

Cambridge IGCSE[™]

CO-ORDINATED SCIENCES

0654/11

Paper 1 Multiple Choice (Core)

May/June 2021

45 minutes

You must answer on the multiple choice answer sheet.

You will need: Multiple choice answer sheet

Soft clean eraser

Soft pencil (type B or HB is recommended)

INSTRUCTIONS

There are forty questions on this paper. Answer all questions.

- For each question there are four possible answers **A**, **B**, **C** and **D**. Choose the **one** you consider correct and record your choice in soft pencil on the multiple choice answer sheet.
- Follow the instructions on the multiple choice answer sheet.
- Write in soft pencil.
- Write your name, centre number and candidate number on the multiple choice answer sheet in the spaces provided unless this has been done for you.
- Do not use correction fluid.
- Do not write on any bar codes.
- You may use a calculator.

INFORMATION

- The total mark for this paper is 40.
- Each correct answer will score one mark.
- Any rough working should be done on this question paper.
- The Periodic Table is printed in the question paper.



- 1 What is respiration?
 - A breakdown of food by enzymes in the alimentary canal
 - **B** breathing to supply oxygen to cells
 - **C** release of carbon dioxide from the lungs
 - D release of energy for body activities
- **2** The photograph shows a bumble bee at a magnification of $\times 6$. The line shows the length of the bumble bee.



What is the actual length of the bumble bee?

- **A** 0.05 mm
- **B** 20 mm
- **C** 126 mm
- **D** 720 mm
- 3 Three food tests are carried out on a sample of food. The results are shown in the table.

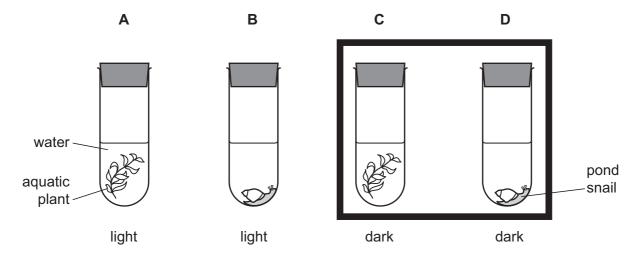
food test	final colour
Benedict's	blue
biuret	blue
iodine	blue-black

From these results, which nutrient is in the food?

- A reducing sugar
- **B** protein
- C starch
- **D** vitamin C

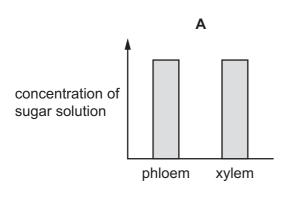
- 4 What is an enzyme?
 - A a carbohydrate that speeds up the rate of a reaction
 - **B** a carbohydrate that alters the activity of a target organ
 - **C** a protein that alters the activity of a target organ
 - **D** a protein that speeds up the rate of a reaction
- **5** Four test-tubes were set up as shown.

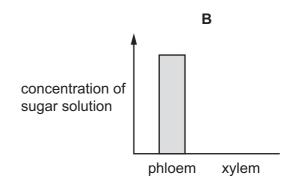
Which test-tube will contain the most dissolved oxygen after 24 hours?

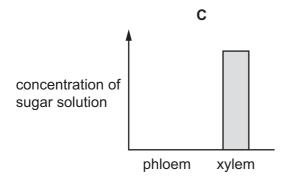


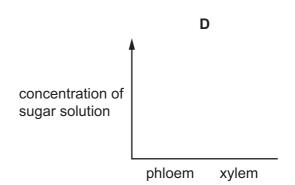
- **6** Which term is defined as the movement of digested food molecules into the cells of the body where they are used?
 - A assimilation
 - **B** absorption
 - C egestion
 - **D** ingestion

7 Which graph shows the concentration of sugar solution found in phloem and xylem?









8 A child blows into a rubber balloon.

What is the percentage of oxygen inside the balloon?

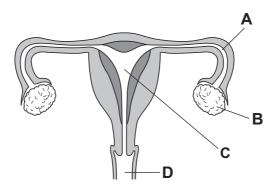
- **A** 0%
- **B** 4%
- **C** 16%
- **D** 21%

9 What is homeostasis?

- A keeping internal conditions constant
- **B** keeping the body at the same temperature as the environment
- **C** sweating to keep the body warm
- D vasoconstriction of arterioles to increase heat loss

10 The diagram shows the female reproductive system.

Where does fertilisation usually occur?



11 Cystic fibrosis is an inherited condition caused by a recessive allele.

A man and woman have a child who has cystic fibrosis. Neither parent has the condition.

What is the probability of their next child also having cystic fibrosis?

