www.PatraCambridge.com

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

MARK SCHEME for the October/November 2011 question paper for the guidance of teachers

5129 COMBINED SCIENCE

5129/21

Paper 2 (Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2011 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

		7
Page 2	Mark Scheme: Teachers' version	Syllabus
	GCE O LEVEL – October/November 2011	5129
(a) red (cells	s) / erythrocytes ;	Candaria
(b) A;		The COM
(c) water ha	s entered the cells ;	

(c) water has entered the cells; by osmosis; cells have burst;

2 (a) velocity has direction / speed does not have direction; [1]

[2]

[3]

(c) kinetic thermal / heat / sound (in order);; [2]

3 [1] (a) alkanes;

(c) glowing splint; relights (incorrect test = 0 marks); [2]

4 1; 1; 0 / neutral;

5 (a) muscle; [1]

(c) pressure rises; blood forced out / leaves; [2]

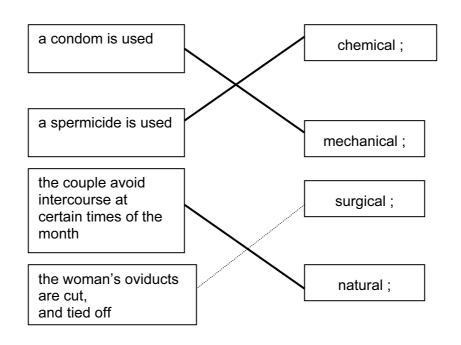
Page 3	Mark Scheme: Teachers' version	Syllabus
	GCE O LEVEL – October/November 2011	5129
blood (in	chamber Z) has less oxygen / is deoxygenated ; chamber Z) has more carbon dioxide ; se argument if no chamber stated assume Z]	Cambridge.c
(a) (i) norr	nal in correct position, perpendicular ;	[1]

- 6 (a) (i) normal in correct position, perpendicular;
 - (ii) ray looks as if coming from pin image left of mirror; [1]
 - (b) left / towards mirror / nearer the mirror; [1]
 - (c) sound / p-waves / push wave on a slinky; [1]

[2]

[3]

- 7 (a) shared pair; three lone pairs on each chlorine atom;
 - (b) increases / higher / bigger; [1]
 - (c) purification of water; [1]
 - (d) potassium chloride / KC1; iodine / I₂ (either order); [2]
- 8 (a) (i)



[1] (ii) the (contraceptive) pill;

Page 4	Mark Scheme: Teachers' version	Syllabus
	GCE O LEVEL – October/November 2011	5129

(iii) condom / abstinence; prevents body (fluid) contact / provides barrier;

(b) the milk is sterile; contains correct (balance of) nutrients; no danger of being made up at the wrong concentration; contains antibodies / provides immunity; at right temperature; promotes bonding between mother and baby; is always / available for the baby / no preparation;

any 2

9 (a) components connected in series; correct symbols for all four components;; correct symbols for two or three components (= 1 mark)

[1] [2]

[2]

(b) (i) V = IR or R = V/I or 1.8/0.2; 9; Ω / ohm (unit independent); (correct answer with unit = 3 marks)

[1] [1]

[1]

(ii) P = VI or 1.8 × 0.2; 0.36;

[1] [1]

10 (a) (i) iron / Fe;

[1]

(ii) 3 2;

[1]

(iii) 8-10;

[1]

(b) nitric acid; neutralisation / exothermic;

[2]

11 (a) protein; catalyst; speeds up reactions; any 1

[2]

(b) (i) faster at higher temperature / at 40° C / best at 40° C;

[1]

(ii) fastest / optimum at pH7; Increases to pH 7 then decreases;

[1]

(c) line drawn below 15°C curve throughout;

[1]

Page 5	Mark Scheme: Teachers' version	Syllabus	
	GCE O LEVEL – October/November 2011	5129	1

12 (a) conduction;

	(b)	heated water expands / becomes less dense; heated (hot) water rises; cooler (cold)water falls; convection (current); any 3	[3]
	(c)	(i) plastic is a poor conductor / insulator ;	[1]
		(ii) white surfaces are poor emitters; (ignore poor absorber / good reflector but good absorber / poor reflector / conduction is incorrect)	[1]
13	(a)	condenser;	[1]
	(b)	filter funnel ;	[1]
	(c)	measuring cylinder / burette ;	[1]
	(d)	pipette; burette correct order;	[2]
14	(a)	thin wall / wall one cell thick; folded / large surface area; many capillaries / moist surface; any 2	[2]
	(b)	carbon dioxide / CO ₂ / water / H ₂ O ;	[1]
	(c)	smoke / soot particles coat gas exchange surface ; or reduce effective surface area ; prevent diffusion	[2]
15	(a)	<pre>A = fermentation; B = water / steam; C = ethanoic acid;</pre>	[3]
	(b)	enzymes / zymase ;	[1]
	(c)	addition / additional ; monomer ;	[2]
16	(a)	magnetic materials are attracted to magnets;	[1]

Page 6 Mark Scheme: Teachers' version Syllabus er
GCE O LEVEL – October/November 2011 5129

		GCE O LEVEL - October/November 2011 5129	~ ~ ~ ~ ·
	(b)	steel is a hard magnetic material; steel retains magnetism / permanent; steel hard to magnetise; iron is soft magnetic; iron easily loses magnetism / temporary; iron easy to magnetise;	Cambridg.
17	(a)	a = F/m or 225 000/50 000 ; = 4.5 ; m/s ² (unit independent) ;	[1] [1] [1]
	(b)	(spacecraft) mass / weight decreases ;	[1]
18	(a)	less (in the cheetahs);	[1]
	(b)	energy losses at each stage; (heat loss due to) respiration; not all organisms being eaten; (energy losses due to) egestion / excretion / movement; any 1	[1] 1 [1]
	(c)	decomposers / bacteria / fungi / microorganisms ;	[1]
19	(a)	S ;	[1]
	(b)	Т;	[1]
	(c)	R; T (either order);	[2]
	(d)	W (accept U);	[1]