

CAMBRIDGE INTERNATIONAL EXAMINATIONS

Cambridge International General Certificate of Secondary Education

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MARK SCHEME for the October/November 2014 series

0653 COMBINED SCIENCE

0653/21

Paper 2 (Core Theory), maximum raw mark 80

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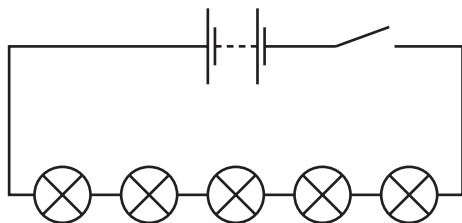
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1 (a)



circuit symbols correct ;
 circuit connected correctly with 5 lamps (allow ± 1) ; [2]

(b) (i) gives p.d. required to light the lamp (for normal use)/what battery to use to make lamp work/to drive correct current ;
 don't exceed stated p.d. or may blow lamp/too high current (owtte) ; [2]

(ii) $R = 1.2/0.1 = 12 (\Omega)$; [1]

(iii) Total resistance = $5 \times 12 = 60 (\Omega)$; *allow ecf* [1]

(iv) ohm ; [1]

[Total: 7]

2 (a) B C ;
 (B C) D A ; [2]

(b)

material	description
element	can be found in the Periodic Table
element	cannot be broken down into simpler substances
mixture	contains different types of molecules
compound	only contains molecules which are identical but each molecule contains more than one type of atom

;;;; [4]

(c) (i) fuels ; [1]

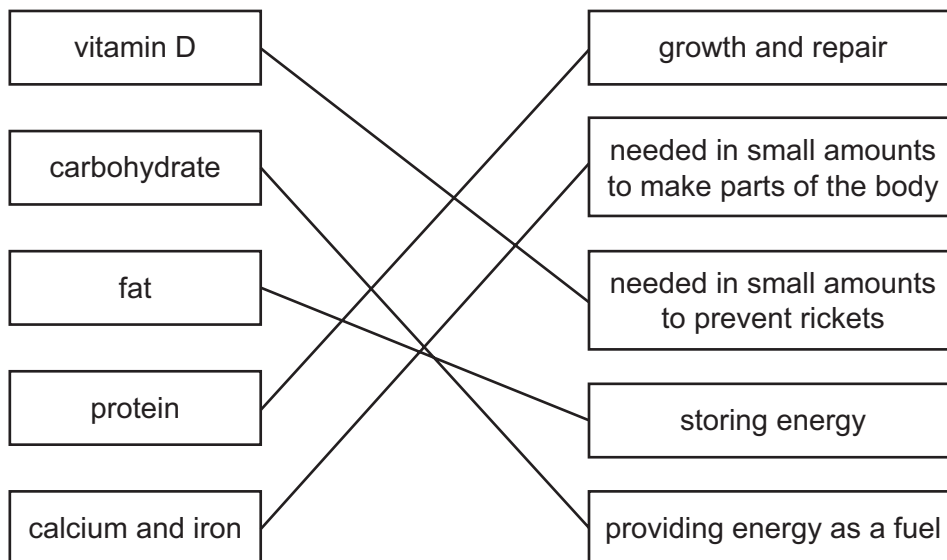
(ii) flammable ; [1]

(d) (i) carbon dioxide ;
 water ; [2]

(ii) C_3H_8 ; [1]

[Total: 11]

3 (a) (i)



(all 4 correct = 3 marks, 3 correct = 2 marks, 2 or 1 correct = 1 mark) [3]

(ii) prevents constipation ; [1]

(b) (i) meal 2 is balanced / meal 1 is not balanced / not enough vitamin C/D / not enough fibre / not enough iron / calcium ;
meal 1 contains too much fat / carbohydrate / energy ;
Jill does not need so much energy in food because of her lifestyle ; [max 2]

(ii) e.g. apple (no mark)
adds fibre / vitamins C/D ;
or
milk (no mark)
adds calcium / vitamins C/D ;
or
any food (no mark) and valid point that addresses the reasons mentioned in (b)(i) ; [1]

[Total: 7]

4 (a) (i) initial any between 8 and 14
final 7 ; [1]

(ii) initial violet / purple
final green ; [1]

(iii) neither acid nor alkali in excess ;
(chemical) reaction / neutralisation ;
Ref. to acid has pH < 7 / low pH ; [max 2]

(b) (i) repeat without indicator ;
using same volumes of solutions ; [2]