A 0%

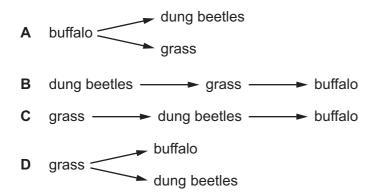
B 25%

C 50%

D 75%

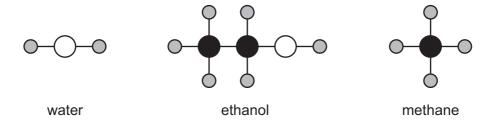
12 Dung beetles lay their eggs in the faeces of plant-eating mammals like buffalo. Both the adult beetles and their young stages eat the **undigested** food in the faeces.

Which diagram shows this food relationship?



- 13 What does **not** contribute to the carbon cycle?
 - **A** combustion
 - **B** deforestation
 - C transpiration
 - **D** photosynthesis

14 The structures of some substances are shown.



Which row shows the total number of different elements and the total number of atoms in the three structures?

	total number of different elements	total number of atoms
Α	3	9
В	3	17
С	7	9
D	7	17

15 Pure substance X has a melting point of 110 °C.

The melting point ranges of four impure samples of substance X are measured.

What is the melting point range of the most impure sample of substance X?

	melting point/°C
Α	81–85
В	86–92
С	98–99
D	102–110

16 Which statement about a carbon dioxide molecule is correct?

A It is composed of metallic elements, which are covalently bonded.

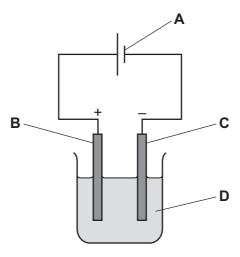
B It is composed of metallic elements, which are ionically bonded.

C It is composed of non-metallic elements, which are covalently bonded.

D It is composed of non-metallic elements, which are ionically bonded.

17 Apparatus used for electrolysis is shown.

Which label identifies the electrolyte?



18 The catalytic converter in the exhaust of a car brings about the reaction shown.

2NO + 2CO
$$\rightarrow$$
 2CO₂ + N₂

Which row about this reaction is correct?

	oxidation	reduction	
Α	✓	✓	key
В	✓	X	✓ = occurs
С	X	✓	x = does not occur
D	X	X	

19 Three different tests are carried out on an aqueous solution of substance X.

The results are shown.

test	result
add a few drops of aqueous sodium hydroxide	blue precipitate
add acidified aqueous silver nitrate	no precipitate
add acidified aqueous barium nitrate	white precipitate

What is X?

- A copper(II) chloride
- B copper(II) sulfate
- **C** iron(II) chloride
- **D** iron(II) sulfate

20 What reacts with ammonia gas?

	hydrochloric acid	sodium hydroxide	
Α	✓	✓	key
В	✓	X	✓ = reacts
С	X	✓	x = does not react
D	x	X	

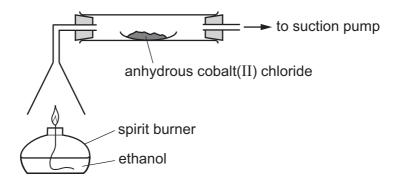
21 Substance Q is added to cold water. It floats on the water and hydrogen gas is made.

What is Q?

- **A** iodine
- **B** lithium
- **C** magnesium
- **D** zinc
- 22 Which statements describe carbon?
 - 1 It forms basic oxides.
 - 2 It is used as an electrode in electrolysis.
 - 3 It is used to extract iron from its ore.
 - **A** 1, 2 and 3 **B** 1 and 2 only **C** 1 and 3 only **D** 2 and 3 only

23 The apparatus used to investigate the combustion of ethanol is shown.

The products of combustion are passed over anhydrous cobalt(II) chloride.



How does the colour of the cobalt(II) chloride change during the investigation?

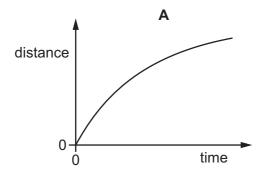
- A blue to pink
- B blue to white
- C pink to blue
- **D** white to blue
- **24** Other than hydrogen and oxygen, which substance provides only **one** of the essential elements for plant growth?
 - **A** K₃PO₄
- B KNO₃
- $C (NH_4)_3PO_4$
- **D** NH₄NO₃

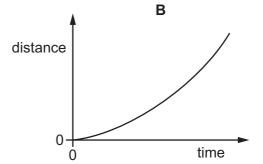
- 25 What is the chemical name for lime?
 - A calcium carbonate
 - B calcium hydroxide
 - C calcium oxide
 - D calcium sulfate
- 26 What is the main constituent of natural gas?
 - A ethane
 - **B** ethanol
 - C ethene
 - **D** methane

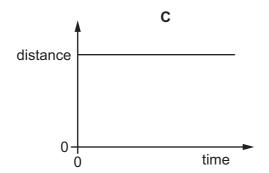
Which row shows the structure of the monomer used to make poly(ethene) and the structure of a section of poly(ethene)?

