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0653 COMBINED SCIENCE

0653/31

Paper 3 (Extended Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

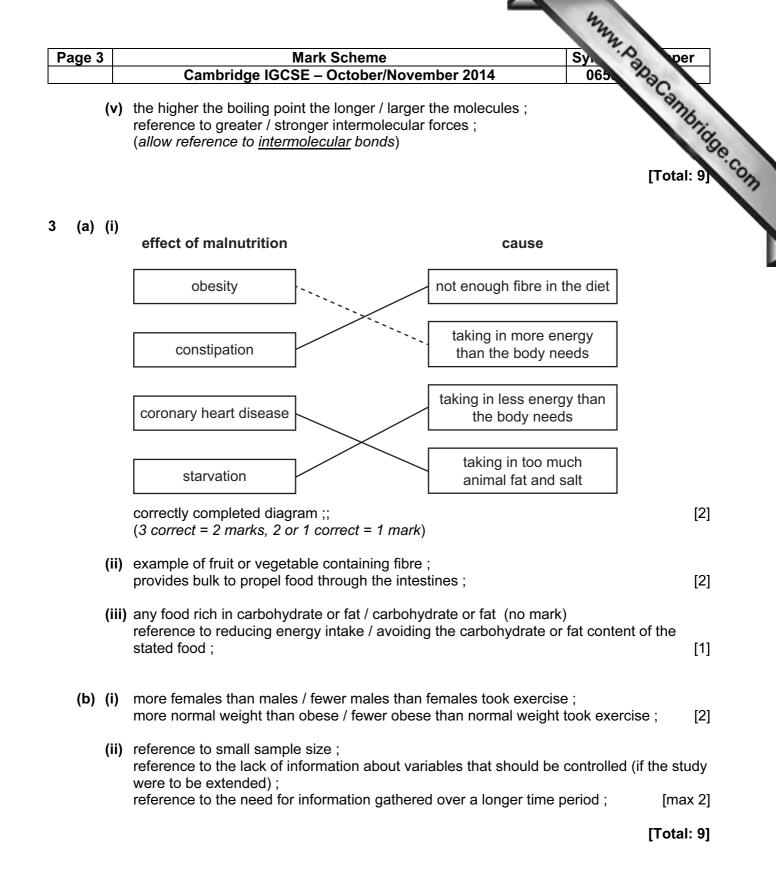
Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

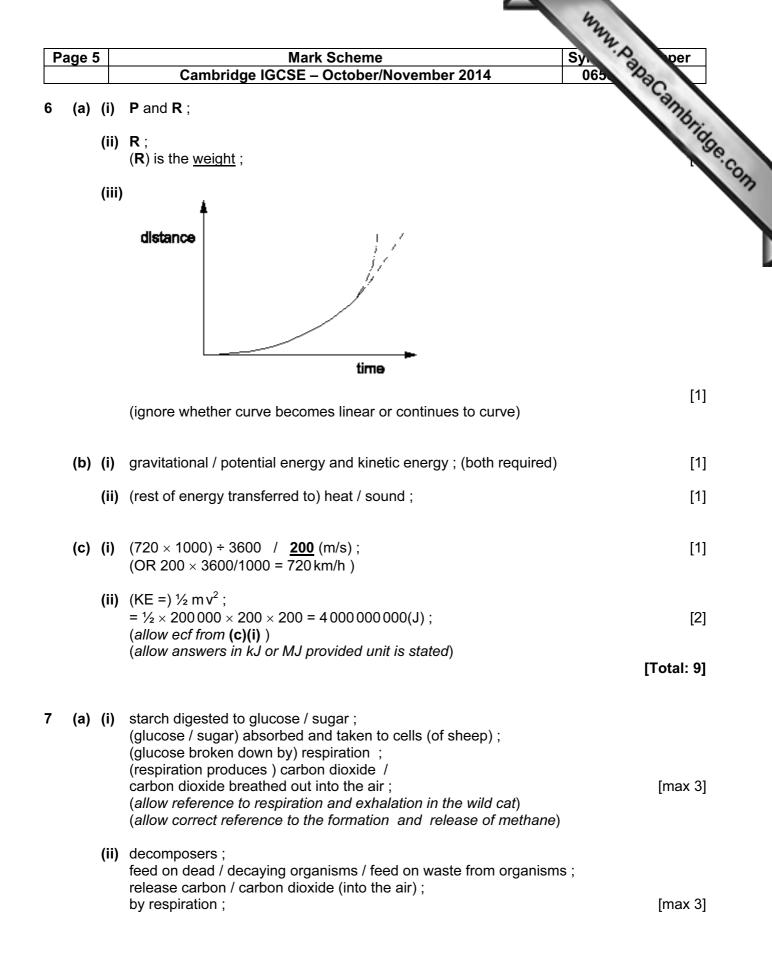
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			www.
P	age 2	2 Mark Scheme	Sv. P. per
		Cambridge IGCSE – October/November 2014	Syn ⁷⁷ , Dath per 065, Phar per
1	(a)		Sy. 065 per 065 per 069 per 069 per 069 per 09 per 00 per
		symbols all correct ;	
		circuit connected correctly (allow ±1 cell or lamp);	[2]
	(b)	(i) $5 \times 0.5 = 2.5$ (A);	[1]
		 (ii) (R =) V/I (or words); = 6 / 2.5 = <u>2.4</u> (Ω); 	[2]
	(c)	series: all bulbs go out AND parallel: rest of bulbs stay alight ;	[1]
			[Total: 6]
2	(a)	BC; (BC)DA; (allow 1 mark if both B and A are correctly located)	[2]
	(b)	(i) catalyst ;	[1]
		 (ii) increases rate / frequency of collision of particles ; increases speed of reaction / increases surface area (of catalyst) ; 	[2]
		 (iii) (petroleum) jelly (diesel) oil (refinery) gas in order ; (iv) (petroleum) jelly (diesel) oil (refinery) gas 	[1]
		(refinery) gas in order ;	[1]



