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UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the NOVEMBER 2004 question paper

0654 CO-ORDINATED SCIENCES

0654/02 Pa

Paper 2 Core (Theory), maximum raw mark 100

Mark schemes must be read in conjunction with the question papers and the *Report on the Examination*.

• CIE will not enter into discussion or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the November 2004 question papers for most IGCSE and GCE Advanced Level syllabuses.

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Grade thresholds taken for Syllabus 0654 (Co-ordinated Sciences) in the November 2004 examination.

	maximum	minimum mark required for grade:				
	mark available	А	С	Е	F	
Component 2	100	n/a	48	36	25	

The threshold (minimum mark) for B is set halfway between those for Grades A and C. The threshold (minimum mark) for D is set halfway between those for Grades C and E. The threshold (minimum mark) for G is set as many marks below the F threshold as the E threshold is above it.

Grade A* does not exist at the level of an individual component.

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November 2004

INTERNATIONAL GCSE

MARK SCHEME

MAXIMUM MARK: 100

SYLLABUS/COMPONENT: 0654/02

CO-ORDINATED SCIENCES
Paper 1 Core (Theory)

www.PapaCambridge.com Page 1 Mark Scheme Syllabu **IGCSE - NOVEMBER 2004** 0654

- 1 (a) tissue
 - label line to cell wall or vacuole and name (b)
 - (c) chloroplasts; which contain chlorophyll; (chlorophyll) absorbs sunlight

(d) near the (upper) surface of the leaf; only one layer/epidermis above them; epidermis cells have no chloroplasts; cells are arranged upright/vertically; so light does not have to pass through several cell walls;

max [2]

(e) down

> the plant is photosynthesizing; faster than it is respiring; using carbon dioxide (from the air)

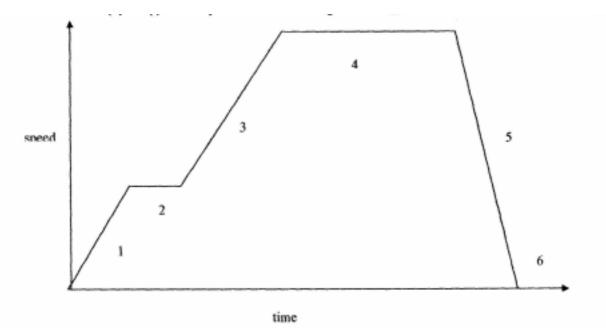
ир

the plant is respiring releasing carbon dioxide (into the air)

[max 3]

Total 9

2 (a) (i) six points to look for see grid below;;; [3]



(ii) speed = distance/time etc;

speed =100(km/hr);

[2]

Page 2	Mark Scheme	Syllabu	 3,	1
	IGCSE – NOVEMBER 2004	0654	0	

 (a) C; A; A; (b) (i) petroleum/crude oil; (ii) named primary product from fractional distillation (c) (i) barrier; prevents air/oxygen and water; 	da
 (ii) there is still a complete circuit for the other bulbs (iii) 13(A) (a) C; A; A; (b) (i) petroleum/crude oil; (ii) named primary product from fractional distillation (c) (i) barrier; prevents air/oxygen and water; 	00
 (ii) there is still a complete circuit for the other bulbs (iii) 13(A) (a) C; A; A; (b) (i) petroleum/crude oil; (ii) named primary product from fractional distillation (c) (i) barrier; prevents air/oxygen and water; 	dine
 (ii) there is still a complete circuit for the other bulbs (iii) 13(A) (a) C; A; A; (b) (i) petroleum/crude oil; (ii) named primary product from fractional distillation (c) (i) barrier; prevents air/oxygen and water; 	[2]
 (a) C; A; A; (b) (i) petroleum/crude oil; (ii) named primary product from fractional distillation (c) (i) barrier; prevents air/oxygen and water; 	[1]
 (a) C; A; A; (b) (i) petroleum/crude oil; (ii) named primary product from fractional distillation (c) (i) barrier; prevents air/oxygen and water; 	[1]
A; A; (b) (i) petroleum/crude oil; (ii) named primary product from fractional distillation (c) (i) barrier; prevents air/oxygen and water;	otal 11
A; (b) (i) petroleum/crude oil; (ii) named primary product from fractional distillation (c) (i) barrier; prevents air/oxygen and water;	
 (b) (i) petroleum/crude oil; (ii) named primary product from fractional distillation (c) (i) barrier; prevents air/oxygen and water; 	
 (ii) named primary product from fractional distillation (c) (i) barrier; prevents air/oxygen and water; 	[3]
(c) (i) barrier; prevents air/oxygen and water;	[1]
prevents air/oxygen and water;	[1]
	nax [2]
(ii) reference to oil as a barrier (to air and water);	[1]
(d) a chemical reaction occurs in the battery;	
(the reaction) provides electricity; reacting chemicals are used up/not a reversible reaction; n	nax [2]
To	otal 10
(a) A lens; B vitreous humour; C choroids layer;	[3]
(b) (i) label line F to retina;	[1]
(ii) lable line P to iris;	[1]
(c) as electrical signal/electrical impulse/action potential; along a neurone; in the optic nerve;	nax [2]
(d) (i) different/longer wavelength;	[1]
(ii) they are warmer (than their surroundings); they regulate their body temperature/they are homeothermic; heat generated by metabolic reactions/respiration/muscle activity; n	nax [2]
To	