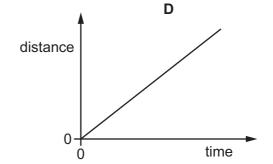
	monomer	section of poly(ethene)
A	H H H—C—C—H H H	H H H H
В	H H H—C—C—H H H	H H H H
С	H H 	H H H H
D	H H 	H H H H

28 Which distance-time graph represents an object that is decelerating?





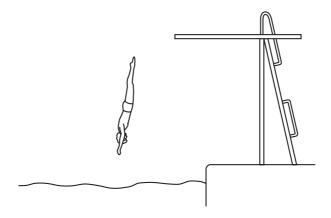




29 A vehicle moves a total distance of 300 m in 40 seconds.

What is its average speed?

- **A** 0.13 m/s
- **B** 7.5 m/s
- **C** 340 m/s
- **D** 12000 m/s
- **30** Which statement applies to a system in equilibrium?
 - **A** There is a resultant force and there is a resultant turning effect on the system.
 - **B** There is a resultant force but there is no resultant turning effect on the system.
 - **C** There is no resultant force but there is a resultant turning effect on the system.
 - **D** There is no resultant force and there is no resultant turning effect on the system.
- **31** The diagram shows a man diving into water.



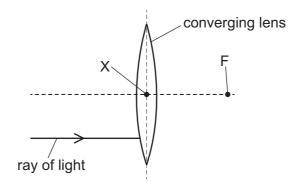
Which form of energy is increasing as he accelerates downwards through the air?

- **A** chemical
- B elastic potential (strain)
- **C** gravitational potential
- **D** kinetic
- 32 Which two quantities are used to calculate power?
 - A distance moved and time taken
 - B speed and distance moved
 - C work done and speed
 - **D** work done and time taken

33 Which row shows a good thermal conductor and a bad thermal conductor?

	good conductor	bad conductor
Α	copper	air
В	copper	silver
С	water	air
D	water	silver

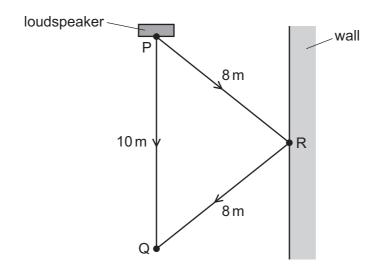
34 The diagram shows a ray of light striking a converging lens.



What happens to the ray after it passes through the lens?

- **A** It continues in its original direction.
- **B** It travels away from the principal focus F of the lens.
- **C** It travels towards the centre X of the lens.
- **D** It travels towards the principal focus F of the lens.

35 Sound from a loudspeaker at P travels directly to Q. Sound also reaches Q after being reflected from a wall at R.

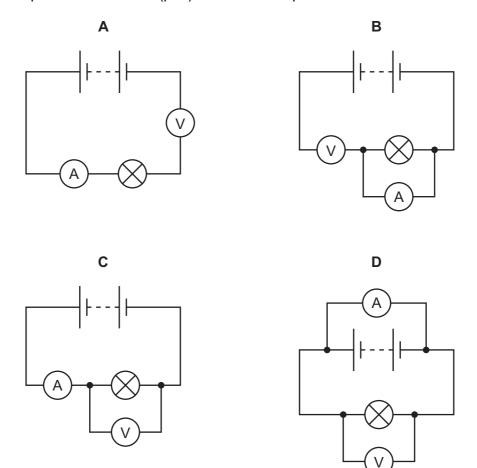


The speed of sound is 330 m/s.

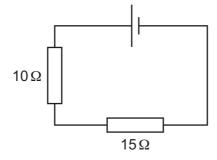
What is the difference in time for sound to travel from P to Q by the two routes?

- **A** $\left(\frac{6}{330}\right)$ s **B** $\left(\frac{16}{330}\right)$ s **C** (6×330) s **D** (16×330) s

36 Which circuit shows an ammeter that measures the current in the lamp and a voltmeter that measures the potential difference (p.d.) across the lamp?



37 A student connects a circuit as shown.



What is the total resistance of the circuit?

A $5.0\,\Omega$

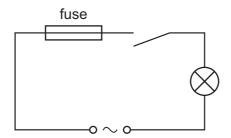
B 10 Ω

C 15Ω

D 25Ω

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38 A student connects the circuit shown.



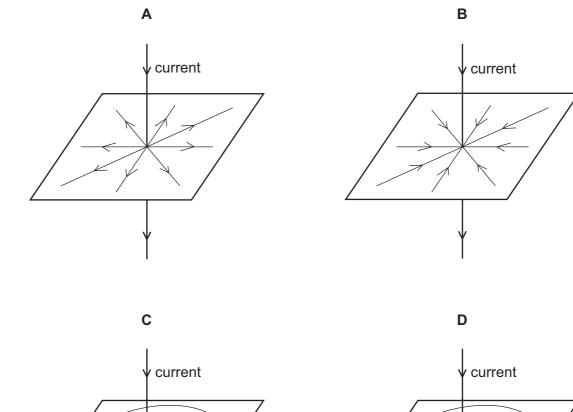
When the switch is closed the fuse blows and stops the current.

