



UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

AL EXAMINATIONS ducation

NAME					
CENTER NUMBER			CANDIDATE NUMBER		

CO-ORDINATED SCIENCES (DOUBLE)(US)

0442/23

Paper 2 (Core)

May/June 2013

2 hours

Candidates answer on the Question Paper.

No Additional Materials are required.

READ THESE INSTRUCTIONS FIRST

Write your Center number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use a pencil for any diagrams or graphs.

Do not use staples, paper clips, glue or correction fluid.

DO **NOT** WRITE IN ANY BARCODES.

Answer all questions.

Electronic calculators may be used.

You may lose marks if you do not show your working or if you do not use appropriate units.

A copy of the Periodic Table is printed on page 28.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [] at the end of each question or part question.



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1 Fig. 1.1 shows an experimental car powered by solar panels.

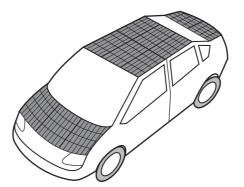


Fig. 1.1

(a) The speed/time graph in Fig. 1.2 shows the motion of the car over a short time.

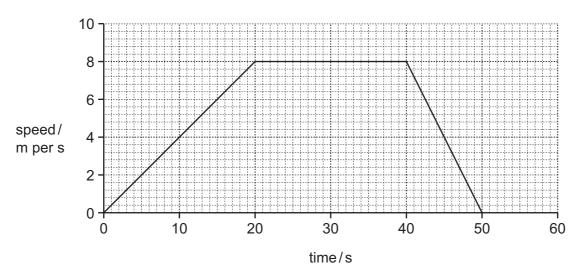


Fig. 1.2

On Fig. 1.2 label

N at a point at which the car was not moving,

A at a point when the car was accelerating,

C at a point at which the car was traveling at constant speed.

[3]

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(b) The energy output from the solar panels was measured during one day. Fig. graph of the results.

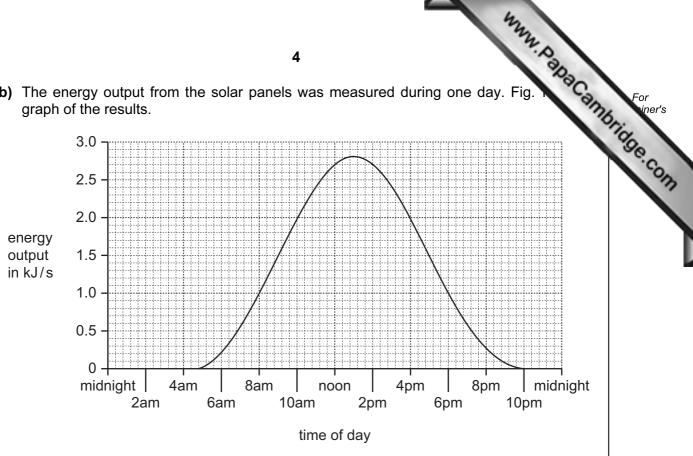


Fig. 1.3

	(i)	Describe how the energy output from the solar panels varies during one day.
		[2]
	(ii)	Explain why the energy output from the solar panels varies during the day.
		[1]
(c)	Gei	nerators are used to produce electricity in power stations.
		scribe how energy from a named fossil fuel is transferred to the generator in a ver station.
		[3]

(d) Fig. 1.4 shows a small photovoltaic cell (solar cell) being investigated.

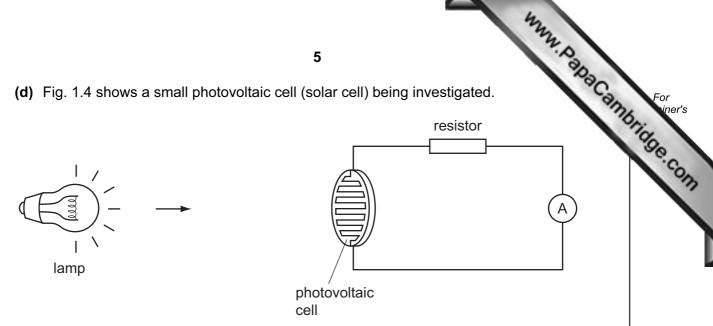


Fig. 1.4

A voltmeter is added to the circuit to measure the voltage across the photovoltaic cell.

Using the correct symbol, draw the voltmeter in the correct position on Fig. 1.4. [2]

(e) The car has mirrors to help the driver see behind the car. The driver sees a truck in his mirror as shown on Fig. 1.5.

