

Cambridge International Examinations Cambridge Ordinary Level

COMBINED SCIENCE

Paper 1 Multiple Choice

5129/11 October/November 2015 1 hour

Additional Materials:

Multiple Choice Answer Sheet Soft clean eraser Soft pencil (type B or HB is recommended)

READ THESE INSTRUCTIONS FIRST

Write in soft pencil.

Do not use staples, paper clips, glue or correction fluid. Write your name, Centre number and candidate number on the Answer Sheet in the spaces provided unless this has been done for you. DO **NOT** WRITE IN ANY BARCODES.

There are **forty** questions on this paper. Answer **all** questions. For each question there are four possible answers **A**, **B**, **C** and **D**.

Choose the **one** you consider correct and record your choice in **soft pencil** on the separate Answer Sheet.

Read the instructions on the Answer Sheet very carefully.

Each correct answer will score one mark. A mark will not be deducted for a wrong answer. Any rough working should be done in this booklet. A copy of the Periodic Table is printed on page 16. Electronic calculators may be used.

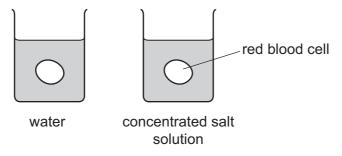
This document consists of 16 printed pages.



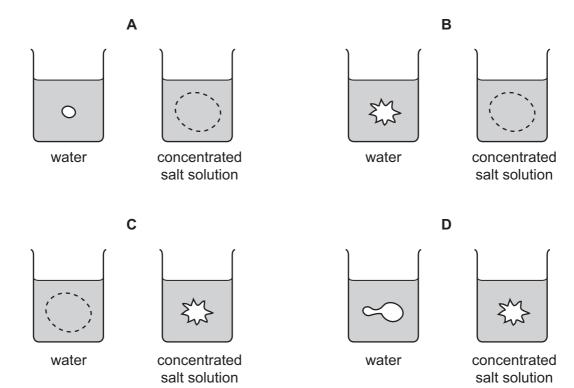
1 Which pair of statements best explains why plant cells are more angular in shape than animal cells?

	plant cells	animal cells
Α	cell wall present	cell wall absent
в	chloroplast present	chloroplast absent
С	one or more large vacuoles	small or no vacuoles
D	thin layer of cytoplasm	dense cytoplasm throughout cell

2 One beaker contains water. Another beaker contains a concentrated salt solution. A red blood cell is placed into each beaker.



Which diagram shows the appearance of the cells after 5 minutes?



3 The following reaction occurs in the human alimentary canal.

catalyst starch → products

What is the catalyst and what is one of the products?

	catalyst	product
Α	acid glucose	
в	acid	maltose
С	amylase	glucose
D	amylase	maltose

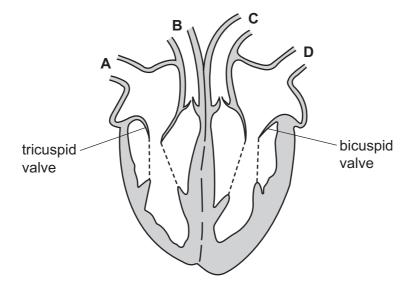
- 4 Which is the equation for photosynthesis?
 - $\textbf{A} \quad 6CO_2 \ \textbf{+} \ 6H_2O \ \rightarrow \ 6O_2 \ \textbf{+} \ C_6H_{12}O_6$
 - $\textbf{B} \quad 6CO_2 \ \textbf{+} \ C_6H_{12}O_6 \ \rightarrow \ 6O_2 \ \textbf{+} \ 6H_2O$
 - $\label{eq:constraint} \begin{array}{ccc} \mbox{6}\mbox{O}_2 \mbox{ + }\mbox{6}\mbox{H}_2\mbox{O} \mbox{ + }\mbox{6}\mbox{H}_{12}\mbox{O}_6 \end{array}$
 - $\textbf{D} \quad 6O_2 \ \textbf{+} \ C_6H_{12}O_6 \ \rightarrow \ 6CO_2 \ \textbf{+} \ 6H_2O$
- 5 Scientists observe that the adult skulls of Roman soldiers from 2000 years ago had perfect teeth.

This is most likely to be because in Roman times, compared to now, they had

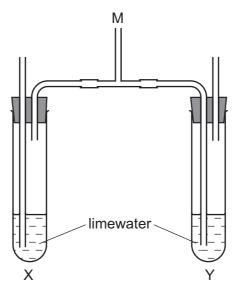
- **A** a diet containing fewer minerals.
- **B** a diet containing less sugar.
- C a diet containing more fat.
- **D** a diet containing more protein.
- 6 What is lost from the stomata during transpiration?
 - A carbon dioxide
 - B nitrogen
 - C oxygen
 - D water vapour

7 The diagram shows the heart.

Which vessel is an artery carrying deoxygenated blood?