What is a possible reason for this?

- **A** The current rating of the fuse is too high.
- **B** The current is too large.
- **C** The lamp is too dim.
- **D** The voltage is too small.

39 The diagrams each show a wire carrying a current in the direction of the arrow.

Which diagram shows the pattern and the direction of the magnetic field around the wire?



40 A radioactive nucleus emits a β -particle.

What happens to the proton number (atomic number) of the nucleus?

- A It stays the same.
- **B** It increases by 1.
- C It decreases by 2.
- **D** It decreases by 4.

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The Periodic Table of Elements

	=>	F 5	helium 4	10	Ne	neon 20	18	Ā	argon 40	36	궃	krypton 84	54	Xe	xenon 131	98	R	radon			
				6	ட	fluorine 19	17	Cl	chlorine 35.5	35	Ŗ	bromine 80	53	П	iodine 127	85	Ą	astatine _			
	>			8	0	oxygen 16	16	ഗ	sulfur 32	34	Se	selenium 79	52	<u>e</u>	tellurium 128	84	Ъ	polonium –	116	^	livermorium -
	>			7	z	nitrogen 14	15	۵	phosphorus 31	33	As	arsenic 75	51	Sp	antimony 122	83	Ξ	bismuth 209			
	≥			9	ပ	carbon 12	14	Si	silicon 28	32	Ge	germanium 73	20	Sn	tin 119	82	Pb	lead 207	114	Εl	flerovium
	=			5	В	boron 11	13	Ρl	aluminium 27	31	Ga	gallium 70	49	In	indium 115	81	11	thallium 204			
										30	Zu	zinc 65	48	ပ္ပ	cadmium 112	80	Нg	mercury 201	112	C	copernicium
										29	Cn	copper 64	47	Ag	silver 108	79	Αn	gold 197	111	Rg	roentgenium -
dn										28	Z	nickel 59	46	Pd	palladium 106	78	പ	platinum 195	110	Ds	darmstadtium -
Group										27	ပိ	cobalt 59	45	몺	rhodium 103	77	'n	iridium 192	109	¥	meitnerium -
		- I	hydrogen 1											Ru	ruthenium 101	9/	Os	osmium 190	108	Hs	hassium -
				-						25	Mn	manganese 55	43	ည	technetium -	75	Re	rhenium 186	107	Bh	bohrium –
					pol	ass						chromium 52		Mo	molybdenum 96	74	≥	tungsten 184	106	Sg	seaborgium -
			Key	atomic number	atomic symbo	name relative atomic mass				23	>	vanadium 51	41	qN	niobium 93	73	<u>Б</u>	tantalum 181	105	Ор	dubnium -
					ato	rela				22	j	titanium 48	40	Zr	zirconium 91	72	Ξ	hafnium 178	104	껖	rutherfordium -
										21	Sc	scandium 45	39	>	yttrium 89	57–71	lanthanoids		89–103	actinoids	
	=			4	Be	beryllium 9	12	Mg	magnesium 24	20	Ca	calcium 40	38	ഗ്	strontium 88	56	Ba	barium 137	88	Ra	radium -
	_			8	:=	lithium 7	1	Na	sodium 23	19	¥	potassium 39	37	Rb	rubidium 85	55	Cs	caesium 133	87	Ē	francium -

r ₁ Lu	lutetium 175	103	۲	lawrencium	I	
70 Yb	ytterbium 173	102	8	nobelium	I	
69 Tm	thulium 169	101	Md	mendelevium	_	
₈₈ П	erbium 167	100	Fm	ferminm	I	
67 Ho	holmium 165	66	Es	einsteinium	_	
66 Dy	dysprosium 163	86	ŭ	californium	I	
65 Tb	terbium 159	26	ă	berkelium	_	
64 G d	gadolinium 157	96	Cm	curium	I	
e3 Eu	europium 152	98	Am	americium	I	
62 Sm	samarium 150	94	Pu	plutonium	_	
Pm	promethium -	66	ď	neptunium	I	
9 P N	neodymium 144	92	\supset	uranium	238	
₅₉	praseodymium 141	91	Ра	protactinium	231	
Ce Ce	cerium 140	06	H	thorium	232	
57 La	lanthanum 139	68	Ac	actinium	1	
lanthanoids			actinoids			

The volume of one mole of any gas is $24\,\mathrm{dm}^3$ at room temperature and pressure (r.t.p.).