

## **Cambridge International Examinations**

Cambridge Ordinary Level

COMBINED SCIENCES 5129/22

Paper 2 Theory May/June 2016

MARK SCHEME
Maximum Mark: 100

## **Published**

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1	(a)	A = cell wall ;		
		B = vacuole ; C = chloroplast ;		[21
		C - Chloropiast ,		[3]
	(b)	difference: no chloroplasts/chlorophyll present;		
	(6)	reason : cell receives no light ;		
		difference : shape ;		
		reason : more water absorption ;		[4]
2	(a)	A;		[1]
	(b)	D;		[1]
3	(a)	chemical; gravitational potential/GPE;		
		kinetic;		[3]
	(b)	(i) 800 (W);		[1]
		(ii) 18 (m);		[1]
4	(a)	Т;		[1]
	(b)	only one spot/colour (in chromatogram);		[1]
	(c)	(i) it does not dissolve/is insoluble;		[1]
		(ii) R;		[1]
5	(a)	(i) 4000 (kg per hectare);		[1]
		(ii) more fertiliser added, the greater the yield;		
		effect becomes less pronounced as more is added;		[2]
	(b)	any <b>one</b> from		
		<ul><li>temperature</li><li>(amount of) light</li></ul>		
		carbon dioxide (concentration);		[1]

**Mark Scheme** 

Syllabus

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**6** (a) 23.7; allow 23.67 [1]

(b) radiation;conduction;[2]

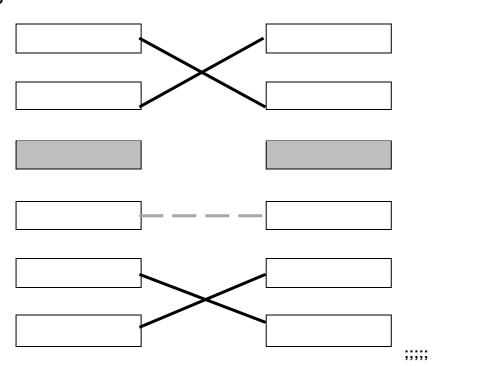
- (c) any one from
  - black absorbs heat
  - shiny reflects heat ; [1]

7 (a) 402; 32; 10.05;

(b) glowing splint;
relights;
[2]

- (c) any one from
  - making steel
  - welding ; [1]
- (d) (s) (l) (g) [1]

8



[5]

Pa	age 4	4	Mark Scheme	Syllabus	Paper
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9	(a)	line	parallel to incident ray from the block ;		[1]
	(b)		= sin <i>i</i> /sin <i>r</i> ; 32.6 ;		[2]
			w sin <i>r</i> = sin54/1.5 w 33		
10	(a)	(i)	9;		[1]
		(ii)	19;		[1]
	(b)	2, 7	(drawn on shells);		[1]
	(c)	(i)	halogens;		[1]
		(ii)	decreases;		[1]
11	(a)	(i)	<ul><li>any three from</li><li>water</li><li>light</li></ul>		
			<ul><li>temperature</li><li>oxygen ;;;</li></ul>		[3]
		(ii)	E; G;		[2]
	(b)	•	three from starch stored in seed/cotyledon starch broken down (amylase) starch to glucose/maltose glucose/maltose soluble		
		•	(glucose used in) respiration energy used for growth (during germination) ;;;		[3]
12	(a)	30			[1]
	(b)	(i)	1.5 ; Volts / <i>V</i> ;		[2]
		(ii)	20(C);		[1]

Pa	age 5	Mark Scheme Cambridge O Level – May/June 2016	Syllabus 5129	Paper 22
	(c)	<ul> <li>any two from</li> <li>current is the same</li> <li>more work done (by the charge against higher resistance)</li> <li>resistance proportional to potential difference</li> <li>R has the highest resistance;;</li> </ul>		[2]
13	(a)	A = oxidation; B = polymerisation; C = steam/water;		[3]
	(b)	speed up the reaction		[1]
	(c)	ECH2 - CHE		
		CH₂ chain open ended bracket showing multiple units		[2]
14	(a)	stronger magnet ; more coils ; louder sound ; answers must be <b>comparative</b>		[3]
	(b)	(i) $v = f\lambda$ ; 5.5 (m); allow $\lambda = v/f$ or $\lambda = 330/60$		[2]
		(ii) at least two waves drawn with same frequency/equal time period; same amplitude;		[2]
		ignore extra waves less than two waves max 1		
15	acid	eria ; ; mel ;		[3]

Pá	age 6	6	Mark Scheme	Syllabus	Paper
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16	(a)	2	H <sub>2</sub> ; <b>both</b> required		[1]
	(b)		cket) fuel ; king margarine ;		[2]
	(c)		aporate to smaller volume ; ol/(allow to) crystallise ;		[2]
		cry	stallisation alone is insufficient		
	(d)	any • •	two from conducts electricity conducts heat malleable ductile		
		•	high melting point/high boiling point high density ;;		[2]
17	E; C; D;				[3]
18	(a)	any • •	one from like poles repel (allow repulsion) align north-south when freely suspended made from iron/steel (allow nickel/cobalt);;		[1]
	(b)	any •	one from induced magnetism steel bar has become magnetised/a magnet;		[1]
	(c)	any •	one from will not attract loses magnetism more quickly;		[1]
19	(a)		air ; acking) hydrocarbons ; allow water/petroleum (crude oil)/named hy	drocarbon	[2]
	(b)	iror	n ; allow Fe		[1]
	(c)	(i)	hydroxide ; allow OH <sup>-</sup>		[1]
		(ii)	sulfuric acid ; allow H <sub>2</sub> SO <sub>4</sub>		[1]

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## 20 any three from

- nerve/brain damage
- mental disorders (e.g. depression/manias/phobias)
- pancreatitis/damage to pancreas
- liver damage/cirrhosis
- (Increased risk of) cancer
- (increase risk of) heart disease
- hypertension/high blood pressure

• strokes ;;;

**21 (a)** 71; [1]

(b) (i) beta; allow electron [1]

(ii) a neutron becomes a proton; allow number of protons increases by 1 and number of neutrons decreases by 1 [1]