MARK SCHEME for the October/November 2015 series

0653 COMBINED SCIENCE

0653/33

Paper 3 (Extended Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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Page 2		2	Mark Scheme Syllabu			
	•		C	Cambridge IGCSE – October/November 2015	0653	Paper 33
1	(a)	larg into cap				[3]
	(b)			y diameter ; nucus obstructing flow ;		[2]
	(c)	(i)	$\frac{\frac{6.7}{5.8} \times 10}{\frac{12.5}{5.8}} = 2$			
			215 – 10	0 = 115 (%) ;		[max 2]
		(ii)	to get mo for <u>respir</u>	pre oxygen (into the blood)/remove more carbon dioxide ; ration ;		[2]
	(d)	mu airv bac or (mo airv mo	cus canno vays beco teria/tar v ore mucus vay becon re mucus	<i>paralysed)</i> In the shifted upwards ; me even more restricted/more mucus for bacteria to bread will not be removed from the lungs/increases risk of bronce <i>is produced)</i> mes even more restricted/blocked ; for bacteria to breed in/remain in lungs/increases chance ns/reduces oxygen supply for the body ;	chitis ; es of	[max 2] [Total: 11]
2	(a)	the	rmal enerç	gy to chemical energy ;		[1]
	(b)	(i)	steeper g	gradient than solid line ;		[1]
		(ii)		g concentration increases rate of reaction ; d frequency of collisions ;		[2]
	(c)	(i)	atoms ions	ions atoms ; <i>(must be in this order)</i>		[1]
		(ii)	silver wri	tten below copper ;		[1]
		(iii)		s of more reactive metals become ions more readily than tive metals ;	those of	[1]
						[Total: 7]

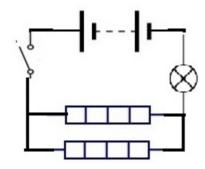
Pa	age 3	3	Mark Scheme	Syllabus	Paper
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3	(a)	we	ight/gravitational force (accept gravity) ;		[1]
	(b)	(gra	avitational) potential (energy) to kinetic (energy) ;		[1]
	(c)	(i)	accelerating ; constant speed ;		[2]
		(ii)	area under graph between A and C or $(\frac{1}{2} \times 2 \times 9) + (2 \times 9)$ or $\frac{1}{2}(2 + 2) = 27$ (m);	⊦4) × 9 ;	[2]
	(d)	particles far apart in air/gas, but close together/touching in water/liquid (accept diagrammatic description); stopping the skateboarder requires loss of KE; more work done/loses more KE/more difficult to push water particles aside than pushing air particles aside/owtte/diagrams; harder to push water molecules apart because of the forces between them/more particles;		aside	[max 3]
					[Total: 9]
4	(a)	(i)	contains the correct proportions of nutrients for an individual ;		[1]
		(ii)	to prevent scurvy/AVP ;		[1]
	(b)	(i)	as temperature increases the amount of vitamin C decreases ;		[1]
		(ii)	rate/amount of decomposition/breakdown/disappearance increas temperature ;	ses with	[1]
		(iii)	temperature will vary in different parts of the world ; this will affect the amount of vitamin C (in fruit before the experime OR amount of water given/contained in1 fruit may vary ; this could affect concentration of the fruit juice ; OR different variety of orange/fruits vary genetically ;	nt) ;	
			(naturally) contains different amounts of vitamin C ; AVPs ;;		[max 2]
	(c)	boi	ling water destroy some/all of the <u>vitamin C</u> ;		[1]
			nvenient if you need to leave the baby / mother may not have enoug	h milk/AVP	;
			sadvantage) es not contain antibodies/any reference to bonding/AVP ;		[2]
					[Total: 9]

Page 4		Mark Scheme Syllabus			Paper
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5	(a) [(b) (i	A 			[1]
		observation	explanation		
		(bubbles of gas)	hydrogen ;		
		(changes from green to purple)	alkaline solution ;		
					[2]
	(ii	more vigorous reaction/hydroge alkali metals become more reac (accept correct references to ea		erature rise ;	[2]
	(iii	one electron in shell ; all Group I elements have 1 elec	ctron in outer shell ;		[2]
					[Total: 7]
6	(a) (i	(<mark>10</mark> =) 0.5 ; Hz/hertz ;			[2]
	(ii	$(\frac{30}{10}) = 3 (m/s)$			[1]
	(iii	$v = f\lambda$ (in any form)/ $\frac{3}{0.5}$;			
		= 6 (m) ; (allow ecf from (i) and/or (ii))			[2]
	(b) 2	× amplitude (0.5) = 1 (m) ;			[1]
	(c) R	placed in right-hand end box ;			[1]
	(d) (i		earer to land so easier access/in sed on access to the energy invo		

Page 5		5	Mark Scheme	Syllabus	Paper
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		(ii)	$(efficiency = \frac{useful \ energy \ output}{energy \ input})$ $= \frac{150}{500} \times 100 = 30(\%);$		[1] [Total: 9]
7	(a)	cor (gr froi infr	ows solar radiation to reach the Earth's surface ; ntains greenhouse gas molecules/example ; eenhouse gases in the atmosphere) absorb infra-red radiation (that i m the Earth's surface) ; ra-red radiation is then (re-)emitted by the greenhouse gases into the ck to the Earth's surface ;		e/ [max 2]
	(b)		o from: bon dioxide/methane/water vapour ;		[1]
	(c)	(ind (ind	o from: creased) use of fossil fuels/example; creased) deforestation ; creased) keeping of cows/growing rice ; P ;		[max 2]
	(d)	use rec rec pla	luced use of fossil fuels/removal of carbon dioxide from exhaust/pro e of public transport ; luced deforestation ; luced agricultural practices that cause methane to be produced ; nt more trees ; e more renewable energy sources ; P ;	omote the	[max 1] [Total: 6]
8	(a)	(i)	$2HCl + (CuCO_3) \rightarrow (CuCl_2) + CO_2 + H_2O$ formulae ; balanced ;		[2]
		(ii)	limewater ; turns milky/cloudy etc. ;		[2]
	(b)	(i)	copper (deposit on cathode) ; <u>chlorine</u> (gas at anode) ;		[2]
		(ii)	(copper ions): move towards cathode/negative (electrode) ; (chloride ions): move towards anode/positive (electrode) ;		[2]

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(c))	(i)	CuC <i>l</i> ; reference to the need for charge balance ;		[2]
	(ii) high density/high melting point/form coloured compounds/act as				[max 1]
(d	(d) correct structure of methane molecule ; correct structure of ethane molecule ;				[2] [Total: 13]
9 (a))	(i)	move towards each other ; unlike charges attract ;		[2]
(b))	(i)	force ;		[1]
		(ii)	any path heading towards the upper positive plate ;		[1]

 (c) (i) complete circuit with 2 extra components included in series and/or in parallel; two heaters in parallel; lamp in series in main circuit;



[3]

(ii) water expands/volume increases/particles get further apart water becomes less dense;
(less dense)/warm water rises (above denser colder water) / owtte;
[2]

[Total: 9]