8 The apparatus shown is used to investigate gas exchange during breathing.



Which describes the appearance of solutions X and Y after a person has exhaled several times through tube M?

	solution X solution Y	
Α	A clear clear	
в	clear cloudy	
С	cloudy	clear
D	cloudy	cloudy

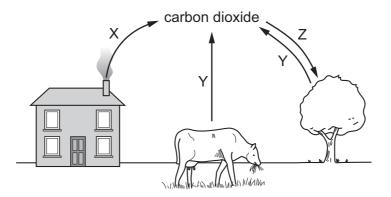
9 Where in the body are amino acids converted to urea and through which organ is urea excreted?

	amino acids converted to urea	urea excreted
Α	kidney	liver
в	kidney	stomach
С	liver	kidney
D	liver	stomach

- 10 Which group of chemicals are produced in human glands to have an effect on target organs?
 - A enzymes
 - **B** hormones
 - **C** proteins
 - D vitamins
- **11** Which row shows possible effects of the drug heroin?

	breathing rate	speed of nerve impulses
Α	decreased	decreased
в	decreased	increased
С	increased	decreased
D	increased	increased

12 The diagram shows part of the carbon cycle.



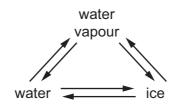
Which row describes processes X, Y and Z?

	Х	Y	Z
Α	combustion	photosynthesis	respiration
в	combustion	respiration	photosynthesis
С	photosynthesis	respiration	combustion
D	respiration	combustion	photosynthesis

13 AIDS (Acquired Immune Deficiency Syndrome) is a serious disease caused by the Human Immunodeficiency Virus (HIV).

Which interaction between people could transmit the virus?

- A being close to an infected person who sneezes
- **B** eating a meal with an infected person
- C having unprotected sexual intercourse with an infected person
- **D** kissing an infected person
- 14 In which conversion do water molecules lose speed?



A ice \rightarrow water

- $\textbf{B} \quad \text{ice} \rightarrow \text{water vapour}$
- **C** water vapour \rightarrow ice
- **D** water \rightarrow water vapour

	proton number	nucleon number
Α	8	16
В	12	24
С	C 21 45	
D	24	52

15 Which row shows an atom containing 24 neutrons?

16 Magnesium metal reacts with chlorine to produce magnesium chloride, an ionic compound. Which statement describes what happens to the atoms during the reaction?

- **A** A magnesium atom gains two electrons and a chlorine atom loses two electrons.
- **B** A magnesium atom gains two electrons and two chlorine atoms each lose one electron.
- **C** A magnesium atom loses two electrons and a chlorine atom gains two electrons.
- **D** A magnesium atom loses two electrons and two chlorine atoms each gain one electron.
- 17 Which 'dot and cross' diagram for ammonia, NH₃, is correct?

Α	В	С	D
H≚N≚H ×● H	H ^{×××} N [×] H ×● H	H × N × H × ● H	H×N•H ×× H

18 The equation shows the reaction between sodium and water.

xNa + yH₂O \rightarrow 2NaOH + H₂

What are the values of *x* and *y* for the equation to be balanced?

	X	У
Α	1	1
в	1	2
с	2	1
D	2	2

19 A spatula measure of compound Z is stirred into a beaker containing some dilute hydrochloric acid.

Another spatula measure of Z is stirred into a beaker of aqueous sodium hydroxide.

In both experiments, Z dissolves and a neutral solution is made.

What is Z?

- **A** a basic oxide
- B a carbonate
- C an acidic oxide
- D an amphoteric oxide
- 20 Which statement describes the elements in Group VII of the Periodic Table?
 - A They are diatomic molecules which get more reactive descending the group.
 - **B** They are diatomic molecules whose melting points increase descending the group.
 - **C** They are soft metals with low melting points.
 - **D** They are unreactive gases used in lamps and balloons.
- **21** Steel objects are often galvanised to prevent rusting.

Galvanising involves coating the object by dipping it in a molten metal.

Which metal is used?

- A chromium
- B lead
- C tin
- D zinc

T reacts slowly with hydrochloric acid.

Q does not react with acid.

R reacts with steam but not with cold water.

S reacts violently with cold water.

What is the order of reactivity of the four metals, most reactive first?

