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## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

## MARK SCHEME for the May/June 2010 question paper for the guidance of teachers

## **5129 COMBINED SCIENCE**

5129/02

Paper 2 (Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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Page 2		Mark Scheme: Teachers' version	Syllabus
	•	GCE O LEVEL – May/June 2010	5129
1	(a) correct s in paralle	symbol el with $6\Omega$ resistor	Cambridge
	(b) (i) V = 1.2 V	IR (or equivalent) or 0.2 × 6  2 (unit independent)	[3]

- 1 (a) correct symbol in parallel with  $6\Omega$  resistor
  - **(b) (i)** V = IR (or equivalent) or  $0.2 \times 6$ ٧ (unit independent)
    - (ii) 9 [1]

[3]

- 2 (a) magnesium and aluminium / Mg and Al metals [2] (answers are independent)
  - (b) blue / purple / violet red / orange / yellow [2]
  - (c) same number of / 2 electrons in outermost shell [1] (allow valence electrons)
- 3 (a) (i) 0 and 1 [1]
  - (ii) 2 and 3 [1]
  - (iii) 3 and 4 [1]
  - (b) iris / circular and/or radial muscles [1]
  - (c) a line which remains above the drawn line throughout [1] (can be along initial line but must not increase)
  - (d) retina [1]
- [1] (a) 27
  - (b) electron / e [1]
  - [1] (c) 32
  - (d) an indication of 2 half lives = 11 400 [2] 2 × 5700 with incorrect answer = 1 mark

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	Page 3		age 3 Mark Scheme: Teachers' version		Syllabus	er	
					y/June 2010	5129	OS .
5		liver amino ad kidneys liver	cids				Da Cambridge
6	(a)	neutralis	ation				[1]
	(b)	17 80					[2]
	(c)	(17 × 2) = 0.425 k (ecf from	kg / 425 g				[2]
7	(a)		ack is a better emitte / white is a poorer er		(of thermal radiation) ator		[1]
	(b)	conducti	on				[1]
	(c)		on iir expands iir is less dense	}	any one		[1]
8	(a)	filter add chlo sedimen		}	any two		[2]

[2]

[2]

[1]

[2]

[2]

(b) remove solids / insoluble particles / dirt

(d) diffusion over short distance / easily diffuse

light penetrates to all parts / more light readily absorbed

kill bacteria / micro-organisms solids settle to the bottom

(a) X = palisade / mesphyll Y = (lower) epidermis

(b) spongy mesophyll

through stomata

(c) by diffusion

9

				May
	Pa	ge 4	Mark Scheme: Teachers' version	Syllabus
			GCE O LEVEL – May/June 2010	5129
10	(a)	no yes r 4 = 2 m	no yes arks 3 = 1 mark	Syllabus Parta Cannbhage
	(b)	(i) plas		[1] [1]
		` ,		
11	(a)	haemati	ite / magnetite	[1]
	(b)		2 3 ow correct multiples)	[1]
		(ii) rem	noval of oxygen / gain of electrons / lower oxidation s	tate [1]
		` ´ carl	te / carbon burns to form carbon dioxide bon dioxide reacts with carbon bon reacts with oxygen = 1 mark	[2]
	(c)	too read	ctive / more reactive than iron / carbon	[1]
12	(a)	(i) 60		
		<b>(ii)</b> 8 (iį	gnore any sign)	[2]
	(b)		on / disturbance / displacement / motion of particles dicular to the direction / motion of the wave	[2]
13	(a)	crush / t increase mix food make sv	own of large pieces to small pieces / tear / grind food e surface area d with saliva / enzymes wallowing easier references to chewing / digestion)	vo [2]
	(b)		ectly identified) ecayed teeth / less decay	[1]
	(c)	brush te visit der use mor eat mor		vo [2]
			<u> </u>	

accept converse in terms of town A

	i age o		CCE O LEVEL May/lune 2010	5120
			GCE O LEVEL – May/June 2010	5129
14	(a)	infra red	/ ir	5129 Phacambridge
	(b)	radio		3
	(c)	300 000	000 / 3 × 10 <sup>8</sup>	[1]
15	(a)	carbon d	lioxide	
		plan	ts animals	[3]
	(b)		osynthesis	
		Y = resp Z = deca	y / decomposition / (bacterial) respiration	[3]
16	(a)	D		[1]
	(b)	В		[1]
				541
	(c)	C		[1]
17	(a)	time = di 4.0	stance/speed or 0.8/0.2	[2]
	(b)	work = fo	orce × distance or 4 × 0.8	
		J / joules	(unit independent)	[3]
18	(a)	carbon d	lioxide team (any order)	[2]
	(b)	carbon n poisonou	nonoxide us / toxic / correct description of mode of action	[2]
	(c)		neral formula	[1]
		gradation	hemical properties  n in physical properties differ by CH <sub>2</sub> any one	[1]
		.5.1110100		[1]

Mark Scheme: Teachers' version

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Syllabus

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	GCE O LEVEL – May/June 2010	5129	100

**19** (a) potential / gravitational kinetic / movement

**(b)** 0.5

20 (a) (i) any date from Feb 28 to Mar 5

(ii) any date from Mar 10 to Mar 17 [1]

[1]

[2]

(b) (i) 28 days / 4 weeks [1]

any 2

(ii) age / menopause stress / emotional state / anxiety diet / malnutrition / starvation pregnancy / breast feeding exercise genetic factors / inheritance

21 (a) fermentation / anaerobic respiration [1]

(b) (i) provides enzymes [1]

(ii) prevents oxidation of ethanol / formation of ethanoic acid [1]

(c) (fractional) distillation [1]

(d) correct structure of ethanol [1]