Use Fig. 1.5 to describe two characteristics of an image seen in a plane mirror apart from size.

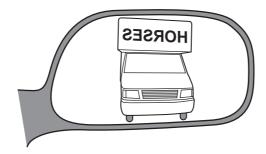


Fig. 1.5

[3]

2 Petroleum (crude oil) is a mixture of hydrocarbons.

.....

	The state of the s	
	6	
role	eum (crude oil) is a mixture of hydrocarbons.	-'c
	aree useful products obtained from petroleum are refinery gas, gasoline (petrol) and esel oil (gas oil).	,
(i)	State one use for each of these products.	m
	refinery gas	
	gasoline	

(ii) Name two compounds that are produced when hydrocarbons undergo complete combustion.

diesel oil

2	[2]
Explain why combustion of hydrocarbons is an example of an oxidation reaction	١.

(b) Fig. 2.1 shows a simplified diagram of a process which is used to convert large saturated hydrocarbon molecules into smaller, more useful molecules.

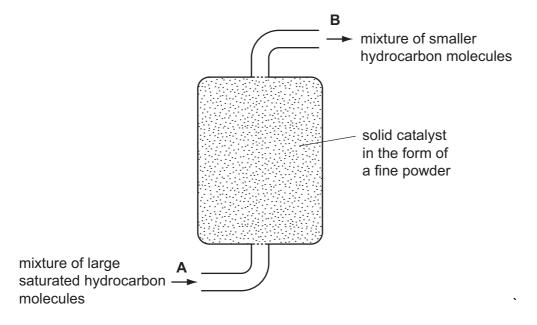


Fig. 2.1

Name the process shown in Fig. 2.1. [1] (i)

(iii)

and policy for iner's

(ii) A chemist takes samples of the mixture of compounds from point **A** and point Fig. 2.1.

He adds bromine solution to each sample and shakes the mixture.

Predict and explain the appearance of each mixture after shaking with bromine solution.

sample from point A
sample from point B
explanation
[4]

www.papaCambridge.com Fig. 3.1 shows part of a food web in a northern forest. The arrows show the direct energy flow.

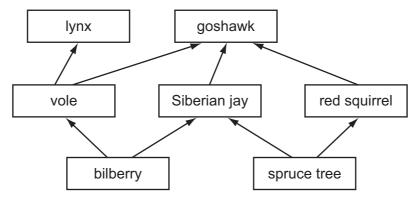


Fig. 3.1

(a) Complete Table 3.1 by selecting two organisms from the food web that belong in each column.

You can use each organism once, more than once or not at all.

Table 3.1

	producer	consumer	herbivore	carnivore
organism 1				
organism 2				

[4]

(b) If the forest is cut down, the species in the food web may not be able to survive.

List **two other** undesirable effects that may occur if the forest is cut down.

1	
2	[2

(c) State three ways in which energy is used in the body of an animal, such as a lynx.

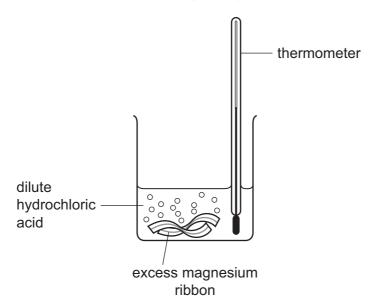
1	
2	

3

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4 A student added excess magnesium to dilute hydrochloric acid.

During the reaction, the thermometer reading changed.



(a) (i) State **two** observations which show that a chemical change occurs when magnesium is added to dilute hydrochloric acid.

	1	
	2	[2]
(ii)	Name the gas that is given off in this reaction and describe a test for this gas.	
	name	
	test	
		[2]
iii)	Explain why the pH of the mixture increases during the reaction.	

(b) The student set up the apparatus shown in Fig. 4.1.

She investigated the rate of reaction between magnesium and dilute hydrochloric acid

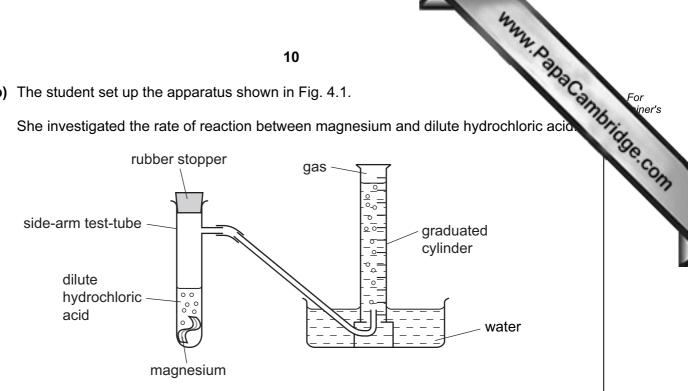


Fig. 4.1

At the start of the experiment, the graduated cylinder contained no gas and was full of water.