- $\mathbf{A} \quad \mathbf{Q} \to \mathbf{T} \to \mathbf{R} \to \mathbf{S}$
- $\textbf{B} \quad \textbf{Q} \rightarrow \textbf{R} \rightarrow \textbf{T} \rightarrow \textbf{S}$
- $\boldsymbol{\mathsf{C}} \quad S \to \mathsf{Q} \to \mathsf{R} \to \mathsf{T}$
- $\textbf{D} \quad S \to R \to T \to Q$
- 23 A number of different reactions produce hydrogen gas.

Which pair of reactions both produce hydrogen?

	reaction 1	reaction 2	
Α	a metal and an acid	a reactive metal and water	
В	a reactive metal and water	an acid and an alkali	
С	an acid and a carbonate	a metal and an acid	
D	an acid and alkali	an acid and a carbonate	

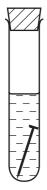
24 The diagrams show an investigation into the rusting of iron.

Which nail would you expect to have the least amount of rusting after one week?



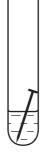
В

D



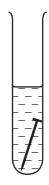
stoppered test-tube nail completely submerged in tap water

stoppered test-tube nail completely submerged in boiled tap water



С

open test-tube nail half submerged in tap water



open test-tube nail completely submerged in boiled tap water

25 The molecular formulae of four organic compounds, W, X, Y and Z, are shown.

W	Х	Y	Z
C_4H_8	C_3H_8	C_3H_6	C_4H_{10}

Which statement is correct?

- **A** W and Y have the same general formula.
- **B** W and Z have the same general formula.
- **C** X and Y belong to the same homologous series.
- **D** Y and Z belong to the same homologous series.
- **26** The fractions obtained from the fractional distillation of petroleum have different boiling point ranges.

Four of the fractions are known as kerosene, diesel, gasoline and bitumen.

Which fraction is correctly matched to its boiling range?

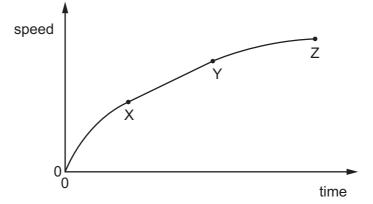
	fraction	boiling point range
Α	bitumen	greater than 350 °C
в	diesel	150 °C to 220 °C
С	gasoline	220 °C to 350 °C
D	kerosene	30 °C to 150 °C

27 Alkenes are a series of unsaturated hydrocarbons containing a double bond.

Which formula does not represent an alkene?

A C_2H_4 **B** C_3H_6 **C** C_4H_{10} **D** C_6H_{12}

28 The speed of a car moving along a straight road varies with time as shown.

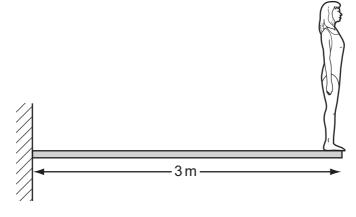


Which statement is correct?

- A Between X and Y the acceleration is constant.
- **B** Between X and Y the speed is constant.
- **C** Between Y and Z the car is slowing down.
- **D** Between 0 and X the acceleration is constant.
- **29** An object is taken from the Earth to the Moon. The gravitational field strength on the Moon is less than the gravitational field strength on the Earth.

Which statement is correct?

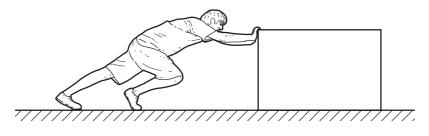
- A Both the mass and the weight of the object decrease.
- **B** The object's mass and weight both stay the same.
- C The object's mass decreases.
- D The object's weight decreases.
- **30** A diver, weighing 720 N, stands at the end of a springboard measuring 3 m long.



What is the moment about the support due to the weight of the diver?



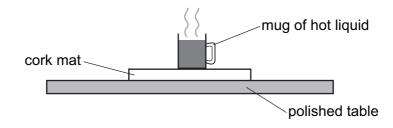
31 A man pushes a heavy box across a floor. He exerts a force of 80 N and the box moves 4.0 m in 5.0 seconds.



What useful power does the man develop?



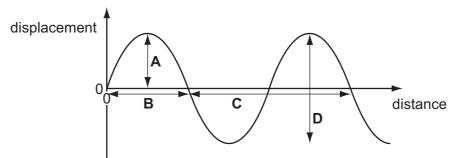
32 To protect a polished table, a cork mat may be put on the table underneath a mug containing hot liquid.



