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UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2011 question paper for the guidance of teachers

0654 CO-ORDINATED SCIENCES

0654/31

Paper 3 (Extended Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2011 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

					mn	Dana Cambridge
	Pa	ge 2		Mark Scheme: Teachers' version	Syllabus	a.
				IGCSE – October/November 2011	0654	Par
1	(a)	(i)	arro	w going from right to left;	·	and.
		(ii)	cent	ral nervous system/spinal cord/brain ;		Tale
		(iii)	(very	ies/transmits, impulses/electrical signals/action pot y) long; of connecting central nervous system with distant p lin speeds up (impulse/transmission);	tentials ;	[max 2]
			e.g. pass	nect to, another cell/organ; muscle/gland/effector; ses on, impulse/electrical signal/action potentials; that branches make many contact points;		[max 2]
	(b)	(i)	on n	es/contains, information/instructions; naking proteins; of hereditary material;		[max 2]
		(ii)	twice	e as much (in motor neurone) ;		[1]
						[Total: 9]
2	(a)	(i)	0.5 v	waves per second/0.5Hz ;		[1]
		(ii)		itudinal – wave motion is in same direction as distur sverse – wave motion at right angles to disturbance		[2]
	(b)	(KE = ½	=) ½ × 50	½ mv²;) × 12 × 12 = 3600 J;		[2]

(c) (i) no – weight is determined by gravitational attraction of a mass/gravity is the

yes – weight is determined by gravitational attraction of a mass but this depends on height/distance between mass and centre of gravitational force;

(ii) KE will be greater because, more potential energy converted into KE/has

been accelerating for longer/his speed is greater;

(d) energy = mass × shc × change in temperature;

[max 1]

[max 1]

[3]

[Total: 10]

same at both points;

 $= 50\ 000 \times 4\ 200 \times 5$; = 1\ 050\ 000\ 000 J;

Page 3	Mark Scheme: Teachers' version	Syllabus	
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- 3 (a) (i) speeds up reactions/provides lower activation energy route;
 - (ii) reaction (to make gases) is reversible; so reactants can never be fully used up/some product changes back to reactants/some gases pass through without reacting;
 - (iii) $2 SO_2 + O_2 \implies 2 SO_3$; (or correct multiple) [1]
 - (iv) sulfur dioxide; [1]
 - (b) symbols shown in correct atoms;
 three bond pairs around central atom;
 lone pair correctly shown and no others;

 [3]
 - (c) (i) calculate M_r of ammonium nitrate = $(14 \times 2) + (1 \times 4) + (16 \times 3)/80$; calculate mass of 0.1 moles = $0.1 \times 80 = 8g$; [2]
 - (ii) NO₃⁻; reference to charge balance given 1:1 ratio of ions; [2]

[Total: 12]

- 4 (a) conduction from filament to filament support/gas;
 radiation from filament (to the materials of the lamp);
 convection of the (noble) gas;
 [3]
 - **(b)** $(60-54) \div 60$; = 10%;
 - (c) (i) decreases; to constant (minimum) value; [2]
 - (ii) 0.2(0)A; [1]
 - (iii) power = voltage × current; 230 × 0.20 = 46W; [2]
 - (d) 1/R = 1/R1 + 1/R2; = 1/1000 + 1/2000; $R = 2000/3 = 666.7 \Omega$; [3]

[Total: 13]

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(a) (i) carbon;

elements contain one type of atom/carbon is listed in the Periodic Table; compounds contain more than one, type of atom/element (bonded); (reject 'mixed')

(ii) (both have a) giant structure/or good attempt to describe; so large numbers of bonds to break (which needs energy); (all these) bonds are strong;

[max 2]

(iii) diamond is harder/has stronger bonds, than sapphires/rubies;

(b) (i) the idea of attraction between opposite charges;

[1]

[1]

(ii) Al^{3+} gain and O^{2-} lose electrons; Al^{3+} gains three and O^{2-} loses two electrons; some relevant maths;

[3]

(e.g. so if six electrons then number of Al atoms is $6 \div 3 = 2$)

[Total: 10]