(i)	The student knew that the speed at which the gas is produced is a good way of measuring the rate of reaction.
	What should the student measure to find the rate at which gas is produced?
	[2]
(ii)	State two variables that affect the rate of reaction between magnesium and dilute hydrochloric acid.
	1

			8
5	(a)	Visi	ble light and γ -(gamma) radiation are two regions of the electromagnetic spec
		(i)	Name a region of the electromagnetic spectrum that is used in remote contributions devices for televisions.
			[1]
		(ii)	State one way in which the waves in different regions of the electromagnetic spectrum differ from each other.
			[1]
	(b)	Fig.	5.1 shows a light ray passing from the air through a glass fiber, and back out into air.
			air A glass fiber
			Fig. 5.1
		Use	e one of the phrases to complete the sentences below.
		Eac	ch phrase can be used once, more than once or not at all.
			hits at an angle greater than the critical angle.
			hits at an angle less than the critical angle.
			is passing into a less dense medium.
			is passing into a more dense medium.
		The	e ray of light changes direction at
		A b	ecause it
		B b	ecause it
		C b	ecause it

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[3]

(c)	One s	source of background ra	diation is cosmic rays.		, Ca	2
	Cosm	nic rays are 90% protons	s, 9% $lpha$ -(alpha) particles ar	nd 1% electrons.		3
	(i) V	What is an $lpha$ -particle?				
					[1]	
	(ii) N	lame a source of backgr	round radiation apart from	cosmic rays.		
					[1]	
(d)		ollowing sentence abou orrect.	ut $lpha$ -particles was written	by a student. The s	tatement is	
		α-particles can p	ass through a thin s	heet of lead		
	Chan	ge the statement to mak	ke it correct.			
	Write	your correct statement I	below.			
					[1]	
(e)	Unde	rline the two pieces of e	equipment that detect ioniz	ing radiations.		
		ammeter	Geiger-Müller tube	litmus paper		
		newton-meter	photographic film	thermometer		
					[2]	
(f)	Three	e of the following statem	ents are true. Tick the corr	ect statements.		
	Both o	α -(alpha) radiation and β	s-(beta) radiation pass easil	y through the body.		
	α -rad	iation damages cells in a	a very localized area of the	e body.		
	Ioniza	ation does not always kil	l cells – sometimes it caus	es them to mutate.		
	Canc	er occurs when a large r				
		J	number of cells are killed.			
	The d	•	number of cells are killed. d depends on the length o	f exposure.	[2]	

(g)	Most atoms contain ele	ctrons, protons and neutrons.	VaCa,
	State which of these pa	rticles	
	has the least mass,		Ì
	has no charge,		
	has a negative charge,		
	are in the nucleus.	and	[4]

6 (a) The words in the list below are all related to human reproduction.

www.PapaCambridge.com Choose words from the list to match each description. You may use each word once more than once or not at all.

	oviduct	prosta	te gland	t	sp	erm	
	testis	urethra	uter	us	7	zygote	
G. CO	ned when the female gamet						
a male ga	ımete						
the organ	in which sper	ms are made					
the place	where fertiliza	ation occurs					

[4]

(b) Fig. 6.1 shows changes in the thickness of a woman's uterus lining over a time interval of 45 days.

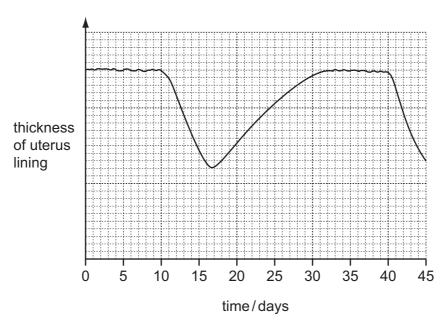


Fig. 6.1

(i)	Use Fig. 6.1 to estimate the number of days for which one menstrual cycle laste	d.
		[1]
(ii)	Suggest the day on which an egg was released from the woman's ovaries.	
		[1]

