

## **Cambridge International Examinations**

Cambridge Ordinary Level

COMBINED SCIENCES 5129/21

Paper 2 Theory May/June 2016

MARK SCHEME
Maximum Mark: 100

## **Published**

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2016 series for most Cambridge IGCSE<sup>®</sup>, Cambridge International A and AS Level components and some Cambridge O Level components.

® IGCSE is the registered trademark of Cambridge International Examinations.



Pa	age 🛚		Syllabus	Paper
		Cambridge O Level – May/June 2016	5129	21
1	(a)	A = cell wall;		
	()	<b>B</b> = vacuole;		
		C = chloroplast ;		[3]
	(b)	difference : no chloroplasts/chlorophyll present ; reason : cell receives no light ;		
		-		
		difference : shape ; reason : more water absorption ;		[4]
		Todoon: more water absorption,		[-1
2	(a)	A :		[1]
_	(ω)	··,		1.1
	(b)	D·		[1]
	(~)			
3	(a)	chemical;		
•	()	gravitational potential/GPE;		
		kinetic;		[3]
		(I) 000 (II)		
	(b)	(i) 800 (W);		[1]
		(ii) 18 (m);		[1]
4	(a)	Т;		[1]
	(b)	only one spot/colour (in chromatogram);		[1]
	(c)	(i) it does not dissolve/is insoluble;		[1]
		(ii) R;		[1]
5	(a)	(i) 4000 (kg per hectare);		[1]
		(ii) more fertiliser added, the greater the yield;		
		effect becomes less pronounced as more is added;		[2]
	(b)	any <b>one</b> from		
		• temperature		
		<ul><li> (amount of) light</li><li> carbon dioxide (concentration);</li></ul>		[1]

**Mark Scheme** 

Syllabus

Paper

Page 2

Page 3	Mark Scheme	Syllabus	Paper
	Cambridge O Level – May/June 2016	5129	21

**6** (a) 23.7; allow 23.67 [1]

(b) radiation; conduction; [2]

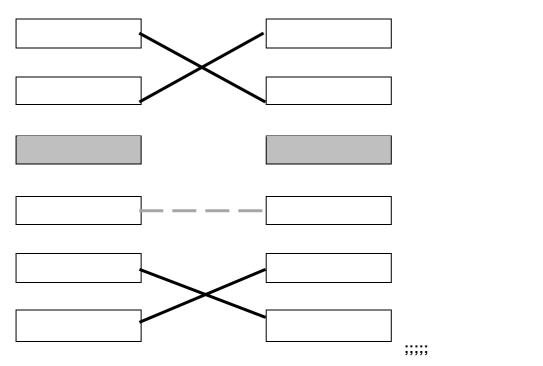
- (c) any one from
  - black absorbs heat
  - shiny reflects heat ; [1]

7 (a) 402; 32; 10.05;

(b) glowing splint;
 relights;
[2]

- (c) any one from
  - making steel
  - welding ; [1]
- (d) (s) (l) (g) [1]

8



[5]

Pa	age 4	4	Mark Scheme	Syllabus	Paper
			Cambridge O Level – May/June 2016	5129	21
9	(a)	line	parallel to incident ray from the block;		[1]
	(b)		$= \sin i / \sin r;$ 32.6;		[2]
			ow $\sin r = \sin 54/1.5$ ow 33		
10	(a)	(i)	9;		[1]
		(ii)	19;		[1]
	(b)	2, 7	′ (drawn on shells) ;		[1]
	(c)	(i)	halogens;		[1]
		(ii)	decreases;		[1]
11	(a)	(i)	any <b>three</b> from  • water		
			<ul><li>light</li><li>temperature</li></ul>		
			• oxygen ;;;		[3]
		(ii)	E; G;		[2]
	(b)	any •	three from starch stored in seed / cotyledon		
		•	starch broken down (amylase) starch to glucose/maltose		
		•	glucose / maltose soluble (glucose used in) respiration		
		•	energy used for growth (during germination) ;;;		[3]
12	(a)	30	;		[1]
	(b)	(i)	1.5 ; Volts / V ;		[2]
			v 51167 v ,		[4]
		(ii)	20(C);		[1]

Page 5		Mark Scheme		Paper
		Cambridge O Level – May/June 2016	5129	21
((		<ul> <li>any two from</li> <li>current is the same</li> <li>more work done (by the charge against higher resistance)</li> <li>resistance proportional to potential difference</li> <li>R has the highest resistance ;;</li> </ul>		[2]
13 (		A = oxidation ; B = polymerisation ; C = steam/water ;		[3]
(1	b)	speed up the reaction		[1]
(1	<b>c</b> )	fcH2 - CHE ]n		
		CH₂ chain open ended bracket showing multiple units		[2]
14 (		stronger magnet ; more coils ; louder sound ; answers must be comparative		[3]
(1	b)	(i) $v = f\lambda$ ; 5.5 (m); allow $\lambda = v/f$ or $\lambda = 330/60$		[2]
		at least two waves drawn with same frequency/equal time period; same amplitude;		[2]
		ignore extra waves less than two waves max 1		
	acid			[3]

Page 6		6	Mark Scheme		Paper	
			Cambridge O Level – May/June 2016	5129	21	
16	(a)	2	H <sub>2</sub> ; <b>both</b> required		[1]	
	(b)		cket) fuel ; king margarine ;		[2]	
	(c)		aporate to smaller volume ; ol/(allow to) crystallise ;		[2]	
		cry	stallisation alone is insufficient			
	(d)	any • •	two from conducts electricity conducts heat malleable ductile			
		•	high melting point/high boiling point high density ;;		[2]	
17	E; C; D;				[3]	
18	(a)	any • •	one from like poles repel (allow repulsion) align north-south when freely suspended made from iron/steel (allow nickel/cobalt);;		[1]	
	(b)	any •	one from induced magnetism steel bar has become magnetised/a magnet;		[1]	
	(c)	any •	one from will not attract loses magnetism more quickly;		[1]	
19	(a)		air ; acking) hydrocarbons ; allow water/petroleum (crude oil)/named hy	drocarbon	[2]	
	(b)	iror	n ; allow Fe		[1]	
	(c)	(i)	hydroxide ; allow OH <sup>-</sup>		[1]	
		(ii)	sulfuric acid ; allow H <sub>2</sub> SO <sub>4</sub>		[1]	

Page 7	Mark Scheme	Syllabus	Paper
	Cambridge O Level – May/June 2016	5129	21

## 20 any three from

- nerve/brain damage
- mental disorders (e.g. depression/manias/phobias)
- pancreatitis/damage to pancreas
- liver damage/cirrhosis
- (Increased risk of) cancer
- (increase risk of) heart disease
- hypertension/high blood pressure

• strokes ;;;

**21 (a)** 71; [1]

(b) (i) beta; allow electron [1]

(ii) a neutron becomes a proton; allow number of protons increases by 1 and number of neutrons decreases by 1 [1]