com	munication;		[1]	
	(b)		•	ver = 1.8 kW; 1800 × 30 × 60;			
				000 J;		[3]	
	(c)			y could be produced by burning fossil fuels;			
				iels release CO ₂ when burned; demand for (fossil) fuels/electricity reduces amount	of CO ₂ released;	[max 2]	
						_	

[Total: 8]

Page 4	Mark Scheme	Syllabus	.0	V
<u>-</u>	IGCSE – May/June 2013	0653	100	

5 (a) sodium atom loses an electron/outer shell; oxygen atom gains two electrons/fills outer shell; idea that two electrons provided by two sodium atoms; reference to ions formed; attraction between positive and negative ions;

[max 3]

(b) (i) glowing splint relights;

[1]

(ii) $24 \div 400 = 0.06$; $20 \text{ cm}^3 \text{ per second}$;

[2]

(iii) hydrogen ions gain electrons;(each ion gains) one electron;atoms join in pairs to form hydrogen molecules;

[max 2]

(iv) the higher the current the higher the rate of production of hydrogen; current is (rate of) flow of electrons/charge; so if electrons arriving at cathode (per second) is higher then more H⁺ ions discharging in given time;

[max 2]

[Total: 10]

		The state of the s					
	Pa	ge 5	Mark Scheme	Syllabus			
			IGCSE – May/June 2013	0653			
6	(a)	chang light e water carbol	synthesis; es light energy to chemical energy; nergy absorbed by chlorophyll; combined with carbon dioxide; nydrates produced; nydrates contain chemical energy;	Syllabus 0653 Rahat Ahhhhitata [max 4]			
	(b)	or not all e.g. shidea the or all so sor	ation; / lost as heat; organisms eaten/not all parts of organisms eaten; neep does not eat grass roots/human does not eat sh nat this energy goes into decomposer food chain; food digested; ne not absorbed into organism's body/some lost in fa nat this energy goes into decomposer food chain;				
	(c)	al	$_{6}H_{12}O_{6} + 6O_{2} \rightarrow 6CO_{2} + 6H_{2}O$ I formulae correct; alanced;	[2]			
		fro so by	ore heat lost in cold environment; om skin/by radiation/by conduction; o more heat needs to be produced within the body/in o respiration; sing food/glucose/carbohydrates (as fuel);	cells;			
			ating more increases fat deposits in the body; r heat insulation;	[max 3] [Total: 11]			
				• •			
7	(a)		t series circuit; nbols correct;	[2]			
	(b)	renew	newable – coal/oil/gas/nuclear; able – geothermal/wave/tidal/hydroelectric; required for mark)	[1]			

(c) (i) $R = R_1 + R_2$; $1200 + 2400 = 3600 \Omega$; [2]

(ii) 800Ω ;

combined resistance of parallel components is less than that of either resistance; ${\bf OR}$ calculation $[{\sf max}\ 2]$

[Total: 7]

Page 6	Mark Scheme	Syllabus
	IGCSE – May/June 2013	0653

8 (a) T; PQR; R(S); P;

> (b) (i) decreases slowly (at the start); sudden steep fall; decreases slowly (after the sudden fall);

[max 2]

(ii) these are the volumes at pH 7/owtte;

[1]

(iii) (A)

lower volume of **A** needed to neutralise the alkali;

[1]

[Total: 8]