(c)	Αw	voman with HIV/AIDS can pass the disease to her child.	Paca
	(i)	What does the abbreviation HIV stand for?	[1]
	(ii)	Describe how a woman can pass the disease to her child.	

www.PapaCambridge.com 7 (a) (i) Copper is used to make water pipes, cooking pots and electrical wires. State three different properties of copper that make it a suitable material for the [3] (ii) Name the family of metals in the Periodic Table which includes copper. **(b)** Bronze is a mixture containing copper and tin. (i) State the general name of materials such as bronze. (ii) State one advantage of bronze compared with copper. [1] (c) Fig. 7.1 shows a process in which a copper compound is split into elements. d.c. power supply orange-brown bubbles layer forming of gas

Fig. 7.1

(i)	Name the process shown in Fig. 7.1.		[1	[ا
-----	-------------------------------------	--	----	-----

(ii) On Fig. 7.1 label the cathode. [1]

www.PapaCambridge.com (iii) One of the products of the process shown in Fig. 7.1 is a gas. This gas bloom damp litmus paper. Name the copper compound that is being separated into its elements. Explain your answer. name of compound explanation

www.PapaCambridge.com 8 Fig. 8.1 shows a washing machine. When the door is closed and the machine is sw on, an electric motor rotates the drum and clothes.

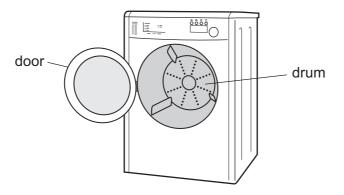


Fig. 8.1

heat

(a) Choose words from the list below to complete the sentences.

chemical

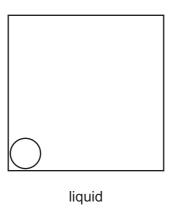
	nuclear	gravitational po	otential	sound	
In an elect	ric motor, the use	ful energy transfer	is electrical en	ergy into	
			energy.		
Some of th	e electrical enerç	gy supplied to the m	otor is wasted	as	
			energy and		
			energy.		[2]

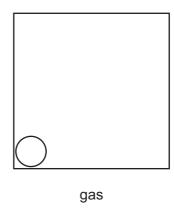
kinetic

light

- (b) Inside the washing machine, some of the water evaporates when the washing machine is being used.
 - (i) During evaporation, water changes state from liquid to gas.

Complete the diagrams to show the arrangement of particles in a liquid and in a gas.





[3]

		19 Explain, in terms of particles, the process of evaporation.	1	
		19	S	1
	(ii)	Explain, in terms of particles, the process of evaporation.		aCam,
				[3]
)	A 0 220	current of 3A passes through the heating element when the voltage ac 0V.	ross	it is
	Ca	culate the resistance of the heating element.		
	Sta	te the formula that you use and show your working.		
		formula		
		working		
			Ω	[2]

9 Fig. 9.1 shows a pitcher plant, which grows in Malaysia and Indonesia.

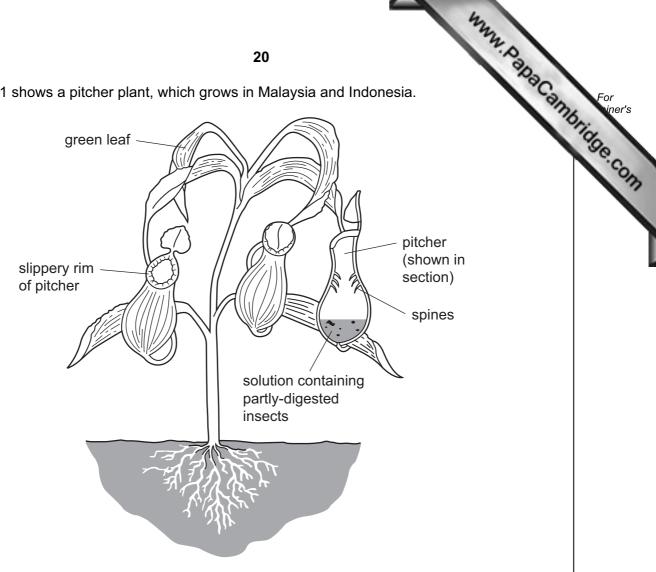


Fig. 9.1

- (a) The leaves of pitcher plants carry out photosynthesis, using carbon dioxide and water to make carbohydrates. They obtain carbon dioxide and water in the same way as other plants.
 - (i) Complete Table 9.1 to show how the leaves obtain carbon dioxide and water. You do not need to write anything in the shaded box.