(a)

enzyme	one site of production	substrate	product	
amylase	salivary glands	starch	maltose	
protease/trypsin /pepsin	stomach/pancreas (see note below)	proteins	amino acids	
lipase	pancreas	fats/lipids	fatty acids and glycerol	

note: if protease given, allow either stomach or pancreas

if trypsin, must be pancreas

if pepsin, must be stomach

one mark for any two correct ;;;;

[4]

(b) villi;

very long/coiled;

increased surface area;

good blood supply/good capillary system;

has thin wall;

[max 2]

(c) (i) hepatic portal vein;

[1]

(ii) urea;

[1]

(iii) kidneys;

[1]

	Page 5		j	Mark Scheme: Teachers' version Syllabus		
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	(d)	(i)	_	ucose, cells would take up water by osmosis ;	•	DaCambridge
		(ii)	resp	energy ; biration ; cose oxidised/glucose combined with oxygen ;		9
				movement/other named use of energy;		[max 3]
						[Total: 14]
7	(a)	(1/2	× 5 ×	may be shown on graph/idea of area under graph; 8) + (15×8) + $(\frac{1}{2} \times 5 \times 8)$;		
		= 10	60 m	• ,		[3]
	(b)			mass × acceleration ; .5 = 105N ;		[2]
				,		
	(c)			power × time ; 5 = 3000 J ;		[2]
	(d)	heat transferred into (water) particles (from surroundings); (water) changes from liquid to gas; ref. attraction between particles in the liquid; fastest moving/more energetic, particles escape;				
		(escape) at surface/ref. to process happening at temperature below boiling point; average energy of rest of particles reduced/heat removed from liquid;			[max 3]	
						[Total: 10]
8	(a)	(C)	high	density and (high) electrical conductivity;		[1]
	(b)	(i)	delo	ocalised electrons/sea of electrons/the outer shell ele	ctrons ;	[1]
		(ii)	word	gram shows atoms of two different sizes; ds or diagram imply layer structure disrupted; ns of different size prevent layers of the other atoms fidea that more force needed to move layers/atoms;	rom sliding ;	[max 3]
	(c)	the	idea	idea that cell voltage is related to relative metal reaction that the greater the difference in reactivity the greaty difference between Cu and Zn is greater than betwe	ater the voltage/the	[2]

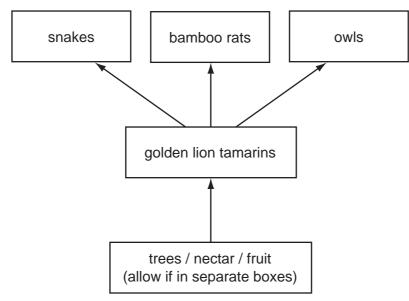
Page (Mark Scheme: Teachers' version	Syllabus
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(d) (i)	2CO + 2NO \longrightarrow 2CO ₂ + N ₂ (formulation (allow one mark for CO + NO \rightarrow CO ₂ + N)	ae + balanced) ;;
(ii)	reference to increased rate of reaction;	46.CO
(iii)	greenhouse effect/global warming/climate chang	

- (d) (i) 2CO + 2NO \longrightarrow 2CO₂ + N₂ (formulae + balanced) ;; (allow one mark for CO + NO \rightarrow CO₂ + N)
 - (ii) reference to increased rate of reaction;
 - (iii) greenhouse effect/global warming/climate change; much carbon dioxide (in exhausts)/carbon dioxide not reduced by converters/carbon dioxide made in converter;

[2]

[Total: 12]





plants and tamarins connections correct;

all three predators in separate boxes and with connections to tamarins correct;

all arrows in right direction;

[3]

- (ii) energy is lost along the food chain; ref. to one way in which energy is lost; idea that there is less energy for, top predators/at end of food chain; [2]
- (b) (i) fewer faeces further from tree/v.v.; furthest distance from tree is 400 m; figures quoted, e.g. 31% of faeces deposited within 50 m of tree; [max 2]
 - (ii) faeces provide nutrients for, young plants/seedlings (**not** seeds); less competition (for seedlings) away from parent tree; example of factor competed for (e.g. light, water, soil nutrients); help to colonise new areas; [max 3]

[Total: 10]