

## MARK SCHEME for the May/June 2015 series

## **0653 COMBINED SCIENCE**

0653/63

Paper 6 (Alternative to Practical), maximum raw mark 60

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Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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Page 2	Mark Scheme
	Cambridge IGCSE – May/June 2015

- 1 (a) yeast dead/(enzyme) no longer active/denatured ;
  - (b)

				Mary Mary
2		Mark Scheme		Syl Syl per
	Ca	mbridge IGCSE – May/	June 2015	065 23
) y )	east dead/(enzym	e) no longer active/dena	tured ;	Sy. And oer 065 And Cambridge colour in tube C
	time/mins	colour in tube A	colour in tube <b>B</b>	colour in tube C
	1	blue	blue	blue
	2	colourless	blue	blue
	3	colourless	blue	blue
	4	colourless	blue	blue
	5	colourless	colourless	blue
	6	colourless	colourless	blue

time/mins;

- A correct ; B correct ;
- C correct ;

2

ALLOW decolourised IGNORE transparent	[4]
(c) (i) constant volume/concentration;	[1]
<ul> <li>(ii) A changes quicker/changes first/respires faster ;</li> <li>(more) glucose/substrate available in A ;</li> </ul>	[2]
M2 dependent on times being considered	
<ul> <li>(colour changes back to) blue ; methylene blue oxidised/reacts with oxygen/oxygen introduced ; oxygen from air above solution ;</li> </ul>	[max2] <b>[Total: 10]</b>
<ul> <li>(a) make a solution in water ;</li> <li>add (aqueous) sodium hydroxide/(aqueous) ammonia ;</li> <li>green (gelatinous) ppt/solid ;</li> </ul>	[3]
<ul> <li>(b) add sodium hydroxide (solution) and heat ;</li> <li>damp ;</li> <li>(red) litmus turns blue ;</li> </ul>	[3]
<ul> <li>(c) make a solution in water ; add hydrochloric/nitric acid ; add barium chloride/nitrate (solution) ; white ppt ;</li> </ul>	[4] [Total: 10]

	e 3	Mark Scheme Syn	ber ber
		Cambridge IGCSE – May/June 2015 065	1000
(a		rect symbol for voltmeter ; inected in parallel between ${\bf X}$ and ${\bf Y}$ or equivalent ;	W. Papacambridg [2]
(k	b) (i)	values in table: 1.81; ALLOW range 1.80 – 1.82 0.7 <b>0</b> ;	[2]
	(ii)	headings: V, A, $\Omega$ (all three required) ;	[1]
	(iii)	3.91, 8.00, 2.59 (allow ecf on third value) all values to 2 d.p ; all correct values ;	[2]
(0	stat AN	e of 3.91 and 2.59 ; tement matches results (expect NO) D tification: e.g. values are too different/not close enough, even allowing	
		experimental error/is 1.5 times ;	[2]
(c		lamps are at different temperatures/lamps have different resistances or cunnersected/this could explain why teacher statement not supported ;	urrents [1]
			[Total: 10]
(a	a) (i)	61 ;	[1]
	(ii)	433 ;	[1]
	(iii)	0.0023 ;	[1]
(k	b) (i)	Correct plotting (allow 1 error) ; SMOOTH curve ;	[2]
	(ii)	52 ± 2 ;	[1]
	(iii)	Do not know the rate either side of $52^{\circ}C$ / need more results in range e.g. $40^{\circ}C$ to $60^{\circ}C$ ;	[1]
(0	c) rep	eat experiment with water instead of acid ;	
	1 cr	m <sup>3</sup> ;	
		ution will remain cloudy ;	[3]

Page 4	•	Mark Scheme Syn	per
		Cambridge IGCSE – May/June 2015 065	20
(a)	(i)	lamp/bulb/ammeter;	andri
	(ii)	correct symbol for cell (or battery);	19
(	(iii)	(explanation) does not react ; (material) e.g. carbon/platinum ;	na Cambridge [2]
(b)	(i)	gives red-brown ppt ;	[1]
	(ii)	<u>damp</u> litmus ; (red then) bleached ;	[2]
(	(iii)	hydrogen ; lit splint ; "pops" ;	[3]
			[Total: 10]
(a)	(i)	21.5 ; 20.5 ;	[2]
	(ii)	axes correct and labelled ; vertical axis NOT starting at zero ; points correct (allow 1 error) ; (e.c.f. from part <b>(i)</b> )	[3]
(	(iii)	no, points scattered/no pattern/no straight line ; (e.c.f. from parts (i) and (ii)) (ignore any line drawn)	[1]
(b)	rod am	y <b>three</b> of) s should be same length and width ; ount of wax should be the same ; eriment repeated and average taken ;	
		er should be stirred ;	[3]
(c)		swer depends upon <b>(b)</b> ) p thickness / length (etc.) means only variable is % magnesium ;	
		eating identifies anomalous results ;	[1]
	•		