Table 9.1

substance	source	part of plant that absorbs it	process by which it is absorbed
carbon dioxide	air		
water			

ГΛ	٦
14	

(11)	write the word equation for photosynthesis.	
		[2]

(b) Pitcher plants grow where the concentration of nitrate ions in the soil is very low plants need nitrate ions to make amino acids and proteins.

	May
	21
	cher plants grow where the concentration of nitrate ions in the soil is very low nts need nitrate ions to make amino acids and proteins.
	cher plants grow where the concentration of nitrate ions in the soil is very low nts need nitrate ions to make amino acids and proteins. Cher plants use a different way of obtaining amino acids. They trap insects in their chers, and produce a solution that digests the proteins in the insects' bodies.
(i)	Describe two features of the pitchers, shown in Fig. 9.1, that help to trap insects inside them.
	1
	2[2]
(ii)	Define the term digestion.
	[2]
iii)	Suggest what is present in the solution that the pitcher plant produces inside its pitchers, to enable digestion to take place.
	[1]

(c) A scientist investigated the hypothesis that a scent produced by the rim of the

acts as a stimulus that attracts insects.

She took several identical Petri dishes.

- www.papaCambridge.com She placed a piece of the rim of a pitcher, or a small amount of solution from inside the pitcher *or* water, on one side of the dish (side **A**).
- She put a small amount of water on the other side (side **B**) as shown in Fig. 9.2.
- She then placed an insect in the center of the dish. She recorded which side of the dish the insect moved to.

She repeated this 19 more times, using a different insect each time.

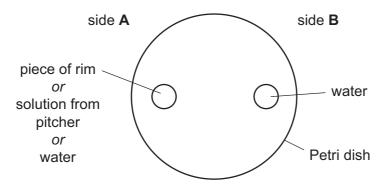


Fig. 9.2

Table 9.2 shows her results.

Table 9.2

substance on side A of dish	substance on side B of dish	number of moved to		
Side A of distr	side b of distr	Α	В	
piece of rim	water	16	4	
solution from pitcher	water	4	16	
water	water	10	10	

(i)	Suggest why the scientist placed water on both sides of some dishes.							
		. [1]						
(ii)	Do the results support the scientist's hypothesis? Explain your answer.							
		[2]						

10 (a) When wood is burnt, a solid material known as wood ash remains.

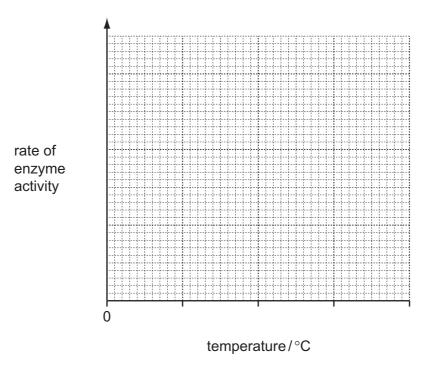
		42	
		23	1
(a)	Wh	en wood is burnt, a solid material known as wood ash remains.	C.S.
		en wood is burnt, a solid material known as wood ash remains. od ash contains calcium carbonate and potassium compounds, which can be use approve the quality of soil. Explain briefly how calcium carbonate and potassium compounds could improve the quality of soil.	3
	(i)	Explain briefly how calcium carbonate and potassium compounds could improve the quality of soil.	ve
		calcium carbonate	
		potassium compounds	
		[3]
	(ii)	Suggest how a sample of wood ash could be tested to show that it contains carbonate ions.	ed
		[2]
(b)		quality is also improved by the addition of nitrogen compounds such as ammoniu ate, $(NH_4)_2SO_4$.	m
	(i)	State the total number of atoms shown combined in the chemical formu $(NH_4)_2SO_4.$	la
		[1]
	(ii)	Ammonium sulfate is the product of a reaction between an alkaline solution ammonia and an acid.	of
		Name the acid that reacts with ammonia to form ammonium sulfate and state the type of chemical reaction that occurs.	ne
		name of acid	
		type of reaction [2]

(iii)	Outline how crystals of ammonium sulfate could be obtained from a solu ammonium sulfate.	Cannie	For viner's
		[2]	COM

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Please turn over for Question 11.

www.PapaCambridge.com 11 (a) Complete the graph in Fig. 11.1 to show how enzyme activity is affect temperature. You should write in at least two values for temperature on 'temperature' axis.