_	1	Mark Scheme Syn	per
		Cambridge IGCSE – October/November 2014 065	000
(a)	(i)	initial between 8 and 14 to 7 (final);	ambr
	(ii)	purple / blue to green ;	19
(b)	(i)	KCl;	pacambrio
		H ₂ O ;	[2]
	(ii)	repeat without indicator / use pH meter / use indicator paper ;	
		using same volume(s) of solution(s) ; evaporate (the water from the neutral mixture) / heat (the solution) then cool ;	[3]
(c)	the	erence to the involvement of ions / ionic compound / particles with opposite char idea of strong forces / bonds between particles that must be broken / s must be separated ;	rges ;
		eaking bonds / separating ions) requires a large amount of energy ;	[max 2]
			[Total: 9]
(a)		abel line to green area and Y label line to white area ;	
	-	en area containing chlorophyll / chloroplasts only in cell X / te area does not contain chlorophyll /chloroplasts shown in cell Y ;	[2]
(b)	(i)	black or shaded in area matching green area of leaf and indicated as black ;	[1]
	(ii)	chlorophyll / chloroplasts traps <u>light</u> energy ;	
		for photosynthesis ; which makes (glucose / sugar which leads to) starch ;	[3]
		d <u>denatures</u> enzyme ;	
(C)		longer optimum pH / owtte ;	
(C)		inges shape of enzyme / active site / substrate no longer fits active site ;	[3]



Page 6	Mark Scheme Sy. A	per
	Cambridge IGCSE – October/November 2014 065	30
(b) (i)	increases level of carbon dioxide / carbon monoxide ; reduces oxygen level ; increases sulfur dioxide level ;	per acambridge.co.
(ii)	<i>carbon dioxide:</i> (increases) global warming / described consequence e.g. changed rainfall patterns / floods and or droughts ;	.00.
	<i>sulfur dioxide:</i> causes acid rain / described consequence e.g. chemical weathering of structures / damage to trees or aquatic organisms reference to harmful effects in relation to breathing ;	
	(allow other valid answers)	[max 1]
		[Total: 9]

8 (a) (i) number of vibrations / waves per second / unit of time ;

[1]

(ii)

	(11)							
		highest freq	uency				lowes	t frequency
		(gamma radiation)	X-rays	ultra- violet	(visible light)	infra-red	(microwaves)	(radio waves)
		all three corre and in correc (<i>allow 1 mark</i>	t positions	•	amed and loc	ated)		[2]
(b)	(i)	move further decrease / we quicker / mor	eaken / ge	t less ;		them ;		[3]
	(ii)	infra-red radia (energy from which move f forces betwee (molecules) e	sun) absc faster / gai en molecu	rbed by wat n kinetic en les are wea	ter (molecule: ergy. ; kened / broke	en;	as / vapour ;	[max 2]
(c)	(i)	sound is a loi sound needs space is a va	medium t	o travel thro	bugh ;			[max 2]
	(ii)	8 minutes / the electromagne		-			ecause all ace / vacuum ;	[1]
								[Total: 11]

Page 7		Mark Scheme Sy.	Der per
		Cambridge IGCSE – October/November 2014 06	1030
(a)	(i)	exothermic ;	annb.
	(ii)	chemical (potential) \rightarrow thermal / heat / kinetic ;	W. Papacambridg
	(iii)	aluminium (gains oxygen and) is oxidised ;	
		iron (oxide) (loses oxygen and) is reduced ;	[2]
		(allow correct references to electron gain by iron and electron loss from a	luminium)
	(iv)	iron will not react with / reduce aluminium oxide ;	
		iron is lower in the reactivity series / less reactive than aluminium ;	[2]
(b)	(i)	cations / aluminium ions migrate / move / are attracted to the cathode	
		/ negative electrode ;	
		electrons flow on to ions / ions gain electrons ;	
		the idea that the ions are discharged as the result of electron gain ;	[max 2]
	(ii)	oxygen ;	[1]
			[Total: 9]