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## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

## MARK SCHEME for the October/November 2010 question paper for the guidance of teachers

## **5129 COMBINED SCIENCE**

5129/02

Paper 2 (Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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Page 2	Mark Scheme: Teachers' version	Syllabus	er	
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- **1** (a) 0.18 (ignore units)
  - (b)  $V = IR \text{ or } 50 \times 0.03 \text{ or } 10 \times 0.15$ = 1.5 (V)
  - (c) Q = It or C = It or 0.15 × 300
     = 45
     C
     0.75 C gains 2 marks, 0.75 gains 1 mark
     unit mark is independent of the numerical answer
- 2 (a) (i) blue / purple / indigo / violet
  - (ii) OH<sup>-</sup> / hydroxide ion [1] ignore OH

[3]

[1]

- (b) pipette
  burette (do not accept biuret)
  neutral / neutralised [3]
- (c) (i)  $(NH_4)_2SO_4$  [1]
- (ii) fertiliser [1]
- 3 (a) v = d/t or speed = distance / time or  $2.7 \times 10^8$  /  $24 \times 60 \times 60$  3125 (m/s) [2] allow  $2.7 \times 10^8$  /24 = 11 250 000 for 1 mark allow  $2.7 \times 10^8$  /  $(24 \times 60) = 187$  500 for 1 mark
  - (b) F = ma or a = F/m or 45/200= 0.225 (m/s<sup>2</sup>) [2]
- 4 (a) anther / stamen (ignore pollen grains)
  sepal
  ovary / carpel [3]
  - (b) to attract insects for pollination [2]
  - (c) anther / stamen / X [1]

Page 3	Mark Scheme: Teachers' version	Syllabus er
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- (a) liquid irregular shape majority of particles touching gas random particles not touching
  - (b) melting condensation

[2]

6 (a) volume
density
length
resistance / resistivity
colour
e.m.f.
pressure

any 2

[2]

(b) smaller range
constriction
retains reading
triangular cross section
narrow bore / tube
more sensitive
(ignore more accurate / narrower alone)

[2]

(c) mercury would freeze / would be solid or alcohol stays liquid / does not freeze (ignore statement that mercury melts at -39°C)

[1]

**7** (a) (i) tubing

[1]

(ii) the water (in the beaker)

[1]

(iii) the starch do not accept starch and amylase

[1]

(b) (amylase is an) enzyme catalyses (breakdown of starch) starch is broken down / digested sugar / maltose diffuses into the water tube is permeable (to maltose / sugar) (allow correct description for diffusion)

[3]

any 3

8 (a) E = Pt or energy = power × time or 1800 × 120 = 216 000 J 3600 J gains 2 marks, 3600 gains 1 mark

[3]

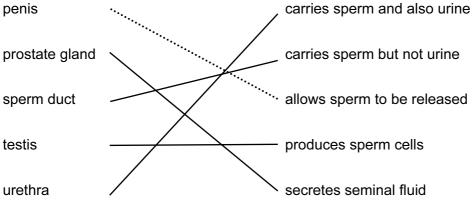
(b) neutral earth ANY order

[2]

			3
	Page 4	Mark Scheme: Teachers' version	Syllabus
		GCE O LEVEL – October/November 2010	5129
9	(a) oxygen		Camb

- 9
  - (b) hydrogen
  - (c) hydrogen [1]
  - (d) carbon monoxide [1]
  - [1] (e) argon
- 10 (a) (i) N [1] S
  - (ii) S Ν [1]
  - (b) current not changing / is constant / in one direction only magnetic field not changing / is constant [2]
- 11 (a) two parents (genetically) different offspring fertilisation / fusion of gametes or nuclei [2] allow converse argument

(b) penis



[4]

	Page 5	Mark Scheme: Teachers' version	Syllabus	er
Ī		GCE O LEVEL – October/November 2010	5129	100

		.900	GCE O LEVEL – October/November 2010	5129
12	(a)	5.4 (accept to	112 11.2 both above divided by 10) 5.6 above divided by 2) answers always gain credit)	5129 ADAC AND TION
	(b)	oxidation	n / redox	[1]
13	(a)	46°		[1]
	(b)	( )	= sin <i>i</i> / sin <i>r</i> (accept 27.79 to 28)	[1] [1]
14	(a)	gas B liquid C	= hydrogen = carbon dioxide = water = water / steam	[4]
	(b)	ethene c	ontains a (carbon to carbon) double bond	[1]
	(c)	speed up	o the reaction / lowers activation energy	[1]
15	(a)	(ii) thro	of water ugh stomata aves	[1]
	(b)		ours lours (accept 23.5 to 24)	[2] [1]
16	(a)	conducti	on / conduct / conductor	[1]

(b) it has expanded

[1]

				2.
Page 6		ge 6	Mark Scheme: Teachers' version	Syllabus
			GCE O LEVEL – October/November 2010	5129
17	(a)	(i)	8 10	Cambric
		(ii)	2 electrons on inner ring and 6 electrons on outer ring	Je CON
	(b)	san	ne element / same number of protons	

- **17 (a) (i)** 8
  - (ii) 2 electrons on inner ring and 6 electrons on outer ring
  - (b) same element / same number of protons different number of neutrons / nucleon number (ignore references to electrons)

(c) oxygen tents in hospital (oxy-acetylene) welding

> oxygen tanks for divers steel manufacture

any two [2] ignore breathing / saving lives / respiration / combustion answers which relate to breathing must say 'how' or 'what' is done

[2]

[4]

- 18 blood gland target organ liver
- **19** (a) d = m/v or 5.4/1.8= 3.0g/cm<sup>3</sup> [3] (correct answer with unit = 3 marks) (unit mark independent of answer)
  - **(b)** 2.8 [1]
- [1] 20 (a) large fish / fishermen
  - (b) mercury into water absorbed by micro-organisms (small) fish eat the micro-organisms [3]
  - (c) because they eat fish [1]