[3]

Fig. 11.1

(b) The internal body temperature of a human is kept constant, allowing enzymes to work efficiently. The skin helps to do this.

Fig. 11.2 shows a section through the skin in two different environmental conditions.

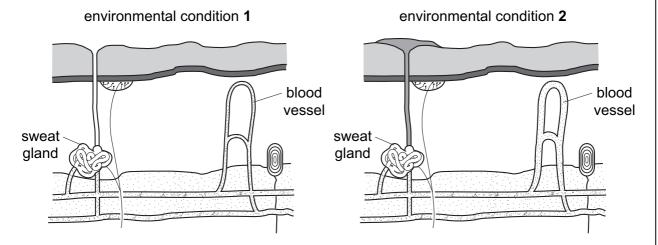


Fig. 11.2

	Describe two ways in which the skin in environmental condition 2 differenvironmental condition 1 .	\
(i)	Describe two ways in which the skin in environmental condition 2 different environmental condition 1 .	Cambi
	1	
	2	[2]
(ii)	Suggest how environmental condition 2 differs from environmental condition 1.	[2]
(iii)	The muscles also help to maintain a constant body temperature.	[1]
	Explain how the muscles can help to return a low body temperature to normal.	
		[2]

DATA SHEET The Periodic Table of the Elements

				2	8				mm.	DanaCambrid
0	4 He Helium	20 Ne Neon 10	40 Ar Argon	84 Kry Krypton 36	131 Xe Xenon	Rn Radon		175 Lu Lutetium 71	Lr Lawrencium 103	Cambri
		19 F Fluorine	35.5 C1 Chlorine	80 Br Bromine 35	127 T lodine	At Astatine 85		173 Yb Ytterbium 70	No Nobelium 102	13
>		16 Oxygen	32 S Sulfur 16	79 Selenium 34	128 Te Telurium	Po Polonium 84		169 Tm Thulium 69	Md Mendelevium 101	
>		14 N itrogen 7	31 P Phosphorus 15	75 As Arsenic 33	Sb Antimony 51	209 Bi Bismuth 83		167 Er Erbium 68	Fm Fermium 100	
≥		12 Carbon 6	28 Si Silicon	73 Ge Germanium	Sn Tin	207 Pb Lead		165 Ho Holmium 67	ES Einsteinium 99	(r.t.p.).
=		11 Boron 5	27 A1 Aluminum 13	70 Ga Gallium 31	115 In Indium	204 T (Thallium		162 Dy Dysprosium 66	Celifornium 98	pressure
				65 Zn Zinc 30	112 Cd Cadmium 48	201 Hg Mercury 80		159 Tb Terbium 65	BK Berkelium 97	The volume of one mole of any gas is $24\mathrm{dm}^3$ at room temperature and pressure (r.t.p.).
				64 Copper 29	108 Ag Silver	197 Au Gold		157 Gd Gadolinium 64	Cm Curium 96	n temper
				59 Nickel 28	106 Pd Palladium	Pt Platinum		152 Eu Europium 63	Am Americium 95	m³ at roor
				59 Co Cobalt 27	Rhodium Rhodium	192 Ir Iridium		Samarium 62	Pu Plutonium 94	as is 24 d
	T Hydrogen			56 Fe Iron	Ruthenium 44	190 OS Osmium 76		Pm Promethium 61	Np Neptunium 93	of any ga
				Mnn Manganese 25	Tc Technetium 43	Re Rhenium 75		Neodymium 60	238 U Uranium 92	one mole
				52 Cr Chromium 24	96 Mo Molybdenum 42	184 W Tungsten 74		Pr Praseodymium 59	Pa Protactinium 91	olume of
				51 Vanadium 23	Niobium 41	181 Ta Tantalum 73		140 Cer ium 58	232 Th Thorium	The v
				48 Ti Titanium	2r Zroonium 40	178 Hf Hafnium		1	mic mass ibol mic) number	
				45 Sc Scandium 21	89 × Yttrium	139 La Lanthanum 57 *	227 Ac Actinium 89	d series series	 a = relative atomic mass X = atomic symbol b = proton (atomic) number 	
=		9 Be Beryllium	Mg Magnesium	40 Ca Calcium	Strontium	137 Ba Barium 56	226 Ra Radium 88	*58-71 Lanthanoid series 190-103 Actinoid series	« × ∞	
-		7 Lithium	23 Na Sodium	39 K Potassium	85 Rb Rubidium 37	Caesium 55	Fr Francium 87	58-71 L 90-103,	Key	

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