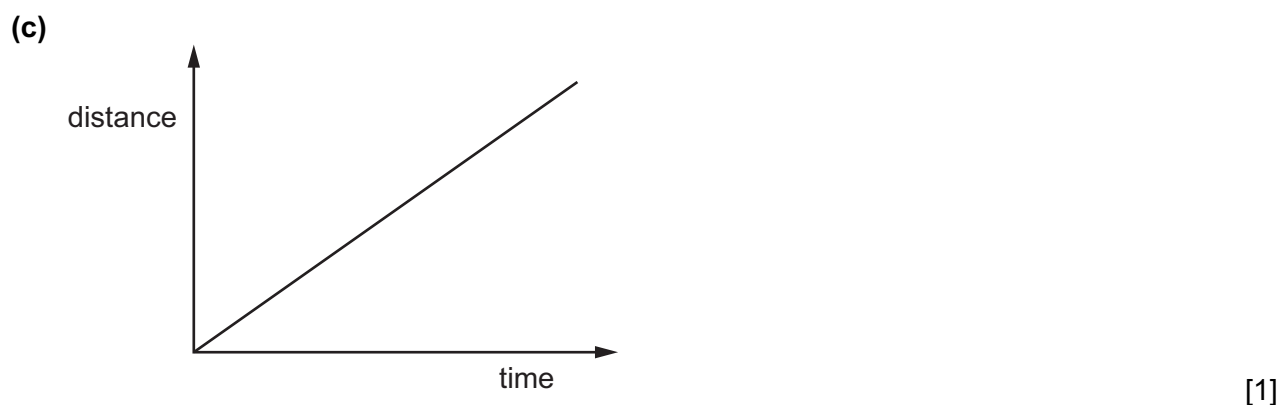
(ii) evaporation / heating to remove water ; [1]

[Total: 7]

Page 4	Mark Scheme	Syllabus Paper
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- 5 (a) (i) **A** cell membrane ;
B vacuole ;
- (ii) **X** label line to green area and **Y** label line to white area ;
green color contained in chloroplasts ;
which are only present in cell X ;
or
white area does not contain chlorophyll/chloroplasts ;
which are absent from cell Y ; [3]
- (iii) black/blue black or shaded area matching green area of leaf and labelled black ; [1]
- (iv) chlorophyll/chloroplasts needed to make starch/for photosynthesis/to trap light ; [1]
- (b) (i) enzyme is denatured ;
not the optimum pH for the enzyme ; [2]
- (ii) pancreas/salivary glands/small intestine/stomach ; [1]
- [Total: 10]

- 6 (a) (i) weight of the aircraft: **R** ;
thrust of the engines: **Q** and force of the brakes **S** (both needed, correct way round) ; [2]
- (ii) thrust and braking force equal and opposite ;
unbalanced forces needed to move/accelerate (the aircraft) ; [2]
- (b) (i) chemical (energy) [1]
- (ii) kinetic (energy) ;
gravitational/potential (energy) ; [2]
- (iii) heat/sound ; [1]



[Total: 9]

Page 5	Mark Scheme	Syllabus Paper
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- 7 (a) (i) returns carbon/carbon dioxide back to air/produces carbon dioxide from sugar/glucose ;
- (ii) sugar/glucose + oxygen → carbon dioxide + water
LHS ; RHS ;
- (iii) grass → rabbit → fox/grass → rabbit → decomposer ;; [2]
(1 mark for correct organisms in right order, 1 mark for correct arrows)
- (iv) feed on dead/decaying bodies/waste ;
release carbon/carbon dioxide back to the air ;
during respiration ; [max 2]
- (b) (i) combustion/burning of fossil fuels (produces carbon dioxide) ;
reduces carbon dioxide removal by photosynthesis; [2]
- (ii) (increases) global warming/description of consequences ; [1]
- [Total: 10]**

- 8 (a) infra-red ;
evaporates ;
molecules ; [3]
- (b) mercury absorbs energy (from Sun)/mercury expands as temperature increases ;
mercury particles move faster/mercury particles get further apart ; [2]

(c)

higher frequency			lower frequency			
gamma radiation	X-rays	ultraviolet	visible	infra-red	microwave	radio waves

any one in its correct box ;
second one in correct box ;
third and fourth in correct boxes ; [3]

- (d) sound needs medium to travel through/can't travel through a vacuum/empty space ;
region between Sun and Earth is a vacuum ; [2]

[Total: 10]

Page 6	Mark Scheme	Syllabus
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- 9 (a) (i) copper oxide + carbon → copper + carbon dioxide
LHS ; RHS ;
- (ii) copper oxide (loses oxygen so) is reduced ;
carbon (gains oxygen so) is oxidised ; [2]
- (b) no reaction (between copper and dilute hydrochloric acid) ;
copper is below hydrogen in the reactivity series ; [2]
- (c) (sulfur dioxide) dissolves in rain / atmospheric water / forms acid rain ;
collects in lakes / rivers etc. ;
damages plants / animals ;
effect of gas on respiratory system ;
other correct ; [max 3]

[Total: 9]