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UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the May/June 2011 question paper for the guidance of teachers

0654 CO-ORDINATED SCIENCES

0654/23

Paper 2 (Core Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2011 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

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(a) (i) reference to: timescale / time to renew; action of heat/pressure; action of microorganisms / decay; (ii) oxygen; (iii) glucose molecules join / link together; to form long chains; [2] [1] **(b)** (C₆H₁₄) largest / heaviest; (c) (i) nitrogen; water (vapour); [2] (ii) (mix gas with) limewater; goes cloudy; [2] (iii) carbon monoxide; [2] nitrogen dioxide; [Total: 12] 2 (a) power = energy/time; = 8000/600 = 13.3; W; [3] **(b) (i)** KE = $\frac{1}{2}$ mv²; $= 0.5 \times 2 \times 40 \times 40 = 1600 (J)$; [2] (ii) 1600 J (or same answer as (i)); energy is conserved; [2] (c) expanded polystyrene / air / gas is a poor conductor of heat;

concrete block is a poor conductor of heat;

aluminium reflects heat back into house;

trapped air cannot carry heat around by convection;

[Total: 10]

[max 3]

	Page 3	Mark Scheme: Teachers' version	Syllabus V
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3		triceps; biceps;	Cambridge
	` '	B: contracts ; A: transmits force from B to bone ;	oe.c

C: relaxes; [3] (b) (i) increases; steady / linear, (increase); from 0.6 to 1.1 (g/cm^3) /by 0.5 (g/cm^3) ; [max 2] (ii) these foods contain calcium; needed for bones; [2] (iii) any citrus fruit / blackcurrants / other valid food source; [1] (c) (i) (bone is) harder/stronger/less elastic/less smooth; [1] (ii) (on the surface of the bones) at the joint; reduces friction / allows bones to move smoothly over each other; [2] [Total: 13] (a) work done = force × distance; $= 700 \times 55 = 38500 (J);$ [2] **(b) (i)** 50 s; [1] (ii) constant speed; [2] of 36 m/s; (c) relationship between pressure, force and area; pointed end has small area and large pressure;

disc has large area and small pressure;

(d) less friction;

[max 2]

[Total: 8]

[1]

	Page 4			Mark Scheme: Teachers' version	Syllabus	0
	raye 4		r	IGCSE – May/June 2011	0654	Ob.
_						S. 1
5	(a)	(i)	hair	;		My.
		(ii)	large	e ears / large eyes / long neck (so eyes high above g	round) /long legs;	O ADAC AMBRIDGE
	(b)	(i)		sion ; alveoli ;		[2]
		(ii)	com	e oxygen can be absorbed (from the air)/t pensates for less oxygen; e oxygen supplied to cells; espiration;	aken in by lungs	[max 2]
	(c)	(i)	not e	to limiting factors ; enough grass to eat ; y eaten by, foxes / pumas ;		[max 2]
		(ii)	idea won'	to species diversity; of their importance in food chain/provide food food food to be extinct; r, e.g. tourism/moral arguments;	or pumas/so puma	[max 2]
						[Total: 10]
6	(a)	(i)	Grou	up 1 , Period 2 ;		[1]
	(ii)			revents reaction with air/oxygen/water/forms a pro		[2]
		(iii)		im atoms have two shells / only have two electrons i im atoms have three electrons ;	n first shell ;	[2]
	(b)	(i)	hydr	ochloric (acid) ;		[1]
		(ii)	carb	on dioxide ;		[1]
		(iii)	chlo	rine ;		[1]
	(c)	(i)	subs	stance which changes the way the body works;		[1]
		(ii)		d unexpected / uncontrolled effects (of impurities); d harming the user;		[may 1]

ensure correct dosage / owtte;

[max 1]

[Total: 10]

Page 5	Mark Scheme: Teachers' version	Syllabus	.0
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(a) straight lines; approx angles of incidence and reflection (correct by eye); (b) correct diagram; (c) (i) red, blue, green; [1] (ii) frequency or wavelength; [max 1] [Total: 5] 8 (a) (i) petals/nectary; [1] (ii) anther/stamen; [1] (iii) ovule; [1] (b) (pollination is) the transfer of pollen from anther to stigma; (fertilisation is) the fusion of male and female gametes; pollination takes place before fertilisation; [max 2] (c) (i) 17; [1] (ii) nucleus; [1] [1] (iii) DNA; (d) (i) sugars produced by photosynthesis in leaves; transported to flowers in phloem; as sucrose; mineral ions in xylem; [max 2] (ii) for respiration / for energy / to make nectar / any energy-requiring process; [1] [Total: 11]

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Page 6	Mark Scheme: Teachers' version	Syllabus
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9	(a)	(i)	temperature; acid concentration; use the same acid; surface area of the metal; volume of acid;	[max	COM
		(ii)	ignites / pops; hydrogen is given off;	[2]	
		(iii)	both A and C did not react/cannot decide between A and C /two of the metals did not react;	ne [1]	L
	(b)	(i)	electrolyte in beaker ; electrodes in electrolyte ; voltmeter connecting electrodes ;	[3]	
		(ii)	voltage changes; because voltage depends on the metals used for electrodes;	[2]	
				[Total: 11]	
10	(a)	(i)	uranium ;	[1]	
		(ii)	nuclei ; energy ; turbine, generator ; (both needed for mark)	[3]	
	(b)	(i)	lead or concrete;	[1]	
		(ii)	damages cells / kills cells / mutation / damages DNA; cancer;	[2]	
			radiation sickness; radiation burns / burns skin;	[max 2]	
	(c)	(i)	Geiger counter / GM tube etc. ;	[1]	
		(ii)	3 half-lives; 300 (years);	[2]	
				[Total: 10]	