**CAMBRIDGE INTERNATIONAL EXAMINATIONS** International General Certificate of Secondary Education

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## 0654 CO-ORDINATED SCIENCES

0654/51

Paper 5 (Practical), maximum raw mark 45

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2012 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.

Page 2	Mark Scheme	Syllabus	Ľ,
	IGCSE – October/November 2012	0654	
	ow: 10, 10, 10, 10 ; d row: 0, 5 – 10, 0, 5 – 10 ;	Syllabus 0654 BabaCo	mbrides
becau water becau do not	ot necessary ; ise seeds in <b>D</b> germinate (as well as in <b>B</b> ) ; is necessary ; ise seeds in <b>A</b> and <b>C</b> do not germinate/because see t germinate/because seeds in <b>B</b> and <b>D</b> do germinate/ i wool do germinate ;	eds in dry cotton wool	[4]
	prove reliability/because just one seed might be dead nt of individual variability ;	d or damaged/to take	[1]
(d) tempe	erature/oxygen/air/carbon dioxide/soil pH/soil type/	minerals ;	[1]
tube S tube F	<b>R1</b> colour recorded as red/orange/yellow/green ; <b>61</b> colour recorded as no change/blue (not blue-green <b>R2</b> colour recorded as orange/brown/yellow ; <b>62</b> colour recorded as blue/black ;	ו) ;	[4]
reduci	e – seeds (not just <b>S2</b> ) ; ing sugar – radicles/roots (not just <b>R1</b> ) ; have correct observations in the table)		[2]
<b>(g)</b> amyla	se/carbohydrase/diastase ;		[1]
		[Tota	1. 451

Page 3		Mark Scheme	Syllabus	
		IGCSE – October/November 2012	0654 230	
(a) (i)	entry	r for <b>d</b> for 50 g (must be < 60) ;	- 31	nor:
(ii) / (iii)	all re	ainder of entries for <b>d</b> (60, 70, 80, 90g) ; adings to nearest cm or all to nearest 0.1 cm <i>(cons</i> ) ues decrease for increasing <b>m</b> ;	Syllabus 0654 vistency) ;	[3]
(b) (i)	0.01 0.014	ree 1/m values: 7/0.0167 (not 0.016) 4/0.0143 (not 0.0142) 3/0.0125 (not 0.012) ;		[1]
(ii)	4 poi best (no g	cal axis linearly numbered <b>AND</b> labelled ; nts plotted correctly within ½ square ; straight line ; graph marks for plotting wrong column from table by grated from a straight line)	ut allow gradient to be	[3]
(iii)	∆ <i>y</i> <b>A</b> gradi	ing shown either in space <b>or</b> on graph as coordinat <b>ND</b> change in <b>d</b> must be at least 10 (or 4 cm of pap ient value from a correct working method ; gradient marks from a graph with a curve or point to	per vertically) ;	[2]
(iv)	value	e using mass of rule = 300 – (gradient from <b>(b) (iii)</b> /	(10) ;	[1]
(c) (i)		ass × distance values calculated and entered in tak w if only four masses in table)	ble ;	[1]
(ii)	avera	age mass × distance value ;		[1]
(iii)	value	e for mass of rule ;		[1]
<b>(d)</b> adv	antag	e of plotting shows anomalous results clearly ;		[1]
			[Tota]	

Page 4	Mark Scheme Syllabus	×
	IGCSE – October/November 2012 0654	2
(a) (i)	Mark Scheme Syllabus   IGCSE – October/November 2012 0654   first value entered in column 2 of table and < 10 ;	ambr
	two more readings in column 2 ; all readings to 1 decimal point ;	19
	2 of the readings within 0.4 cm <sup>3</sup> ;	[3]
(iii)	column 3 completed (10 – column 2) ;	[1]
(iv)	average calculation for $V_{av}$ ;	[1]
	correct values used ( $V_{av} = (a)(iv)$ , $c_a = 0.013$ and $V_a = 10$ ); correct rearranging $c_s = 2 \times c_a \times V_a/V_{av}$ ; correct $c_s$ calculated value to 2 (or more) significant figures; (correct value only scores all 3 marks) (calculation mark may be awarded following wrong substitution and/or wrong rearrangement providing all terms included)	[3]
	colour = red/orange <b>AND</b> pH = 1 – 4 ;	[1]
(ii)	colour = yellow (or orange if <b>(b) (i)</b> is red) <b>AND</b> pH > <b>(b) (i)</b> pH and < 7 ;	[1]
(iii)	colour = yellow/green <b>AND</b> pH = 6 – 7 but not < <b>(b) (ii)</b> pH ;	[1]
<b>(c)</b> colo	ur = purple <b>AND</b> pH = 10 – 14 ;	[1]
2 sp 1 sp <b>OR</b>	ium hydroxide because) atula loads calcium carbonate and still not neutral (reference to <b>(b)</b> ) ; atula load calcium hydroxide produced greater increase in pH (ref to <b>(c)</b> ) ; atula load calcium hydroxide produced greater increase in pH than 1 spatula	

[Total: 15]