	Page 3		Mark Sche	Syllabus				
		IGCSE – NOVEMBER 2004 0654			0654			
5	(a)	(i)	Syllabus Add October 1997					
		(ii)	(ii) pass radiation between oppositely charged plates; attracted to positive plate; passes through paper, absorbed by aluminium;					
	(b)		evidence of working; 6000 years;					
	(c)	dama caus	[2]					
					Total 7			
6	(a)	(i)						
			description	name of element				
			most common metal	aluminium				
		m	ost common transition metal	iron				
			most common halogen	chlorine				
					[3]			
		(ii)	Na;		[1]			
		(iii)	silicon; oxygen;		[2]			
	(b)	oxygen in air is free element/exists as oxygen moleculres/O ₂ simple molecule oxygen in earth exists in compounds/as oxides/in giant structures;						
	(c)	(i)	reference to weathering/erosion; description of a weathering process;					
		(ii)	provision of minerals/trace eler	[2] [1]				
		(iii)	air; organic material/humus; water;					
			correct named substance;		max [2]			
					Total13			
7	(a)	teste ovari			[2]			
	(b)	they if sm	[2]					
	(c)	they chror	mber of [1]					

	Page 4		Mark Scheme	Syllaby
	Faye 4		IGCSE – NOVEMBER 2004	Syllabus 0654
	(d)	in the	hromosomes; e nucleus; enes; e of DNA;	Syllabus M. Barbarcannia (2) Total 7
8	(a)	yes; yes; no; no; yes;		all correct [2] [four correct [1]]
				[.ou. cocot[.]]
	(b)	stays at 0°	s the same; 'C;	[2]
	(c)	gene	erator;	[1]
	(d)	mea	sure of energy output to energy input/useful energy;	[1]
				Total 6
9	(a)	skier grea	r; ter area in contact with ground;	[2]
	(b)	(b) pressure = 720/360;		
		= 2 1	N/cm2;	[2]
	(c) redu		ce friction;	[1]
	(d)	(i)	the same; momentum is conserved;	[2]
		(ii)	speed of woman greater than that of man; momentum = mass x velocity;	
			ration of 3:2;	max [2]
				Total 9
10	(a)	(i)	exothermic means reaction gives out heat/reference to i from 20°C;	ncreased temperature [1]
		(ii)	experiment 4; mixture is (still) acidic/pH is below 7/is 1;	[2]
		(iii)	pH is 7/mixture is neutral; (this only happens) when amounts of acid and alkali are	equal; [2]
		(iv)	4000 dm ³ ;	[1]
		. ,	·	

this would not produce a neutral mixture/mixture would be alkaline; alkaline solution causes pollution;

[2]

(v)

	Dogo 5		Morde Schome			
	Page 5		Mark Scheme IGCSE – NOVEMBER 2004	Syllabus 0654	20	
	(b)	(i)	run-off from agricultural land may contain pollutants; illegal dumping;	0004	Na Canne	
		(ii)	chlorination/use of ozone;		[1]	
				т	otal 10	
11	(a)	(i)	grass → hog deer → tiger;		[1]	
		(ii)	energy (transfer);		[1]	
		(iii)	grass;		[1]	
	(b)	(i)	digest proteins; to amino acids/polypeptides;		[2]	
		(ii)	amylase digests starch; no starch in tiger's diet/meat does not contain starch/sta	rch only found i		
			plants;		[2]	
	(c)	hair/	fur;		[1]	

Total 8