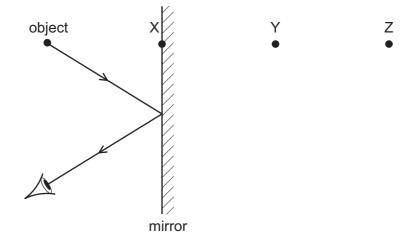
Why is this effective?

- **A** Cork is a good conductor.
- B Cork is a good radiator.
- **C** Cork is a poor conductor.
- **D** Cork is a poor radiator.
- **33** The diagram shows the displacement across a wave.

Which value is multiplied by the frequency to give the speed of the wave?

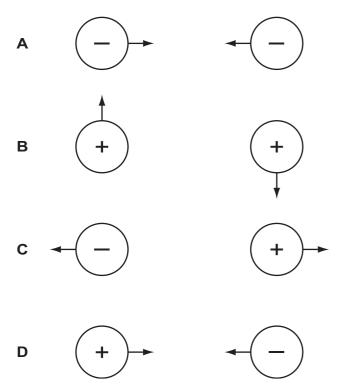


34 The diagram shows the reflection, in a plane mirror, of a ray of light from an object.

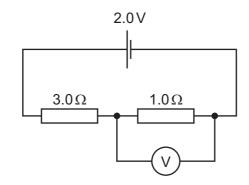


Which statement is correct?

- A The image is at X.
- **B** The image is between X and Y.
- **C** The image is at Y.
- **D** The image is between Y and Z.
- **35** Which diagram correctly shows the directions of the electrostatic forces on a pair of charged spheres?



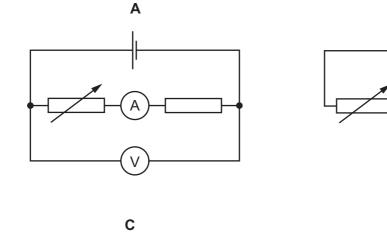
36 The diagram shows a voltmeter connected across a 1.0Ω resistor in a circuit.

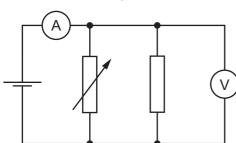


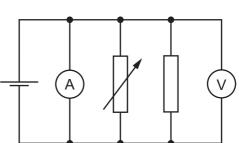
What is the reading on the voltmeter?

A 0.5V **B** 1.0V **C** 1.5V **D** 2.0V

37 Which circuit may be used to measure the resistance of a fixed resistor?







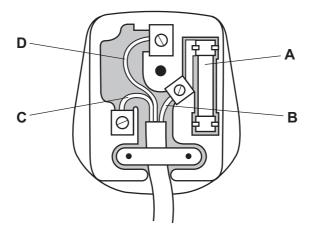
D

В

V

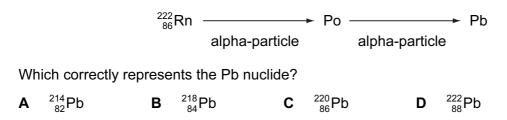
38 The diagram shows a plug for a device. The device has an outer metal case.

Which part is connected to the metal case of the device?



- 39 Which particle is positively-charged?
 - A electron
 - B neutral atom
 - **C** neutron
 - **D** proton
- **40** Radon is a naturally occurring radioactive gas.

The first two steps in the decay process of radon each involves the emission of an alpha-particle.



	0	4	He	Helium	20	Ne		40	Ar	Argon	84	Kr	Krypton 36	131	Xe	Xenon		Rn	Radon				175	Lu			Ľ	Lawrencium 103
	١١			2	19	ш	Fluorine 10	35.5	Cl	Chlorine 18	80	Br	Ø	127	н	53 bodine 54		At	Astatine 85				173	Yb Viterbium	70 71			102 Nobelium L
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	>						Nitrogen 7	31	٩	Phosphorus 15	75	As		122	Sb	Antimony 51	209	ä	Bismuth 83				167	Er Erbium				Fermium 100
	≥						Carbon 6		Si	Silicon 14	73	Ge	Germanium 32	119	Sn	50 Tin	207	Pb	Lead 82				165	Holmium Holmium	67		Es	Einsteinium 99
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	=				6	Be	Beryllium 4	24	Mg	Magnesium 12	40	Ca	50	88	Sr	Strontium 38	137	Ba	Barium 56	226	Ra	Radium 88	*58-71 Lanthanoid series	190-103 Actinoid series			×	
	_				2	:	Lithium 3	23	Na	Sodium 11	39	¥	Potassium 19	85	Rb	Rubidium 37	133	Cs	Caesium 55		ŗ	Francium 87	*58-71	t 90-103	L		Key	q



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